

# The breakdown of diopside to $(\text{Ca}, \text{Mg})\text{SiO}_3$ perovskite– $(\text{Mg}, \text{Ca}, \text{Fe})\text{SiO}_3$ glass– $(\text{Mg}, \text{Ca})\text{SiO}_3$ glass– $(\text{Mg}, \text{Ca})\text{SiO}_3$ majorite in a melt vein the Suizhou L6 chondrite

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**Abstract** The Suizhou meteorite is a heavily shocked and melted vein-containing L6 chondrite. It contains a minor amount of diopside with a  $(\text{Ca}_{0.419}\text{Mg}_{0.466}\text{Fe}_{0.088})\text{SiO}_3$  composition, and a shock-metamorphosed diopside grain associated with ringwoodite and lingunite was found in a melt vein of this meteorite. Our electron microprobe, transmission electron microscopic and Raman spectroscopic analyses revealed four silicate phases with different compositions and structures inside this shock-metamorphosed diopside grain, termed phase A, B, C and D in this paper. Phase A is identified as orthorhombic  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$ -perovskite which is closely associated with phase B, the vitrified  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$  perovskite. Phase D is assigned to be  $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$  majorite which is associated with phase C, the vetrified Ca-rich Mg-perovskite with a  $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$  composition. Based on high-pressure and high-temperature experiments, the diopside grain in the melt vein of the Suizhou meteorite would have experienced a  $P$ – $T$  regime of 20–24 GPa and 1800–>2000 °C. Such  $P$ – $T$  conditions are high enough for the decomposition of the diopside and the formation of four different silicate phases. The orthorhombic  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$  perovskite found in the Suizhou L6 chondrite might be considered as the third lower-mantle silicate mineral after bridgmanite and

davemaoite after the detailed analyses of its crystal structure and physical properties being completed.

**Keywords** Suizhou chondrite · Diopside · Shock metamorphism ·  $(\text{Ca}, \text{Mg})\text{SiO}_3$ -perovskite ·  $(\text{Mg}, \text{Ca}, \text{Fe})\text{SiO}_3$  glass ·  $(\text{Mg}, \text{Ca})\text{SiO}_3$  glass ·  $(\text{Mg}, \text{Ca})\text{SiO}_3$  majorite

## 1 Introduction

Diopside is a monoclinic pyroxene mineral of  $\text{CaMgSi}_2\text{O}_6$  composition. It is an important constituent silicate mineral in both pyrolite and basalt, and is believed to be one of the most abundant Ca-bearing minerals in the upper mantle. Numerous experimental studies on the phase relations of diopside  $\text{CaMgSi}_2\text{O}_6$  under high pressures and high temperatures have been performed (Mao et al. 1977; Liu 1987; Tamai and Yagi 1989; Kim et al. 1994; Oguri et al. 1997; Irfune et al. 2000). The final results of all these experiments using diopside crystal as starting material under pressure up to 50 GPa are very similar, namely, the diopside  $\text{CaMgSi}_2\text{O}_6$  had broken down into cubic  $\text{CaSiO}_3$ -perovskite and orthorhombic  $\text{MgSiO}_3$ -perovskite.

Although diopside is found widely in ordinary chondrites (Brearley and Jones 1998), its abundance is comparatively low, and very few occurrences of its high-pressure polymorph have been reported, namely, Ca-rich majorite as its high-pressure polymorph was found in just three ordinary chondrites: Yamato 75,100 (H6) (Tomioka and Kimura 2003), Tenham (L6) (Xie and Sharp 2007) and Villalbeto de la Pena (L6) (Martinez et al. 2019). No high-pressure polymorphs of diopside other than Ca-rich majorite were found in shocked meteorites up to now. In

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this paper, we report the finding of Ca-rich orthorhombic perovskite, vitrified  $(\text{Mg}, \text{Ca}, \text{Fe})\text{SiO}_3$  and  $(\text{Mg}, \text{Ca})\text{SiO}_3$  perovskites, and  $(\text{Mg}, \text{Ca})\text{SiO}_3$ -majorite as the shock-induced decomposition products of diopside in a shock melt vein of the Suizhou L6 chondrite.

## 2 Samples and analytical methods

The Suizhou L6 chondrite fell on April 15, 1986, at Dayanpo, Suizhou City, Hubei Province, China. This meteorite consists of olivine, low-Ca pyroxene, diopside, plagioclase, FeNi-metal, troilite, merrillite, chlorapatite, chromite, ilmenite, pyrophanite, native copper, and shenzhuangite (Xie et al. 2001a, 2011, 2022; Xie and Chen 2016; Bindu and Xie 2017). The Suizhou meteorite was classified as a strongly shock-metamorphosed (S5) meteorite due to the presence of transformation of plagioclase to maskelynite (Xie et al. 2001a; 2011). This meteorite contains a few very thin shock-produced melt veins, and up to 17 high-pressure phases were identified inside or adjacent to melt veins. They are ringwoodite, majorite, lingunite, magnesiowüstite, majorite-pyrope<sub>s</sub>, akimotoite, ahrensite,  $\text{TiO}_2$ -II, and new high-pressure minerals tuite, xieite, chenminite, wangdaodeite, hemleyite, asimowite, poirierite, hiroseite and elgoresyite (Xie et al. 2001a, b, 2002, 2003, 2011, 2016, 2019, 2022; Chen and Xie 2015, Chen et al. 2003, 2008; Ma et al. 2019; Xie and Chen 2016; Bindu et al. 2017, 2019, 2020, 2021; Tomioka et al. 2021).

Polished thin sections were prepared from fragments of the Suizhou meteorite. All observations and physical and chemical analyses are performed *in-situ* on thin sections. The mineral assemblages in polished thin sections of the samples were investigated by optical microscopy using a Leica DM 2500p microscope. A Shimadzu 1720 electron microprobe (EPMA) was used to study the mineral occurrence in back-scattered electron (BSE) mode, and to quantitatively determine the chemical composition of minerals using the wavelength dispersive technique at 15 kV accelerating voltage and beam current of 10 nA. Natural and synthetic phases of well-known compositions were used as standards, such as MgO for Mg,  $\text{Fe}_2\text{O}_3$  for Fe,  $\text{TiO}_2$  for Ti,  $\text{Al}_2\text{O}_3$  for Al,  $\text{MnSiO}_3$  for Mn,  $\text{Cr}_2\text{O}_3$  for Cr, and the data were corrected using a ZAF program. Raman spectra of minerals in the polished thin sections were recorded with a Horiba Labram Aramis instrument. A microscope was used to focus the excitation beam (Ar + laser, 633 nm line) to 1  $\mu\text{m}$  wide spots and to collect the Raman signal. Accumulations of the signal lasted 300 s. The laser power was 2 mW. Transmission electron microscopy (TEM) and selected area electron diffraction (SAED) were performed using FEI Titan G2 60-300 TEM operated at 300 kV accelerating voltage. The

identifications of nano-sized phases were performed with an FEI Talos F200s scanning/transmission electron microscope (STEM) operating at 200 kV, equipped with a field emission gun. Bright-field, high-resolution bright-field, and STEM images of minerals as well as selected area electron diffractions in TEM were obtained from the samples. Compositional analysis of nano-sized crystallites and glass was conducted by using energy dispersive X-ray spectroscopy (EDXS) with an accelerating voltage of 200 kV, a beam current of 1 nA, and a beam size of  $\sim 5$  nm in STEM mode. Accumulation time of X-ray signals was 5 s per analysis.

## 3 Results

### 3.1 Description of diopside in the Suizhou chondritic rock

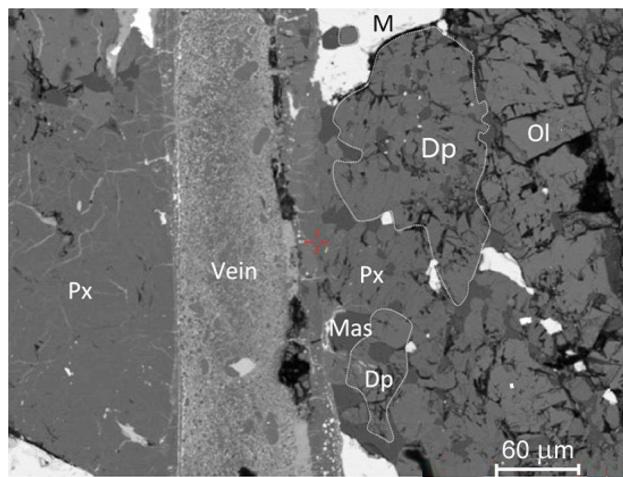
Diopside is a rare silicate phase in the Suizhou meteorite. We only observed five diopside grains with  $(\text{Ca}_{0.419}\text{Mg}_{0.466}\text{Fe}_{0.088}\text{Ti}_{0.007}\text{Cr}_{0.018}\text{Na}_{0.024})_{1.022}(\text{Si}_{0.977}\text{Al}_{0.014})_{0.991}\text{O}_3$  average composition in the Suizhou chondritic rock (Table 1). They have an irregular shape and dark-grey color (Fig. 1), and their Raman spectra show four strong peaks at 322–324, 390–391, 664–666, and 1012–1014  $\text{cm}^{-1}$ , and some weaker bands at 134–135, 509–511 and 1044–1045  $\text{cm}^{-1}$  (Fig. 2), which are identical to that of other natural diopsides (Kubicki et al. 1992).

Interestingly, an unusual diopside grain coexisting with lingunite and ringwoodite was found in a shock melt vein of the Suizhou meteorite (Fig. 3). Our EDXS analysis shows this grain is a CaMgFe-silicate (Fig. 4). The Raman spectrum of this grain shows a few very wide peaks which do not belong to diopside, but the three weak Raman peaks

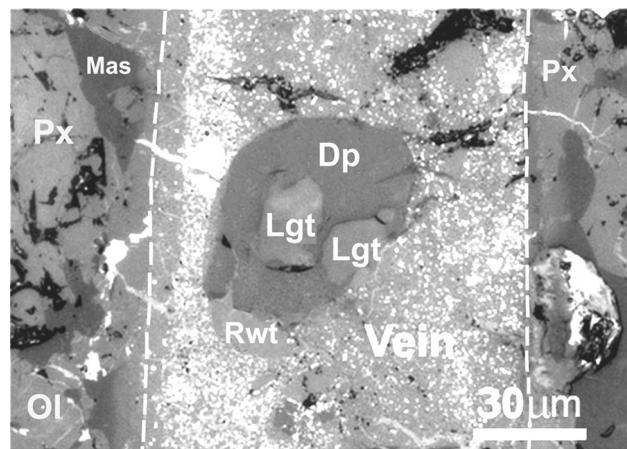
**Table 1** Chemical composition of diopside in the Suizhou chondritic rock (wt%)

| No                      | 1     | 2      | 3      | 4      | 5      | Average |
|-------------------------|-------|--------|--------|--------|--------|---------|
| $\text{SiO}_2$          | 54.17 | 53.75  | 53.40  | 51.60  | 53.24  | 53.23   |
| $\text{TiO}_2$          | –     | 0.63   | 0.85   | 0.73   | 0.55   | 0.52    |
| FeO                     | 4.59  | 5.66   | 5.55   | 6.88   | 5.87   | 5.71    |
| MgO                     | 16.25 | 17.28  | 17.21  | 17.10  | 17.38  | 17.04   |
| CaO                     | 22.96 | 20.77  | 21.11  | 20.78  | 21.06  | 21.34   |
| MnO                     | –     | –      | –      | –      | –      | –       |
| $\text{Al}_2\text{O}_3$ | 0.47  | 0.49   | 0.74   | 0.85   | 0.60   | 0.63    |
| $\text{Cr}_2\text{O}_3$ | 0.85  | 1.23   | 1.33   | 1.52   | 1.20   | 1.23    |
| $\text{Na}_2\text{O}$   | 0.58  | 0.69   | 0.66   | 0.90   | 0.59   | 0.68    |
| Total                   | 99.87 | 100.50 | 100.85 | 100.36 | 100.49 | 100.38  |

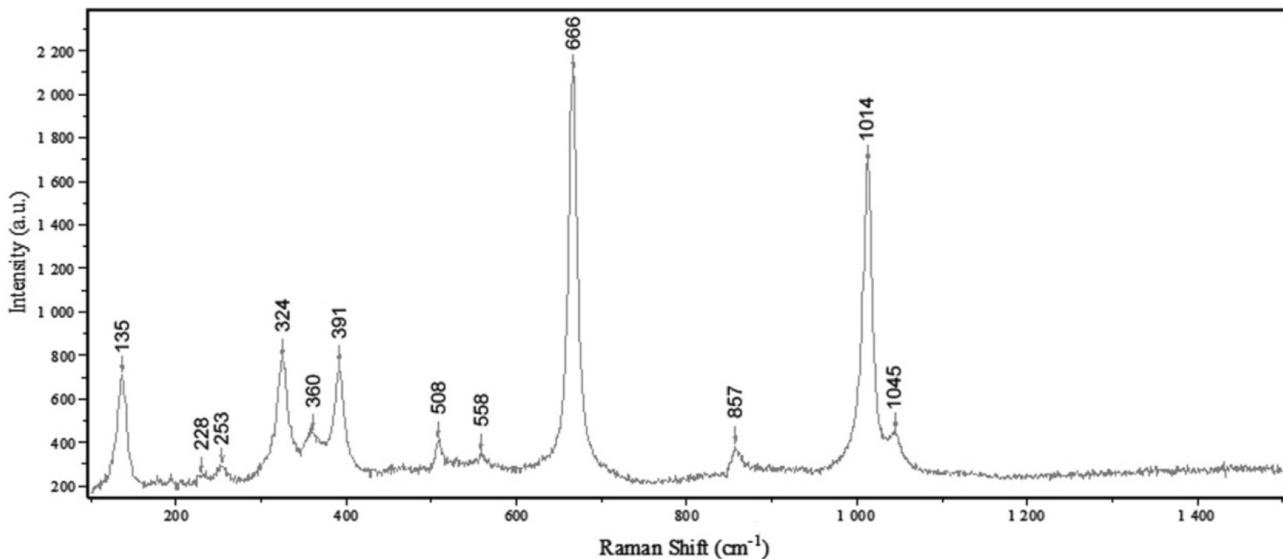
Formula:  $(\text{Ca}_{0.419}\text{Mg}_{0.466}\text{Fe}_{0.088}\text{Ti}_{0.007}\text{Cr}_{0.018}\text{Na}_{0.024})_{1.022}(\text{Si}_{0.977}\text{Al}_{0.014})_{0.991}\text{O}_3$   
Simplified formula:  $(\text{Ca}_{0.419}\text{Mg}_{0.466}\text{Fe}_{0.088})\text{Si}_1\text{O}_3$



**Fig. 1** Back-scattered electron image showing the occurrence of diopside (Dp) in the Suizhou meteorite. Vein = shock melt vein, Ol = olivine, Px = orthopyroxene, Mas = maskelynite, M = FeNi metal



**Fig. 3** Photomicrograph of the shock-metamorphosed diopside (Dp) grain in a shock melt vein (vein) of the Suizhou meteorite. Lgt = lingunite, Rwd = ringwoodite, Ol = olivine, Px = pyroxene, Mas = maskelynite



**Fig. 2** Raman spectrum of a diopside grain in the Suizhou chondritic rock

at 1016,660 and 390  $\text{cm}^{-1}$  for diopside can still be observed on this spectrum (Fig. 5). This indicates that some diopside relicts are remained in this grain. Hence, we assume that this CaMgFe-silicate grain could be a shock-metamorphosed diopside that has broken down to some other silicate phases and glassy materials. It is this grain that draws our attention to conduct a detailed study on it.

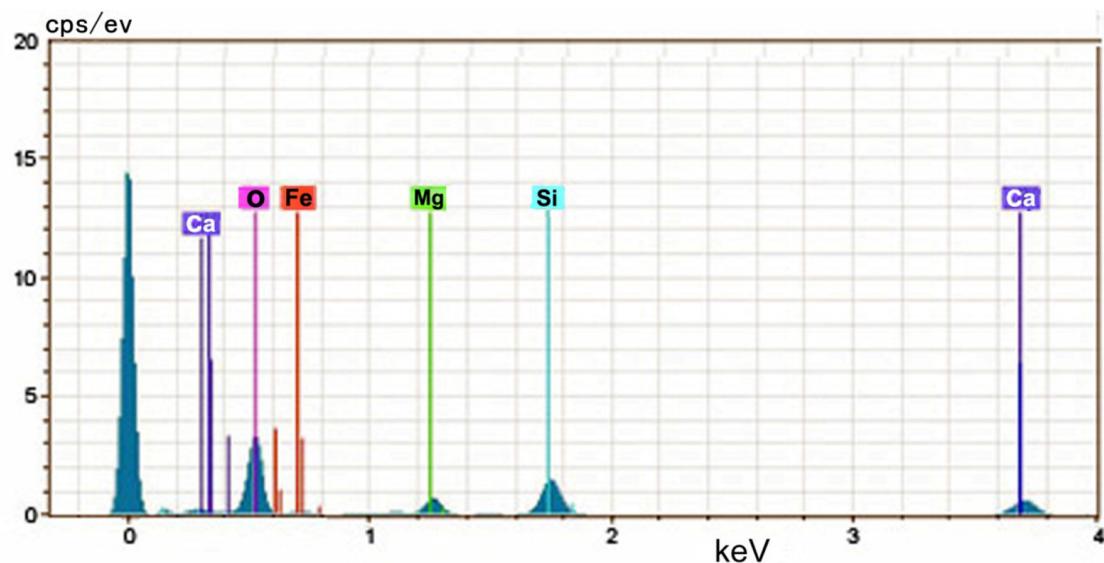
### 3.2 Breakdown of the shock-metamorphosed diopside grain

#### 3.2.1 Occurrence

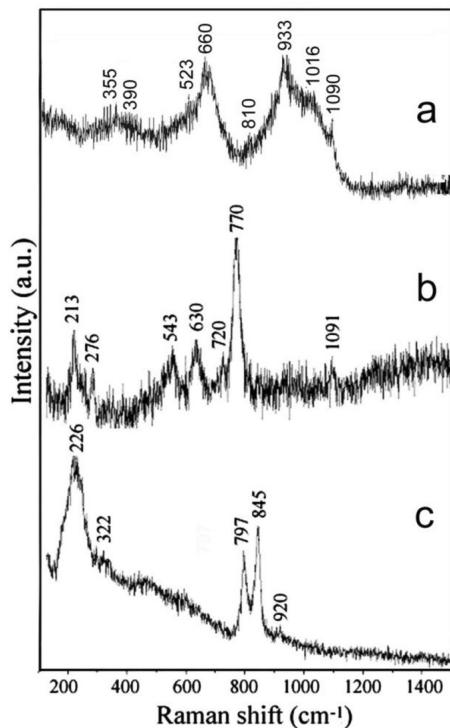
The studied shock-metamorphosed diopside grain occurs in a shock melt vein of the Suizhou meteorite (Fig. 3). It has a

rounded outline and grain size of 40  $\mu\text{m}$  in diameter and closely associates with other two high-pressure minerals lingunite and ringwoodite. This multi-phase grain is embedded in the fine-grained vein matrix consisting of fine-grained majorite-pyrope<sub>s.s.</sub>, magnesiowüstite and small eutectic FeNi-FeS blebs.

Our TEM and STEM studies revealed that this shock-metamorphosed diopside grain is not a mono-mineral phase but an aggregate of different CaMg-silicate phases (Fig. 6). This implies that this grain would have decomposed to several phases by shock. The identification of these phases by STEM, EDXS and Raman spectroscopy will be given in the next sections.



**Fig. 4** EDXS diagram showing the presence of Mg, Ca, Fe, Si, and O in the shock-metamorphosed diopside grain



**Fig. 5** Raman spectrum of the shock-metamorphosed diopside grain (a) coexisting with lingunite (b) and ringwoodite (c) in a shock melt vein of the Suizhou meteorite

### 3.2.2 Chemical compositions

The analysis of chemical compositions of diopside breakdown products was carried out by EDXS equipped with the FEI Talos F200s STEM. The results of analyses on 10 grains, from No.1 to No.10 marked in Fig. 6, are listed in

Table 2. From this table, we can see that four silicate phases of different compositions in this shock-metamorphosed diopside grain are observed, namely, phase A, phase B, phase C, and phase D. We use the average element contents of each phase and the fixed number of oxygens per formula ( $O = 3$ ) to calculate their chemical formulas and the results are listed below the table.

#### 3.2.3 Phase A—the orthorhombic $(Ca_{0.663}Mg_{0.314})SiO_3$ -perovskite

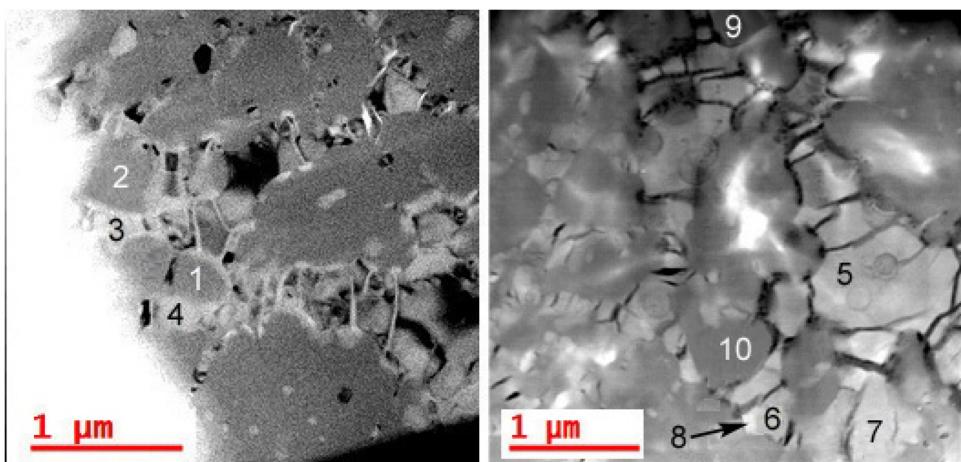
Phase A is the main breakdown product of diopside in the Suizhou meteorite upon shock. Figure 7 are the bright field, SAED, and high-resolution (HR) TEM images of this phase. Figure 7a and 7b show the close association of phase A with phase B. Phase A has a rounded shape and grey color with a smooth surface.

As it has been shown in Table 2, the chemical formula of phase A is  $(Ca_{0.663}Mg_{0.314})_{0.977}(Si_{0.982}Al_{0.06})_{0.988}O_3$ , and its simplified formula is  $(Ca_{0.663}Mg_{0.314})SiO_3$ , or even  $(Ca, Mg)SiO_3$ . Here the atom units of Ca in phase A is as twice larger than that of Mg, and phase A does not contain FeO (Table 2).

Based on the results of our TEM study (Fig. 7) and EDXS analyses (Table 2), we can identify that phase A (grain No. 1 in Fig. 6) is a crystalline silicate with  $(Ca_{0.663}Mg_{0.314})SiO_3$  composition, for it gives electron diffraction spots and lattice fringes on its SAED and HRTEM images (Fig. 7c, e, and f), respectively.

For further identification of phase A, besides the HRTEM image and the SAED pattern taken from grain No.1 (Fig. 7e and f), we also obtained HRTEM images and related SAED patterns on another 4 phase A grains.

**Fig. 6** STEM bright field images of the shock-metamorphosed diopside grain in a shock melt vein of the Suizhou meteorite. The numbers from 1 to 10 show the tiny grains of different phases contained in this diopside grain where the STEM study and EDXS analysis were conducted



**Table 2** Chemical composition of diopside breakdown products in Suizhou meteorite (wt%)

| Grain no.                      | Phase A |        |       | Phase B |       |        |        |       | Phase C |       |       | Phase D |       |       |
|--------------------------------|---------|--------|-------|---------|-------|--------|--------|-------|---------|-------|-------|---------|-------|-------|
|                                | 1       | 2      | Ave.  | 3       | 4     | 5      | 6      | Ave.  | 7       | 8     | Ave.  | 9       | 10    | Ave.  |
| SiO <sub>2</sub>               | 53.93   | 54.10  | 54.02 | 51.83   | 51.37 | 55.87  | 55.50  | 53.64 | 57.83   | 56.48 | 57.16 | 55.29   | 54.58 | 54.94 |
| TiO <sub>2</sub>               | –       | –      | –     | 0.06    | –     | –      | 0.35   | 0.10  | –       | 0.32  | 0.16  | –       | –     | –     |
| FeO                            | –       | –      | –     | 8.81    | 8.86  | 4.99   | 3.25   | 6.48  | –       | –     | –     | –       | –     | –     |
| MgO                            | 14.09   | 11.71  | 12.95 | 25.29   | 23.20 | 23.74  | 23.33  | 23.89 | 33.35   | 31.55 | 32.45 | 23.53   | 19.88 | 21.71 |
| CaO                            | 31.33   | 34.20  | 32.77 | 8.12    | 10.12 | 14.35  | 17.09  | 12.41 | 7.72    | 9.68  | 8.70  | 19.44   | 23.77 | 21.61 |
| Al <sub>2</sub> O <sub>3</sub> | 0.56    | –      | 0.28  | 3.72    | 3.84  | –      | 0.48   | 2.01  | 0.51    | 1.35  | 0.93  | 0.98    | 0.95  | 0.97  |
| Cr <sub>2</sub> O <sub>3</sub> | –       | –      | –     | 0.11    | 0.07  | 1.17   | –      | 0.34  | 0.58    | 0.53  | 0.56  | 0.67    | 0.75  | 0.71  |
| Na <sub>2</sub> O              | –       | –      | –     | 1.19    | 1.55  | –      | –      | 0.69  | –       | –     | –     | –       | –     | –     |
| Total                          | 99.91   | 100.07 | 99.42 | 99.13   | 99.01 | 100.12 | 100.00 | 99.56 | 99.99   | 99.91 | 99.96 | 99.91   | 99.93 | 99.95 |

Chemical formula: Phase A— $(\text{Ca}_{0.663}\text{Mg}_{0.314})_{0.977}(\text{Si}_{0.982}\text{Al}_{0.06})_{0.988}\text{O}_3$ ;

Phase B— $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098}\text{Ti}_{0.001}\text{Na}_{0.024}\text{Cr}_{0.005})_{1.060}(\text{Si}_{0.967}\text{Al}_{0.043})_{1.010}\text{O}_3$ ;

Phase C— $(\text{Mg}_{0.853}\text{Ca}_{0.167}\text{Ti}_{0.001}\text{Na}_{0.023}\text{Cr}_{0.004})_{1.048}(\text{Si}_{0.949}\text{Al}_{0.041})_{0.990}\text{O}_3$ ;

Phase D— $(\text{Mg}_{0.578}\text{Ca}_{0.414}\text{Cr}_{0.010})_{1.002}(\text{Si}_{0.981}\text{Al}_{0.020})_{1.001}\text{O}_3$

Figure 8 shows a representative HRTEM image and SAED pattern obtained from grain A4. The *d* spacing values of phase A measured from above mentioned 5 grains are listed in Table 3. In this table, grain No.1 in Table 2 is marked as A1.

The *d* spacing values of 5 phase A grains show the following characteristics: (1) All 5 grains gave very similar *d* spacing values in 2.671–2.690 Å (2.678 Å in average), 1.902–1.914 Å (1.909 Å in average), and 1.329–1.355 Å (1.339 Å in average). (2) The *d* spacing values measured for the strongest electron diffraction spots are 2.671–2.690 Å (2.678 Å on average). (3) The *d* spacing values measured for the less strong electron diffraction spots are 1.902–1.914 Å (1.909 Å in average) and 1.329–1.355 Å (1.339 Å on average).

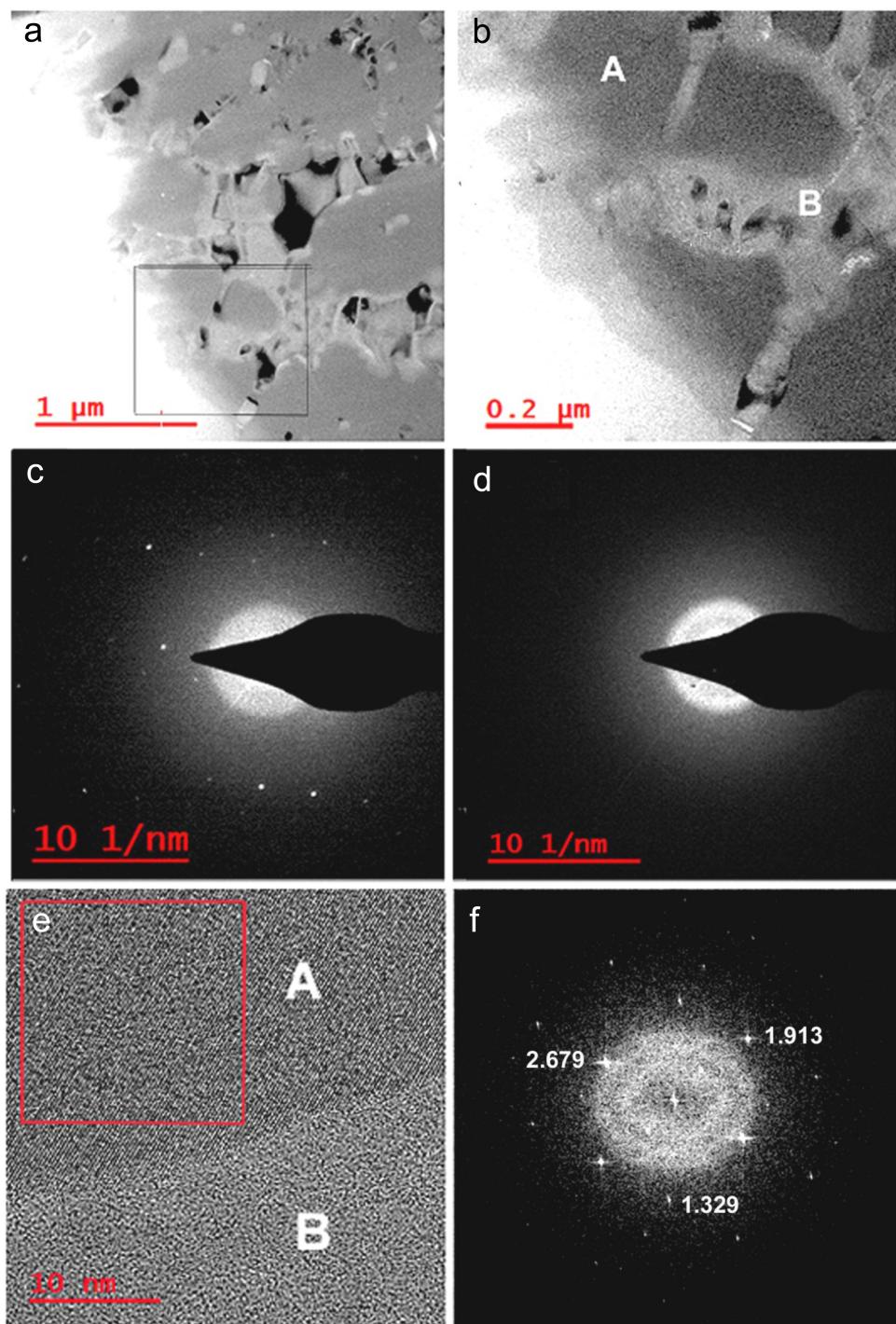
The obtained *d* spacing values and cell parameters (*a* = 5.358 Å, *b* = 7.636 Å, *c* = 5.400 Å, *V* = 222.934 Å<sup>3</sup>) for phase A can be compared with those of orthorhombic

CaTiO<sub>3</sub>-perovskite (PDF 72·1192) (Table 4), and are close to those of the new high-pressure mineral hiroseite, the Fe-rich analog of bridgemanite with orthorhombic perovskite structure and  $(\text{Fe}_{0.54}\text{Mg}_{0.37}\text{Na}_{0.03}\text{Ca}_{0.02})(\text{Si}_{0.89}\text{Al}_{0.15})\text{O}_3$  composition discovered in a shock vein of the same Suizhou meteorite (Bindi et al. 2020). Hence, we identify phase A in the Suizhou shock-metamorphosed diopside as orthorhombic  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$ -perovskite.

### 3.2.4 Phase B—the vitrified $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$ perovskite

Phase B is also a main breakdown product of the shock-metamorphosed diopside in the Suizhou melt vein. Figure 7 shows the bright field, SAED, and high-resolution (HR) TEM images of this phase. Figure 7a and b demonstrate its occurrence and close association with phase A. Phase B is of irregular or ladder shape and light color, and

**Fig. 7** TEM images showing the breakdown of a diopside grain to phase A (grain No. 1 in Fig. 6) and Phase B (grain No. 4 in Fig. 6). **a** Bright-field image of a shock-metamorphosed diopside grain. **b** Enlarged image from the frame area of 7a showing the occurrence of phase A and phase B. **c** SAED pattern taken from phase A. **d** SAED pattern taken from phase B. **e** HRTEM image showing the crystalline phase A and glassy phase B. **f** SAED pattern taken from the frame area of phase A

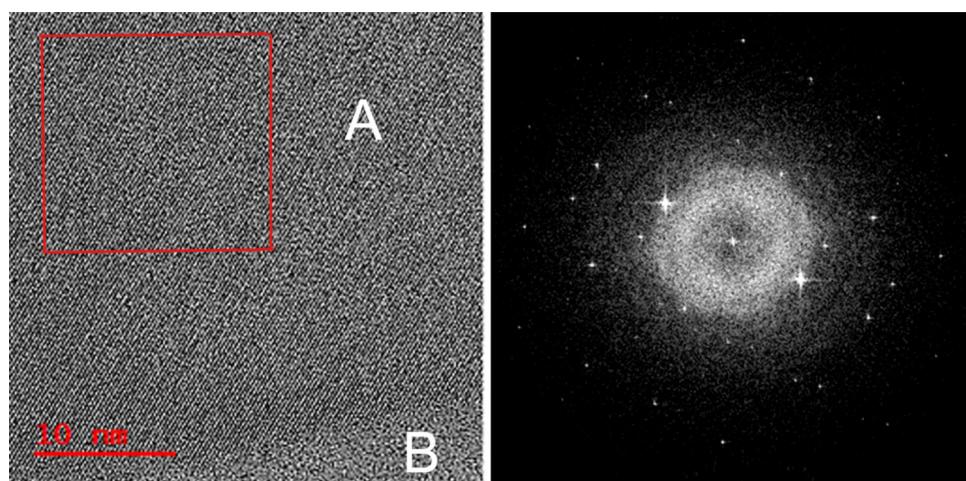


occurs around or in the interstices of phase A grains. Because of the volume shrinkage during cooling and solidification, some fractures/cracks crosscutting the phase B grains are observed (Fig. 6, grains 3, 4, 5, and 6).

As it has been shown in Table 2, the chemical formula of phase B is  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$ , or  $(\text{Mg}, \text{Ca}, \text{Fe})\text{SiO}_3$ . It must be pointed out that FeO is always present in phase B (Table 2), and the FeO content (6.48 wt%) in phase B is a little higher than that in the un-metamorphosed diopside (5.40–5.71 wt%) occurred in the Suizhou chondritic rock (Table 1).

formula is  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$ , or  $(\text{Mg}, \text{Ca}, \text{Fe})\text{SiO}_3$ . It must be pointed out that FeO is always present in phase B (Table 2), and the FeO content (6.48 wt%) in phase B is a little higher than that in the un-metamorphosed diopside (5.40–5.71 wt%) occurred in the Suizhou chondritic rock (Table 1).

**Fig. 8** The HRTEM image of grain A4 (Left, see Table 3) and the SAED pattern taken from the frame area (Right). A = phase A, B = phase B



**Table 3** The  $d$  spacing of phase A measured from SAED patterns (Å)

| Grains | A.1          | A2           | A3           | A4           | A5           | Average      | Intensity |
|--------|--------------|--------------|--------------|--------------|--------------|--------------|-----------|
| 1      | <b>2.679</b> | <b>2.671</b> | <b>2.690</b> | <b>2.671</b> | <b>2.679</b> | <b>2.678</b> | VS        |
| 2      | 2.133        | 2.127        | 2.133        | 2.127        |              | 2.130        | W         |
| 3      | <b>1.913</b> | <b>1.914</b> | <b>1.903</b> | <b>1.912</b> | <b>1.902</b> | <b>1.909</b> | S         |
| 4      | 1.492        | 1.489        |              | 1.490        |              | 1.490        | W         |
| 5      | <b>1.329</b> | <b>1.355</b> | <b>1.336</b> | <b>1.334</b> | <b>1.342</b> | <b>1.339</b> | M         |
| 6      |              | 1.283        | 1.289        | 1.287        | 1.285        | 1.286        | W         |
| 7      |              |              |              | 1.196        | 1.196        | 1.196        | W         |
| 8      |              |              | 1.087        | 1.103        | 1.095        | 1.095        | W         |
| 9      |              |              |              | 1.063        | 1.063        | 1.063        | W         |
| 10     |              |              |              | 1.019        | 1.018        | 1.019        | W         |
| 11     |              |              |              | 0.958        | 0.958        | 0.958        | W         |
| 12     |              |              |              | <b>0.911</b> | <b>0.911</b> | <b>0.911</b> | W         |
| 13     |              |              |              | 0.896        | 0.896        | 0.896        | W         |

The main d values are shown in bold

**Table 4** Comparison of electron diffraction data of Ca-perovskite in Suizhou meteorite with standard orthorhombic  $\text{CaTiO}_3$ -perovskite

| CaSiO <sub>3</sub> -perovskite in Suizhou             |          |           | CaTiO <sub>3</sub> -perovskite (PDF 72-1192)          |          |          |
|---|----------|-----------|---|----------|----------|
| <i>hkl</i>  | <i>d</i> | <i>I*</i> | <i>hkl</i>  | <i>d</i> | <i>I</i> |
| 200   | 2.678    | VS        | 121   | 2.7025   | 999      |
|   |          |           | 200   | 2.6835   | 272      |
| 040   | 1.909    | S         | 040   | 1.9109   | 669      |
|   |          |           | 321   | 1.5529   | 181      |
| 220   | 1.339    | M         | 220   | 1.3512   | 166      |
| 310   | 1.196    | M         | 310   | 1.2085   | 78       |
| Orthorhombic,   |          |           | Orthorhombic,   |          |          |
| $a = 5.358 \text{ \AA}$ , $b = 7.636 \text{ \AA}$ ,   |          |           | $a = 5.367 \text{ \AA}$ , $b = 7.643 \text{ \AA}$ ,   |          |          |
| $c = 5.400 \text{ \AA}$ , $V = 222.934 \text{ \AA}^3$ |          |           | $c = 5.443 \text{ \AA}$ , $V = 223.272 \text{ \AA}^3$ |          |          |

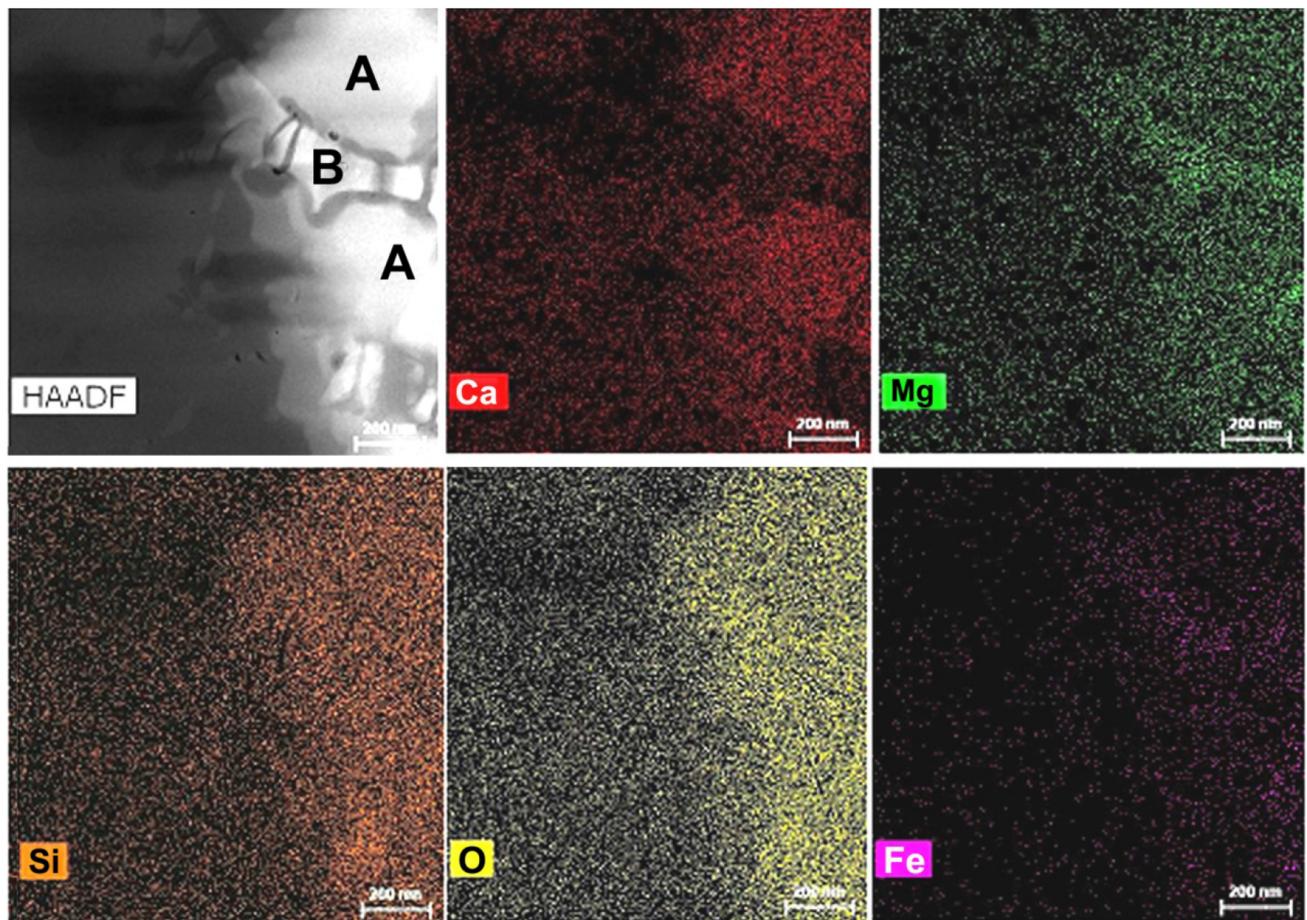
\*VS = very strong, S = strong, M = moderate

Interestingly, the chemical composition of phase B and phase A are complementary with each other in Mg and Ca. The atom units of Mg and Ca for phase A almost equal to those of Ca and Mg for phase B (Table 2). This future can also be seen on STEM element maps of these two phases (Fig. 9), where the Mg-rich and Fe-bearing phase B occurs in between two Ca-rich phase A grains. Hence, we assume that both phase B and phase A are the main decomposition products of diopside with  $(\text{Ca}_{0.435}\text{Mg}_{0.449}\text{Fe}_{0.084})\text{SiO}_3$  composition in the Suizhou meteorite.

Based on the results of our TEM study (Fig. 7) and EDXS analyses (Table 2), we could identify that phase B is a silicate glass with  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$  composition, for it does not give any electron diffraction spots and lattice fringes on its SAED and HRTEM images, respectively (Fig. 7d and e).

Chen et al (2004a) reported the finding of a unique occurrence of fine-grained shock-produced  $(\text{Mg}, \text{Fe})\text{SiO}_3$  glass that intimately coexisted with majorite in the shock veins of the Suizhou meteorite. They indicated that this  $(\text{Mg}, \text{Fe})\text{SiO}_3$  glass is the vitrified product of  $(\text{Mg}, \text{Fe})\text{SiO}_3$  perovskite after pressure release at post-shock temperature. They also indicated that the heating experiments and the molecular- and lattice dynamics calculations indicated that the crystalline  $\text{MgSiO}_3$  perovskite would be decompressed to an amorphous phase near the ambient pressure from its high-pressure stability fields at modest temperatures (Durben and Wolf 1992; Hemmati et al. 1995). The vitrification of  $\text{MgSiO}_3$  perovskite begins above 127 °C and is complete by 477 °C at ambient pressure (Durben and Wolf 1992). The post-shock temperature must have been higher than 477 °C inducing a rapid vitrification of crystalline  $\text{MgSiO}_3$  perovskite.

Our phase B is a shock-produced Ca-rich and Fe-bearing  $\text{MgSiO}_3$  glass with  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$  composition, it would have similar behavior with the shock-produced  $(\text{Mg}, \text{Fe})\text{SiO}_3$  glass reported by Chen et al (2004a) in



**Fig. 9** STEM element maps of phase B and phase A in the shock-metamorphosed diopside grain. The Mg-rich and Fe-bearing phase B occur in between two Ca-rich phase A grains

the shock veins of the same Suizhou meteorite after pressure release at post-shock temperature. Hence, we assume that the glassy phase B would be the vitrified  $(\text{Mg}, \text{Ca}, \text{Fe})\text{SiO}_3$  perovskite.

### 3.2.5 Phase C—the verified Ca-rich $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$ perovskite

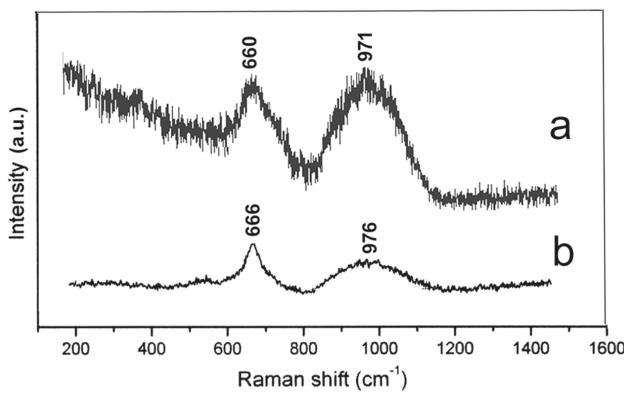
Phase C is the third breakdown product of the shock-metamorphosed diopside in the Suizhou meteorite (Fig. 6, grains 7 and 8). It has an irregular shape and light grey color. Some volume shrinkage-produced fractures/cracks crosscutting the phase C grains are also observed. The chemical formula of phase C is  $(\text{Mg}_{0.853}\text{Ca}_{0.167}\text{Ti}_{0.001}\text{Na}_{0.023}\text{Cr}_{0.004})_{1.048}(\text{Si}_{0.949}\text{Al}_{0.041})_{0.990}\text{O}_3$  (Table 2), and its simplified formula is  $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$ , or  $(\text{Mg}, \text{Ca})\text{SiO}_3$ . Although the atom units of Mg in phase C is as 5 times larger than that of Ca, but the Ca content is still rich enough ( $\text{CaO} = 8.70 \text{ wt\%}$ ), and phase C does not contain  $\text{FeO}$  (Table 2).

The Raman spectrum of phase C only shows two wide bands at  $660$  and  $971 \text{ cm}^{-1}$  (Fig. 10a), indicating that

phase C in the Suizhou shock-metamorphosed diopside is a silicate glass. The Raman spectrum of phase C can be compared with that of the shock-produced  $(\text{Mg}_{0.759}\text{Fe}_{0.202}\text{Ca}_{0.014})\text{SiO}_3$  glass, the verified perovskite, in a melt vein of the same Suizhou meteorite reported by Chen et al. in 2004a (Fig. 10b). Hence, we identify that phase C in the Suizhou shock-metamorphosed diopside is a vitrified Ca-rich Mg-perovskite with  $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$  composition.

### 3.2.6 Phase D—the Mg–Ca majorite with $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$ composition

Phase D is the fourth breakdown product of the shock-metamorphosed diopside in the Suizhou meteorite (Fig. 6, grains 9 and 10). It has an irregular or granular shape and dark grey color. Phase D grains are commonly surrounded by glassy phase C. The chemical formula of phase D is  $(\text{Mg}_{0.578}\text{Ca}_{0.414}\text{Cr}_{0.010})_{1.002}(\text{Si}_{0.981}\text{Al}_{0.020})_{1.001}\text{O}_3$  (Table 2), and its simplified formula is  $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$ , or  $(\text{Mg}, \text{Ca})\text{SiO}_3$ . Here we can see that the atom unit of Mg in phase D is just a litter larger than that of Ca, and phase D does not contain  $\text{FeO}$  (Table 2).



**Fig. 10** **a** Raman spectrum of phase C in the Suizhou shock-metamorphosed diopside grain. **b** Raman spectrum of  $(\text{Mg}_{0.759}\text{Fe}_{0.202}\text{Ca}_{0.014})\text{SiO}_3$  glass in a Suizhou shock melt vein (Chen et al. 2004a)

The Raman spectrum of the phase D grain surrounded by glassy phase C in the Suizhou shock-metamorphosed diopside grain shows Raman peaks at 976, 927, 806, 660, 590, and 352 cm<sup>-1</sup> (Fig. 11a). The two wide bands at 976 and 660 cm<sup>-1</sup> can be assigned to the coexisting glassy phase C (see Fig. 10a), and the peaks at 927, 806, 660, 590, and 352 cm<sup>-1</sup> can be compared with that of majorite, the  $(\text{Mg}, \text{Fe})\text{SiO}_3$ -garnet, in a Suizhou melt vein reported by Xie et al. in 2001a (Fig. 11b). Hence, we assume that phase D in the Suizhou shock-metamorphosed diopside is  $(\text{Mg}, \text{Ca})\text{SiO}_3$  majorite with  $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$  composition.

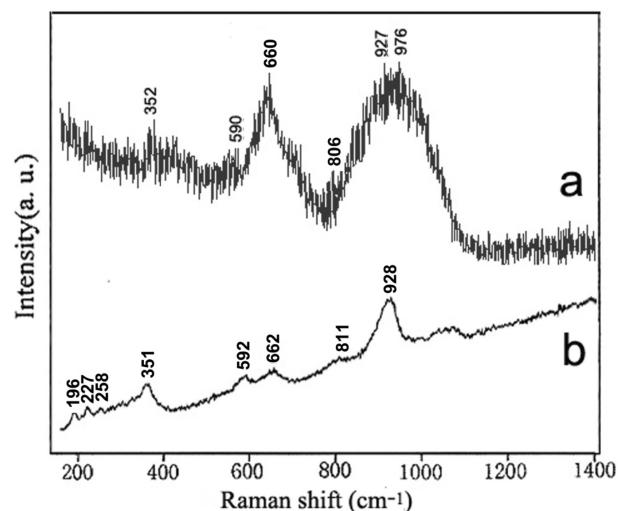
## 4 Discussion

### 4.1 The coexisting of orthorhombic crystalline Ca-perovskite and vitrified

#### 4.1.1 Ca-rich and Fe-bearing Mg-perovskite

As we know that a series of experimental studies on the phase relations of diopside  $\text{CaMgSi}_2\text{O}_6$  under high pressures and high temperatures have been performed. The most pioneering experiments on phase transformations in diopside were performed by Mao et al. (1977) using a diamond anvil cell and laser heating techniques, and samples were quenched from 21.7 and 42.1 GPa. Their results show that, under pressure, the diopside had disproportionated into two perovskite phases, with chemical compositions close to  $\text{CaSiO}_3$  and  $\text{MgSiO}_3$ .

Tamai and Yagi (1989) then conducted the study of phase transformation of synthetic  $\text{CaMgSi}_2\text{O}_6$  single crystal up to 50 GPa. They indicated that  $\text{CaMgSi}_2\text{O}_6$  was also found to break down into  $\text{CaSiO}_3$  perovskite and  $\text{MgSiO}_3$  perovskite. Kim et al. (1994) also carried out a phase transformation study on natural crystalline diopside at a pressure of up to 34.5 and 30 GPa, respectively, and



**Fig. 11** **a** Raman spectrum of phase D + glassy phase C in the shock-metamorphosed diopside grain. **b** Raman spectrum of majorite in a Suizhou shock melt vein (Xie et al. 2001a)

at  $\sim 1000$  °C. They found that the crystalline diopside breaks down to cubic  $\text{CaSiO}_3$  perovskite and orthorhombic  $\text{MgSiO}_3$  perovskite. Oguri et al. (1997) studied phase transformations in diopside  $\text{Ca}_{0.92}\text{Mg}_{1.08}\text{Si}_2\text{O}_6$  at pressures of 18–24 GPa and temperatures of 1000–2400 K. Their results show that diopside transformed into  $(\text{Mg}, \text{Ca})\text{SiO}_3$  garnet and  $\text{CaSiO}_3$  perovskite, and then to  $\text{MgSiO}_3$ -perovskite and  $\text{CaSiO}_3$ -perovskite. Irfiune et al. (2000) performed experiments on high-pressure phase transformation in diopside  $\text{CaMgSi}_2\text{O}_6$  at pressures 23–27 GPa and temperatures 1000–1900 °C. They found that all run products consisted of  $\text{MgSiO}_3$ -rich orthorhombic perovskite and amorphous  $\text{CaSiO}_3$ , the latter having been converted from cubic perovskite.

To sum up, the results of all above-listed experiments performed on diopside  $\text{CaMgSi}_2\text{O}_6$  at pressures range of 18–50 GPa are very similar: the diopside had finely broken down into cubic  $\text{CaSiO}_3$ -perovskite and orthorhombic  $\text{MgSiO}_3$ -perovskite, and one of the intermediate products is  $(\text{Mg}, \text{Ca})\text{SiO}_3$  garnet (Ca-rich Mg-majorite).

The natural  $(\text{Mg}, \text{Fe})\text{SiO}_3$  perovskite, as the high-pressure mineral that partially replaced strongly deformed pyroxene, was found by Tomioka and Fujino (1997) in the shock veins of the Tenham (L6) chondrite. The natural cubic  $\text{CaSiO}_3$ -perovskite (daveomaite) was found as inclusion in a diamond from Orapa kimberlite, Botswana (Tschauner et al. 2021). However, No crystalline Ca-perovskite was found in any shocked meteorites before (Chen et al. 2004a; Xie and Chen 2016; Martinez et al. 2019).

The natural crystalline Mg-rich Ca-perovskite with  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$  composition (phase A) now is found for the first time in a shock-metamorphosed diopside grain of the Suizhou L6 chondrite. This crystalline Ca-perovskite

is closely associated with vitrified Ca-rich and Fe-bearing Mg-perovskite with  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$  composition (phase B). Both phases constitute the main products of diopside breakdown in the Suizhou melt vein. Although the breakdown products of natural diopside in Suizhou chondrite are consistent with those obtained by high-pressure and high-temperature experiments, we could still observe some differences between them: (1) our crystalline Mg-rich Ca-perovskite belongs to an orthorhombic system that is different with the cubic davemaoite and cubic crystalline Ca-perovskite experimentally obtained using diopside as the starting material. The reason for this is still unknown, but it may be related to the different  $P-T$  history between natural and experimental phase transition processes. (2) our Ca-rich and Fe-bearing Mg-perovskite have been vitrified after pressure release at post-shock temperature, while the experimentally obtained  $\text{MgSiO}_3$  perovskite is the crystalline phase after rapid annealing. (3) our vitrified Ca-rich Mg-perovskite contains 6.48 wt% of FeO (Table 2) indicating that the Fe-bearing diopside ( $\text{FeO} = 5.40\text{--}5.71$  wt%) in the Suizhou chondritic rock shifted its total Fe into vitrified Mg-perovskite during its breakdown upon shock, whereas the experimentally obtained  $\text{MgSiO}_3$  perovskite does not, for its starting material is Fe-free diopside.

#### 4.2 The coexisting of $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$ majorite with vitrified Ca-rich Mg-perovskite with $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$ composition

Natural  $(\text{Mg},\text{Fe})\text{SiO}_3$  majorite has been found in melt veins of many shocked L6 chondrites (Coleman 1977; Xie et al. 2001a; Chen et al. 2004a). Natural Ca-rich Mg-majorite, as the breakdown product of diopside, was first reported in the Shergotty Martian meteorite (Malavergne et al. 2001). Then Ca-rich Mg-majorite has also been found in two ordinary chondrites: Yamato 75,100 (H6) (Tomioaka and Kimura 2003) and Tenham (L6) (Xie and Sharp 2007). The Ca-rich Mg-majorite ( $\text{En}_{65}\text{Fs}_9\text{Wo}_{26}$ ) in Yamato 75,100 has a granular texture and is associated with a Ca-rich glass ( $\text{En}_{23}\text{Fs}_5\text{Wo}_{72}$ ), whereas in Tenham, Ca-rich Mg-majorite ( $\text{En}_{64}\text{Fs}_{10}\text{Wo}_{27}$ ) occurs as a nanometer symplectic intergrowth with amorphous Ca-bearing glass (Xie and Sharp 2007). More recently, Martinez et al. (2019) reported the third finding of Ca-rich ( $\text{En}_{47}\text{Fs}_7\text{Wo}_{45}$ ) Mg-majorite together with Ca-poor ( $\text{En}_{77}\text{Fs}_{21}\text{Wo}_2$ ) Mg-majorite in the melt veins of the Villalbeto de la Pena (L6) chondrite. Their electron microprobe analyses and Raman spectra show that original low-Ca pyroxene, high-Ca pyroxene, olivine, and plagioclase in the shock vein are transformed into their high-pressure polymorphs, low-Ca Mg-majorite, high-Ca Mg-majorite, ringwoodite, and a mixture of jadeite-lingunite, respectively. Chen et al (2004a) reported the finding of the majorite with  $(\text{Mg}_{0.761}\text{Fe}_{0.202}\text{Ca}_{0.015})\text{SiO}_3$

composition that intimately coexisted with vitrified perovskite with  $(\text{Mg}_{0.759}\text{Fe}_{0.202}\text{Ca}_{0.014})\text{SiO}_3$  composition in shock veins of the Suizhou meteorite.

In the Suizhou shock-metamorphosed diopside grain, we observed MgCa-majorite with  $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$  composition (phase D) that coexisted with vitrified Ca-rich Mg-perovskite with  $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$  composition (phase C). This assemblage is similar to that in the Yamato 75,100 (H6) chondrite, where the Ca-rich Mg-majorite ( $\text{En}_{65}\text{Fs}_9\text{Wo}_{26}$ ) is associated with a Ca-rich silicate glass ( $\text{En}_{23}\text{Fs}_5\text{Wo}_{72}$ ). Our Suizhou MgCa-majorite is also very similar to the Ca-rich ( $\text{En}_{47}\text{Fs}_7\text{Wo}_{45}$ ) Mg-majorite in the melt veins of the Villalbeto de la Pena (L6) chondrite (Martinez et al. 2019). Furthermore, one of the intermediate products experimentally obtained on the diopside is  $(\text{Mg},\text{Ca})\text{SiO}_3$  majorite, (Oguri et al. 1997). Therefore, we assume that the MgCa-majorite associated with vitrified Ca-rich Mg-perovskite in the Suizhou meteorite might also be the common product formed by decomposition of the shock-metamorphosed diopside.

#### 4.3 The $P-T$ history of diopside decomposition in the Suizhou melt vein

High-pressure experiments indicated that  $\text{MgSiO}_3$  crystallizes in perovskite structures above 23 GPa and at  $\sim 2000$  °C (Liu 1974, 1975; Gasparik 1996; Chen et al. 2004b), and that  $\text{MgSiO}_3$  majorite has a  $P-T$  stability field between 16 and 22.5 GPa and 1600–2500 °C (Gasparik 1996; Presnall 2000; Chen et al. 2004b). The crystalline  $\text{MgSiO}_3$  perovskite would be decompressed to an amorphous phase near the ambient pressure from its high-pressure stability fields at temperatures higher than 477 °C (Durben and Wolf 1992; Hemmati et al. 1995). Quench experiments have confirmed the decomposition of diopside in a wide temperature range of 1000–1900 °C and at pressures 23–27 GPa (Irifune et al. 2000).

The pressure and temperature history in the shock veins of meteorites can be well-constrained by high-pressure mineral assemblage (Agee et al. 1995; Chen et al. 1996; Xie and Chen 2016, 2020). For instance, on the bases of the high-pressure mineral assemblage of Ca-rich Mg-majorite, Ca-poor Mg-majorite, ringwoodite, jadeite-lingunite, and magnesiowüstite in the melt veins of the Villalbeto de la Pena (L6) chondrite, the  $P-T$  conditions of between 16 and 28 GPa and 2000–2200 °C were estimated (Martinez et al. 2019).

The assemblage of ringwoodite, majorite, lingunite, and magnesiowüstite in the shock veins of the Suizhou meteorite should have crystallized at pressures of 20–22 GPa and temperatures of 1800–2000 °C (Chen et al. 1996; Xie et al. 2001a), however, the recent discovery of hiroseite, the Fe-rich analog of perovskite-structured bridgemanite, in

a Suizhou melt vein, raised its  $P-T$  conditions up to 24 GPa and  $> 2000$  °C (Bindi et al. 2020; Agee et al. 1995). Hence, we conclude that the diopside-containing melt vein in the Suizhou meteorite would have experienced a high-pressure and high-temperature regime of 20–24 GPa and  $1800 - > 2000$  °C. This high enough  $P-T$  regime is satisfied for the decomposition of diopside, as well as for the formation of crystalline orthorhombic  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$ -perovskite (phase A), Ca-rich and Fe-bearing Mg-perovskite with  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})$  composition (phase B), and Ca-rich and Fe-free Mg-perovskite with  $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$  composition (phase C). Both phase B and Phase C then turned to amorphous phases after pressure release and upon cooling, while phase A is remained stable. This is because the high-pressure experiments on the dioipside revealed that decomposed  $\text{CaSiO}_3$  is stable at pressure release, but the crystalline  $\text{MgSiO}_3$  perovskite would be decompressed to an amorphous phase near the ambient pressure from its high-pressure stability fields at temperatures higher than 477 °C (Durben and Wolf 1992; Hemmati et al. 1995). As for the  $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$  majorite (phase D), it would be formed at 16–22.5 GPa and 1600–2500 °C (Gasparik 1996; Presnall 2000; Chen et al. 2004b). This  $P-T$  condition is lower than that for Ca- and Mg-perovskites, but is within the  $P-T$  regime that the Suizhou melt vein experienced.

As we know, davemaoite, the natural cubic  $\text{CaSiO}_3$ -perovskite, is the high-pressure silicate recovered from the lower mantle in 2021 (Tschauner et al. 2021), and bridgemanite, another lower-mantle mineral, was found inside a melt vein of the Tenham L6 chondrite in 2014 (Tschauner et al. 2014). Both formed an exclusive club as the only lower-mantle silicate minerals confirmed in nature. But, our orthorhombic  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$  perovskite found in the Suizhou L6 chondrite might serve as the third lower-mantle silicate mineral, a new member of this exclusive club after its crystal structure and physical property analyses being completed.

## 5 Conclusion

- (1) The Suizhou L6 chondrite contains a minor amount of diopside with  $(\text{Ca}_{0.419}\text{Mg}_{0.466}\text{Fe}_{0.088})\text{SiO}_3$  composition, and a unique shock-metamorphosed diopside grain associated with lingunitite and ringwoodite was found in a shock melt vein of the Suizhou meteorite.
- (2) Four silicate phases of different composition and structures in this shock-metamorphosed diopside grain are observed, namely, phase A, phase B, phase C and phase D.

- (3) Phase A is identified as orthorhombic  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$ -perovskite which is closely associated with phase B, the vitrified  $(\text{Mg}_{0.642}\text{Ca}_{0.290}\text{Fe}_{0.098})\text{SiO}_3$  perovskite. The chemical composition of phase A and phase B are complementary with each other in Ca and Mg.
- (4) Phase D is assigned to be Ca-rich  $(\text{Mg}_{0.578}\text{Ca}_{0.414})\text{SiO}_3$ -majorite which is associated with phase C, the verified Ca-rich Mg-perovskite with  $(\text{Mg}_{0.853}\text{Ca}_{0.167})\text{SiO}_3$  composition.
- (5) The diopside-containing melt vein in the Suizhou meteorite would have experienced a high-pressure and high-temperature regime of 20–24 GPa and  $1800 - > 2000$  °C. This  $P-T$  regime is high enough for the decomposition of the diopside and the formation of different silicate phases.
- (6) After the crystal structure and physical property analyses are completed the orthorhombic  $(\text{Ca}_{0.663}\text{Mg}_{0.314})\text{SiO}_3$  perovskite found in the Suizhou L6 chondrite might serve as the third lower-mantle silicate mineral after bridgemanite and dave Maoite.

## Declarations

**Conflict of interest** We declare no conflict of interest in this study. The manuscript has not been submitted to more than one journal for simultaneous consideration.

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