

Fractionation mechanism of iron isotopes in highly fractionated granites from the Xinxian Pluton, Western Dabie Orogen, Central China

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Abstract Iron isotopes are important for tracing the magmatic process. The fractionation of iron isotopes in granite is up to 0.55 ‰. In this study, Wangjiagou (XWJ) granite and Tayueping (XTY) granite in the Xinxian pluton of the Western Dabie orogen were evaluated. Both the XTY and XWJ granite belong to monzogranites, with high SiO₂ (74.42–76.82 wt.%) contents. The granites are depleted of Nb and Ti but enriched with Pb and K, and they display negative Eu anomalies (Eu/Eu* = 0.40–0.52) on REE plots that are normalized by chondrite. The δ⁵⁶Fe values of the XTY granites vary from 0.19 ± 0.03 ‰ to 0.27 ± 0.04 ‰, and the δ⁵⁶Fe values of the XWJ granites are 0.34 ± 0.02 ‰ and 0.36 ± 0.01 ‰, respectively. Both the XTY and the XWJ granites belong to highly fractionated granites due to their SI (solidification index), DI (differentiation index), and content of CaO. Evidence from the iron isotopes shows that neither fluid exsolution, alteration, weathering, nor partial melting can explain the enrichment of the heavy iron isotopes. The results modeled using the Rayleigh equation showed that fractional crystallization can produce Δ⁵⁶Fe_{melt-crystal} with the value of 0.08–0.15 ‰. In conclusion, fractional crystallization was the main factor controlling the fractionation of iron

isotopes, and the change of melt composition may also lead to the enrichment of heavy iron isotopes in the residual melt.

Keywords Iron isotope · Fractionated granite · Dabie orogen · Fractionation mechanism

1 Introduction

Iron has four isotopes (⁵⁴Fe, ⁵⁶Fe, ⁵⁷Fe, and ⁵⁸Fe) and three common valence states (0, + 2, + 3). Significant fractionation of iron isotopes occurred during the magmatic process in the range of –0.03‰ to 0.55‰ (Zhu et al. 2016; Du et al. 2017). Xu et al. (2017) reported that seven migmatites produced Δ⁵⁶Fe_{leucosome-melosome} of about 0.09 ± 0.06 ‰, which showed that the partial melting of crustal rocks promoted the enrichment of heavy iron isotopes in melts. Wu et al. (2017a, b) argued that the high δ⁵⁶Fe values of most high-silica granitic rocks may not be caused by the accumulation of feldspar, by investigating the δ⁵⁶Fe values of the main diagenetic minerals (e.g., plagioclase, K-feldspar, and biotite) in high-silica granites, and leucosomes derived from the different source rock. Recently, Du et al. (2019) analyzed three different types of high-silica granites and suggested that the crystallization of iron-titanium oxide minerals containing ferrous ions was the main reason for the high δ⁵⁶Fe values in granite. Generally, previous studies have proposed four possible mechanisms that govern iron isotopic fractionation: (1) partial melting (Williams et al. 2005; Weyer and Ionov 2007; Telus et al. 2012; Xu et al. 2017); (2) fractional crystallization (Schuessler et al. 2009; Teng et al. 2011; Sossi et al. 2012; Collinet et al. 2017; Du et al. 2019; Chen

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et al. 2021; Deng et al. 2021; Liang et al. 2022); (3) exsolved fluids (Poitrasson and Freydier 2005; Heimann et al. 2008; Du et al. 2019; Deng et al. 2022); and (4) alteration and weathering (Poitrasson et al. 2008; Liu et al. 2014).

Highly fractionated granites (HFG) have usually undergone a high degree of “crystal-melt differentiation” processing such as fractional crystallization, which may have led to the fractionation of iron isotopes. Wu et al. (2017a, b) argued that the presence of characteristic accessory minerals such as tourmaline and lepidolite indicates that the sample is highly fractionated granite (Wu et al. 2017a, b). After integrating a large amount of granite data, it was found that the above characteristic accessory minerals were not universal (Yang 2019), however, they demonstrated a common feature of having dark minerals constituting less than 10% of the whole. The controlling fractional mechanism of iron isotopes in highly fractionated granites remains an open question. The iron isotopic composition of seven highly fractionated granites from the Western Dabie orogen is presented in this study, to discover the controlling fractional mechanism of iron isotopes in highly fractionated granites.

2 Geological background and samples

The Dabie orogen is usually divided into two areas by the Shangcheng–Macheng fault from east to west, namely the Eastern Dabie orogen and the Western Dabie orogen, respectively (Fig. 1). The Western Dabie orogen was located in the west segment of the whole Dabie orogen, and was extremely active in magmatism during the Mesozoic era as a result of the northern frontier of the Yangtze Block (YB) subducts into the North China Block (NCB) (Li et al. 1999; Zheng et al. 2003a; Zhao et al. 2007; Suo et al. 2012; Xu et al. 2012; Liu et al. 2014; Niu and Freydier 2020). The geological activities between NCB and YB changed from collision and compression into extension until the Early Cretaceous (~ 130 Ma), thus resulting in the collapse of thickened crust and the ascent of mantle material (Xu et al. 2007; Zheng 2008; He et al. 2011, 2013; Chen et al. 2013b, a; Zhu et al. 2019).

As reported in many studies, numerous high-silica granite plutons were exposed in the study area (Fig. 1) (Meng 2013; Chen et al. 2013b, 2015; Huang et al. 2018). Reported Sr–Nd isotope data has indicated that the Xinxian pluton was derived from the lower crust of the northern margin of the YB without assimilation or contamination (Chen et al. 2013a; Yang et al. 2020). In terms of the identification of source rock and the age of the Xinxian pluton, previous studies on Sr–Nd–Pb–Hf isotopes have been carried out in detail (Chen et al. 2013a, 2020; Zhou

et al. 2013). It is believed that the Xinxian pluton was formed by the partial melting of TTG type magmatic rocks with a crustal depth of less than 35 km, and the zircon U–Pb age ranges from 134.3 ± 1.4 to 125.5 ± 1.5 Ma, which indicates that they were formed during the tectonic mechanism transition of the Dabie orogenic belt. Zhao et al. (2013) reported that the diagenetic temperature of the Xinxian pluton may be lower than 759 °C, the pressure may be at 1.15–2.07 kbar, and the corresponding depth may be 4.3–7.6 km, by calculation with a mineral thermobarometer.

3 Analytical methods

The whole-rock major element compositions were analyzed by X-ray fluorescence spectrometry (XRF) and the whole-rock trace elements were determined by Inductively Coupled Plasma Mass Spectrometer (ICP-MS) at the ALS Chemex (Guangzhou) Co., Ltd. The relative standard deviation (RSD) for the major elements was 3%–5%, and it was 10% lower for trace elements. The sample was decomposed with $\text{HF-H}_2\text{SO}_4$, the remaining F in the solution was added to H_3BO_3 for complexation, the content of ferrous (Fe^{2+}) was titrated with reference $\text{K}_2\text{Cr}_2\text{O}_7$ with sodium diphenylamine sulfonate as an indicator, and the amount of FeO was calculated. The content of ferric iron was calculated based on the measured total iron and ferrous iron content.

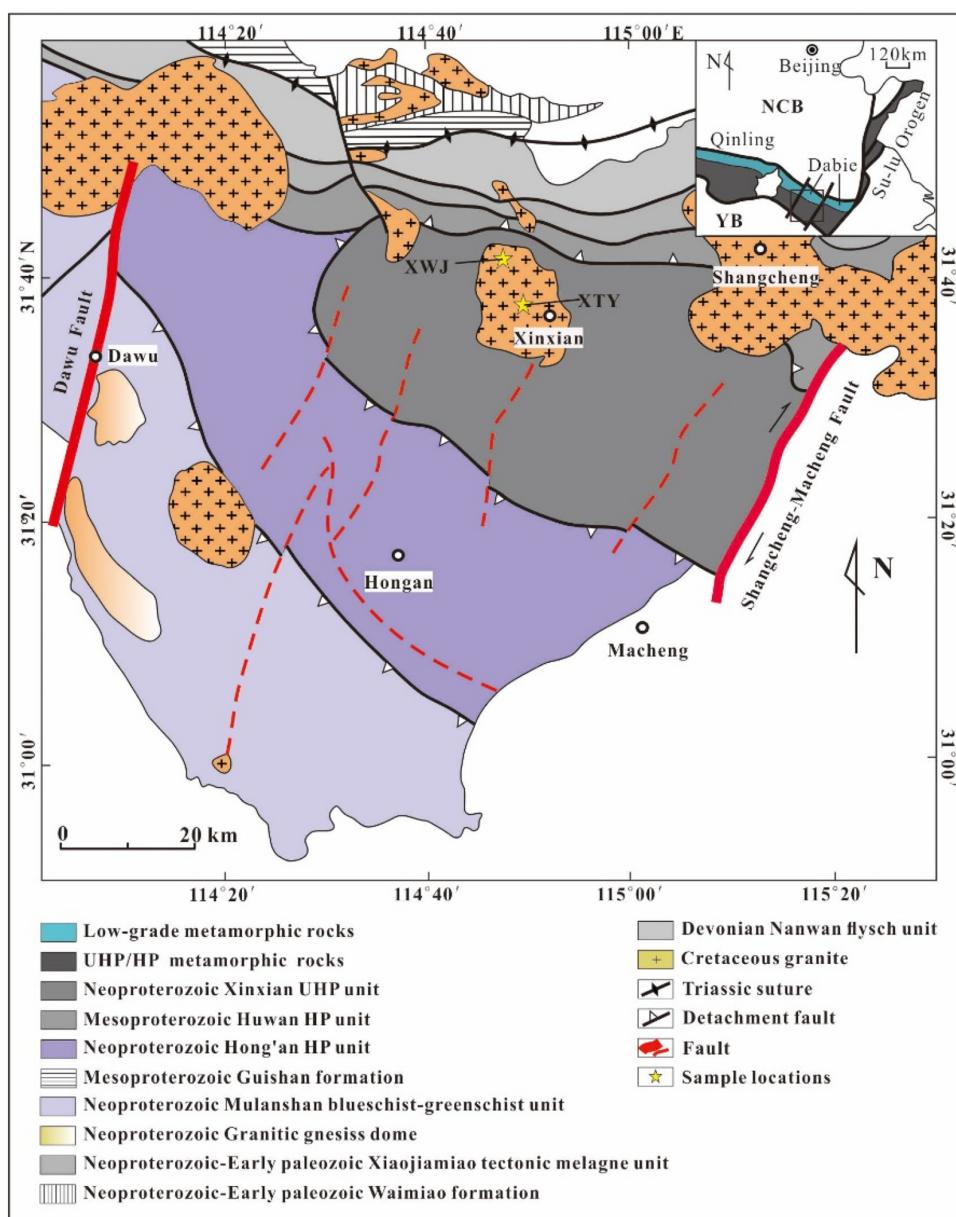
Fe isotope compositions were measured in the Metal Stable Isotopes Geochemistry Laboratory at the University of Science and Technology of China (USTC), Hefei, China. The digestion and evaporation procedures were carried out as described by Xia et al. (2017). The purification procedures for the iron isotopes were described by Huang et al. (2011). The sample solution was performed on the Neptune-Plus MC-ICP-MS and analyzed with the sample-standard bracketing (SSB) method. The analytical procedures for Fe isotopes were as described by Xia et al. (2017). Each sample was typically measured three times. Isotopic data were reported using the delta (δ) notation and expressed as

$$\delta^i\text{Fe} = \left[\left(\frac{i\text{Fe}}{\text{Fe}^{54}} \right)_{\text{sample}} / \left(\frac{i\text{Fe}}{\text{Fe}^{54}} \right)_{\text{standard}} - 1 \right] \times 1000 (\%) \quad (1)$$

where i refers to mass 56 or 57.

The international standards BIR-1 and RGM-1 were purified and measured together with unknown samples, yielding $\delta^{56}\text{Fe}$ values of 0.05 ± 0.01 % ($n = 3$, 2SD) and 0.19 ± 0.02 % ($n = 3$, 2SD). The measured standard values agree with previously published data within error for both Fe isotopes (Xia et al. 2017).

Fig. 1 Geological sketch map of Western Dabie area
(modified after Zheng et al.
2003b; Liu et al. 2004)



4 Results

4.1 Major elements of samples

The analytical results of the major element composition of the samples are given in Table 1. All seven samples contained high SiO₂ (74.42–76.82 wt.%, average: 75.53 wt.%), K₂O (4.40–4.81 wt.%, average: 4.64 wt.%), and Na₂O (3.76–4.25 wt.%, average: 4.01 wt.); and low MgO (0.07–0.29 wt.%, average: 0.17 wt.%). It has previously been established that the content of Fe₂O₃^T in XTY granite (average: 1.16 wt.%) was twice that of the content of Fe₂O₃^T in XWJ granite (average: 0.54 wt.%).

4.2 Trace elements of samples

The analytical results of the trace element composition of the samples are listed in Table 1. The total REE concentration of the seven granitic samples ranged from 74 to 223 ppm and enriched in LREE with the values of (La/Yb)_N ranging from 10.33 to 15.71, which demonstrated that negative Eu anomalies (Eu/Eu* = 0.40–0.52) on the REE plots were normalized by chondrite (Fig. 2a). They were characterized by being enriched in Pb, K, U, and Th, and lacking Nb, Ti, Sr, and Ba, as shown in the trace element spider diagram that was normalized by a primitive-mantle (Fig. 2b).

Table 1 Major (wt.%) and trace elemental (ppm) compositions of granites from Xinxian pluton

Composition	The XTY granite					The XWJ granite	
	XTY-1-001	XTY-2-001	XTY-3-002	XTY-3-003	XTY-4-002	XWJ-001	XWJ-003
SiO ₂	75.75	75.05	75.68	74.42	74.84	76.11	76.82
Al ₂ O ₃	13.14	13.25	12.99	13.40	13.42	13.34	12.81
BaO	0.04	0.06	0.06	0.07	0.07	0.02	0.03
CaO	0.43	0.65	0.84	1.14	0.86	0.45	0.43
Fe ₂ O ₃ ^T	0.88	1.14	1.12	1.35	1.33	0.55	0.53
K ₂ O	4.73	4.81	4.65	4.40	4.80	4.53	4.45
MgO	0.13	0.18	0.17	0.29	0.27	0.07	0.09
MnO	0.04	0.05	0.05	0.05	0.06	0.05	0.02
Na ₂ O	4.11	3.76	3.86	4.07	3.89	4.13	4.25
P ₂ O ₅	0.01	0.02	0.03	0.05	0.04	< 0.01	0.01
TiO ₂	0.11	0.15	0.15	0.19	0.18	0.06	0.07
FeO	0.29	0.32	0.38	0.54	0.45	0.06	0.07
Fe ³⁺	0.39	0.55	0.49	0.52	0.58	0.34	0.32
Fe ^{3+/\sumFe}}	0.44	0.48	0.43	0.38	0.43	0.61	0.60
SI	0.01	0.02	0.02	0.03	0.03	0.01	0.01
LOI	0.35	0.45	0.44	0.67	0.65	0.53	0.35
Ce	58.80	76.70	81.90	94.50	98.50	24.50	28.10
Dy	1.84	2.26	2.43	2.98	2.97	0.84	0.89
Er	1.14	1.32	1.36	1.71	1.76	0.57	0.68
Eu	0.42	0.55	0.59	0.72	0.67	0.16	0.16
Ga	17.40	20.80	19.15	20.50	20.80	22.90	21.50
Gd	1.99	2.71	2.76	3.60	3.42	0.80	1.00
Ge	0.24	0.38	0.42	0.46	0.51	0.33	0.33
Hf	4.30	4.80	4.20	5.40	5.20	2.70	3.50
Ho	0.38	0.45	0.49	0.61	0.60	0.18	0.21
La	32.20	36.20	45.00	49.20	55.70	12.90	14.40
Lu	0.26	0.29	0.26	0.32	0.35	0.17	0.19
Nb	19.40	28.00	18.50	20.00	23.80	28.10	28.00
Nd	19.10	24.50	26.10	32.20	33.30	7.50	8.40
Pb	28.70	29.50	30.60	28.40	29.30	38.00	34.80
Pr	6.12	7.50	8.38	9.93	10.65	2.57	2.86
Rb	192	244	237	220	244	352	320
Sm	2.74	3.90	3.97	4.96	4.96	1.15	1.34
Sr	81.30	132.00	155.00	193.00	171.50	46.80	57.90
Ta	2.10	2.40	1.40	1.50	1.90	2.70	2.30
Th	24.30	37.50	30.20	41.30	44.30	22.00	21.10
Tm	0.19	0.22	0.21	0.27	0.29	0.11	0.13
U	4.40	10.30	3.71	3.51	6.31	7.83	5.16
Yb	1.47	1.63	1.53	1.90	2.06	0.90	1.00
Zr	130	144	148	186	178	54	71
ΣREE	142.40	176.70	191.80	220.30	233.10	74.60	80.00
(La/Yb) _N	15.71	15.93	21.13	18.57	19.39	10.28	10.33
Eu/Eu*	0.52	0.49	0.52	0.50	0.47	0.48	0.40
10 ⁴ × Ga/Al	2.51	2.96	2.78	2.88	2.92	3.21	3.17

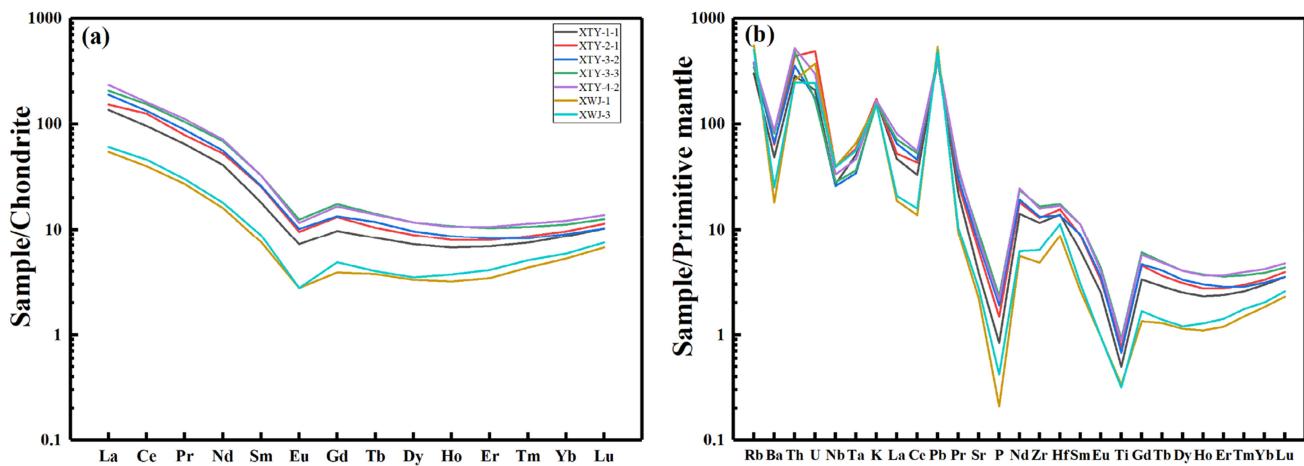


Fig. 2 Plots of chondrite-normalized REE (a) and primitive mantle-normalized trace elements (b) of the Xinxian pluton. Chondrite and primitive mantle-normalized values are from McDonough and Sun (1995) and Sun and McDonough (1989)

4.3 Fe isotopes of samples

Fe isotope data are presented in Table 2. The $\delta^{56}\text{Fe}$ values for all the samples in this study ranged from 0.19 to 0.36‰. The XWJ granite had the highest $\delta^{56}\text{Fe}$ values ($0.36 \pm 0.01\text{‰}$, 2SD, $n = 3$). However, with the $\delta^{56}\text{Fe}$ values ranging from $0.19 \pm 0.03\text{‰}$ to $0.27 \pm 0.04\text{‰}$, the

composition of Fe isotopes in the XTY granite was slightly less than those in the XWJ granite.

5 Discussion

5.1 Petrogenesis of highly fractionated granites

Seven samples were collected from two sites in the Xinxian pluton; five samples were obtained near the Xinxian county; the other two were obtained from the south of the Wangjiagou reservoir. They are referred to as XTY granite and XWJ granite, respectively, hereafter. The XTY granite (Fig. 3) consisted of plagioclase (30%–35%), K-feldspar (~35%), quartz (20%–25%), and biotite (~5%). The composition of XWJ granite was similar to the XTY granite, but the content of biotite in XWJ granite was less than 5%, which was less than the biotite content in XTY granite. Both were monzogranite.

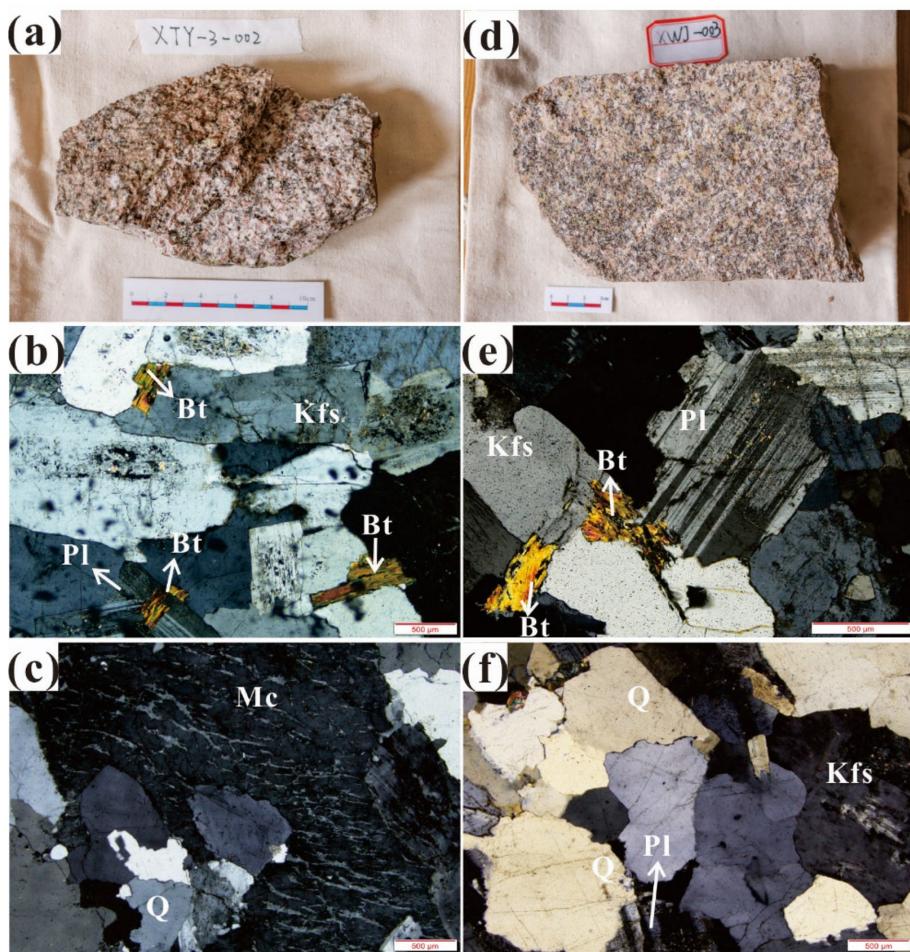
Granites can be classified into four types, namely I, S, M, and A (Chappel and White 1992; Loiselle and Wones 1979). Chen et al. (2013a) discovered that the Xinxian pluton comprises I-type granite, and its classification has been described in detail and is not repeated in this study. However, the degree of fractionation can be further interpreted, to provide more constraints from the perspective of mineralogy and major elements in this study.

The main dark mineral in the seven samples was biotite with the absence of hornblende, and its content ranged from 3 to 10% (Fig. 10a). The $10^4 \times \text{Ga}/\text{Al}$ ratios were out of the range of A-type granite ($10^4 \times \text{Ga}/\text{Al} = 2.6$, Whalen et al. 1987) which varied from 2.51 to 3.20 in this study. Previous studies (Cin-ty and Douglas 2015; Yang 2019) revealed that the major elemental content in highly fractionated granite was $\text{CaO} < 1\text{ wt.\%}$, $\text{SiO}_2 > 72\text{ wt.\%}$, and

Table 2 Iron isotopes composition of HFG from Xinxian pluton and NG from eastern Dabie. Superscript * denote data from He et al. 2017

Sample	Rock type	$\delta^{56}\text{Fe}(\text{\textperthousand})$	2SD	$\delta^{57}\text{Fe}(\text{\textperthousand})$	2SD	n
XTY-1-001	HFG	0.27	0.02	0.4	0.06	3
XTY-2-001		0.2	0.05	0.29	0.1	3
XTY-3-002		0.22	0.02	0.34	0.14	3
XTY-3-003		0.27	0.04	0.34	0.05	3
XTY-4-002		0.19	0.03	0.33	0.05	3
XWJ-001		0.34	0.02	0.49	0.06	3
XWJ-003		0.36	0.01	0.63	0.05	3
06FS-1*	NG	0.061	0.051	0.102	0.072	9
06FS-4*		0.122	0.051	0.178	0.072	9
06ZY-2*		0.099	0.029	0.128	0.042	9
07YC-1*		0.139	0.023	0.229	0.04	9
07YC-3*		0.153	0.023	0.24	0.04	9
07XM-3*		0.143	0.023	0.211	0.04	9
BIR-1	Basalt	0.05	0.01	0.07	0.07	3
RGM-1	Rhyolite	0.19	0.02	0.28	0.06	3

Fig. 3 Specimen photographs and photomicrographs of the Xinxian pluton. **a** The specimen photographs of the XTY granite. **b** and **c** are the minerals in the XTY granite, **d** is the specimen photographs of the XWJ granite, and **e** and **f** are the polysynthetic twin of plagioclase and biotite in the XWJ granite. Pl, plagioclase; Kfs, K-feldspar; Q, quartz; Bt, Biotite; Mc, Microlite



$\text{Na}_2\text{O} + \text{K}_2\text{O} > 7$ wt.%. Meanwhile, its SI [solidification index, $\text{MgO}/(\text{MgO} + \text{Fe}_2\text{O}_3 + \text{Na}_2\text{O} + \text{K}_2\text{O})$] should be less than 6, and the DI (differentiation index, $\text{Q} + \text{Or} + \text{Ab} + \text{Ne} + \text{Lc} + \text{Kp}$, all six minerals were calculated by CIPW, the results are listed in Table 3) should be more than 88. According to Table 1, the CaO contents of the seven samples were all less than 1 wt.%. The SI ranged from 0.007 to 0.029, by calculation. According to the results of the CIPW calculation (Table 3), the contents of Q, Or, and Ab were more than 88, not including the whole content of the DI. As shown in Fig. 4c, and d, the samples in our study were all plotted within the FG area. In

summary, it was confirmed that the samples belong to highly differentiated I-type granite.

In this study, the U–Pb ages, as measured by a previous study, were collected (Fig. 5). As shown in Fig. 5, the diagenetic age of the Xinxian pluton is between 153.4 ± 1.1 and 125.5 ± 1.5 Ma, and the corresponding geological background is from the tectonic transformative stage of the YB (Yangtze Block) and NCB (North China Block), with the two blocks turning from the collision to the extension. In Fig. 6, the samples from the Xinxian pluton were obtained from the syn-collisional granite area. Combined with the geological background, this indicates

Table 3 Selected calculation results of CIPW standard minerals

	XTY-1-001	XTY-2-001	XTY-3-002	XTY-3-003	XTY-4-002	XWJ-001	XWJ-003
Q	32.89	33.57	33.63	31.31	31.83	33.99	34.26
Or	28.20	28.77	27.67	26.24	28.53	27.00	26.47
Ab	35.08	32.20	32.89	34.76	33.11	35.24	36.19
An	2.08	3.13	4.00	5.35	4.03	2.18	2.08
Mt	0.57	0.77	0.71	0.76	0.84	0.18	0.08
Il	0.21	0.28	0.28	0.36	0.34	0.11	0.1
Ap	0.02	0.04	0.07	0.11	0.09	0.02	0.02

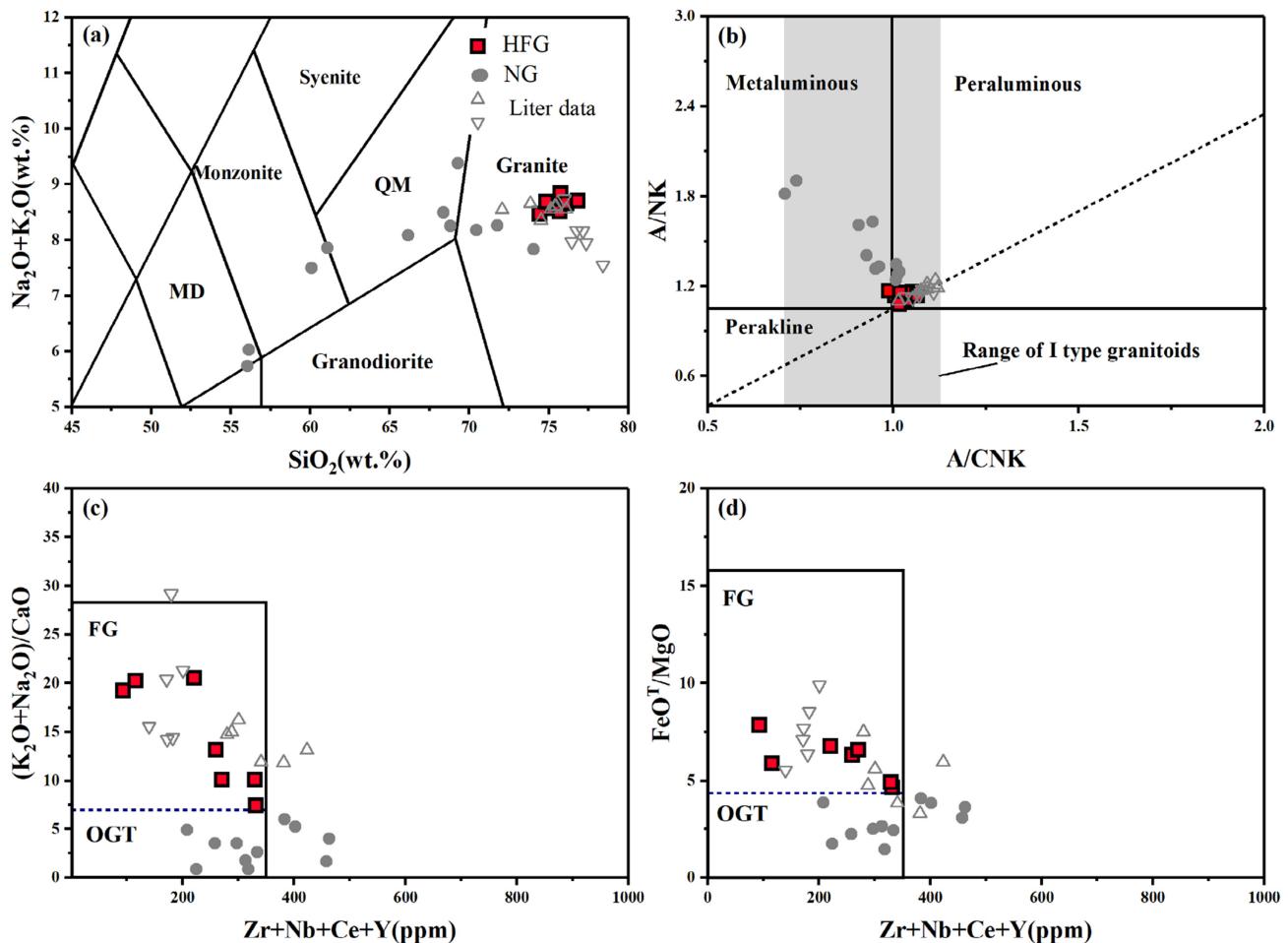


Fig. 4 Total alkali versus SiO₂ (Middlemost 1985) where MD and QM stand for mozodiorite and quartz monzonite, ANK versus ACNK, and the discrimination diagrams of (FeO^T + MgO) versus (Zr + Nb + Ce + Y), in which, $x = 350$, $y = 4$ and 16 (c) and (Na₂O + K₂O)/CaO versus (Zr + Nb + Ce + Y), the $x = 350$, $y = 7$ and 28 (d) for the XTY and XWJ granite (after Whalen et al 1987)

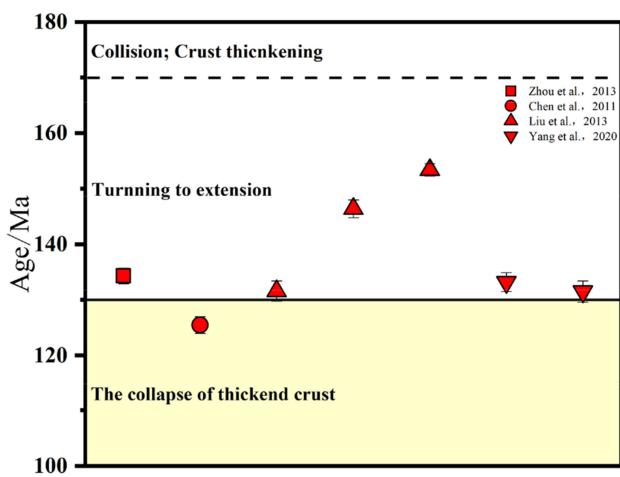


Fig. 5 Statistical map of isotopic age of the Xinxian pluton

that the Xinxian pluton was likely to have formed in the syn-collisional tectonic environment.

5.2 Fractional mechanism of iron isotopes in HFG

The value of $\delta^{56}\text{Fe}$ from all seven samples ranged from 0.19 to 0.36 ‰, which is consistent with the range of iron isotopes of granite (0.07–0.39 ‰) that has been reported in previous studies (Zhu et al. 2016). Some data with the content of SiO₂ > 72 wt.% were presented by Foden et al. (2015) and He et al. (2017). An obvious trend can be observed in Fig. 7a, which shows the HFG from the Xinxian pluton and that it is enriched in heavy iron isotopes with an increase in SiO₂. Compared with previous reports (Foden et al. 2015; He et al. 2017), the HFG from the Xinxian pluton is enriched with heavier iron isotopes, which may imply the existence of invisible mechanisms that have led to iron isotope fractionation, apart from fractional crystallization.

In the following section, the role of alteration and weathering, fluid exsolution, fractional crystallization, and partial melting is presented.

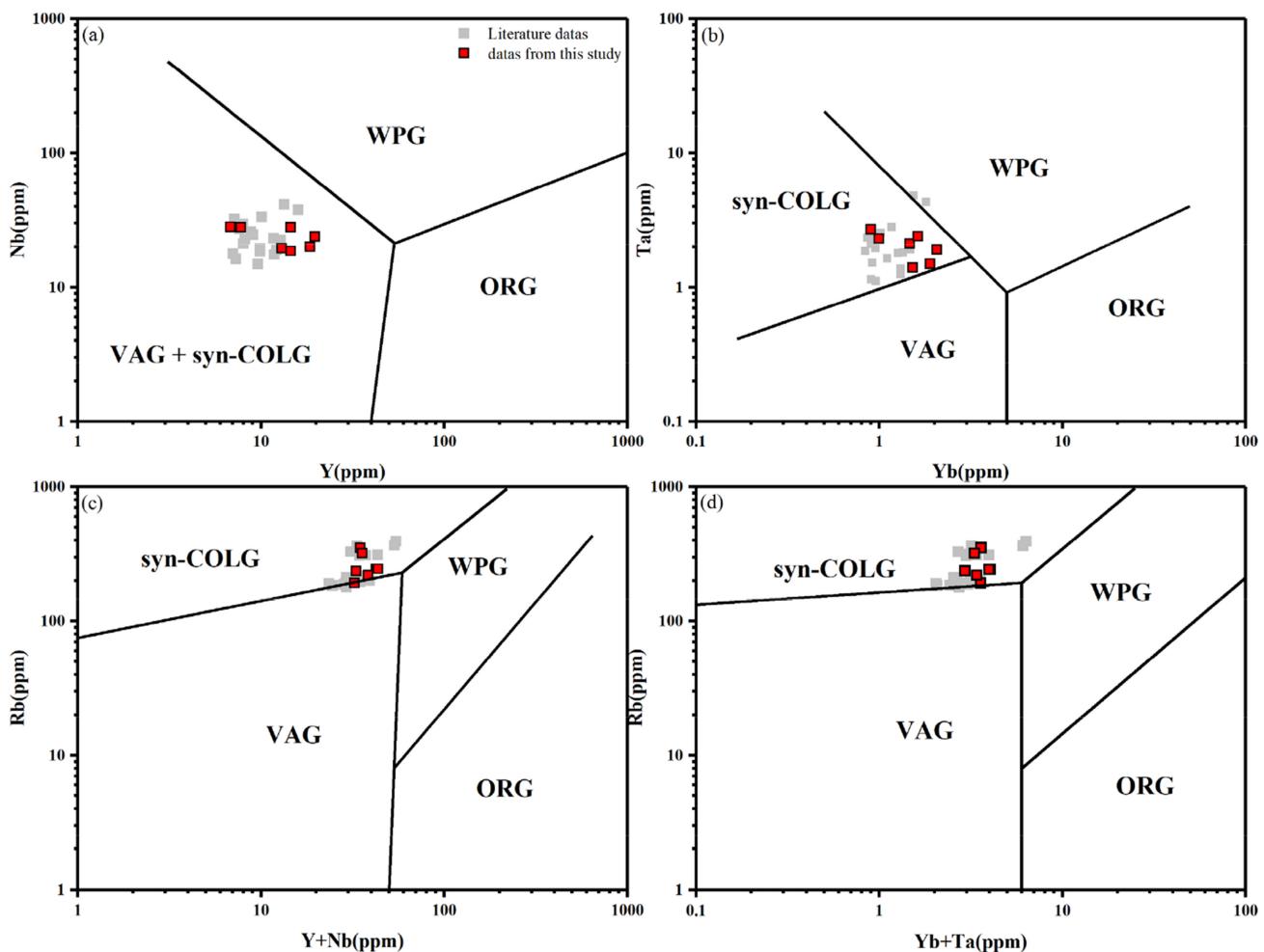


Fig. 6 Tectonic background discrimination diagram of the Xinxian pluton. WPG, intraplate granite. VAG, volcanic arc granite. ORG, oceanic ridge granite. and syn-COLG, syn-collisional granite

5.2.1 Alteration and weathering

Mineralogical observation (Fig. 3) showed that the feldspar and biotite in the samples were altered to different extents. Under the conditions of reductive weathering, the iron, decomposed by the weathering of primary minerals, can migrate in the form of Fe^{2+} , thus resulting in the enrichment of heavy iron isotopes (Liu et al. 2014; Poitrasson et al. 2008). However, under the conditions of oxidizing weathering, the free iron released by the weathering exists in the form of Fe^{3+} and leads to the presence of secondary minerals, without the loss of Fe or a change in iron isotopic composition (Liu et al. 2014). Notably, it is necessary to evaluate the influences of alteration and weathering on the Fe isotopic composition of the samples.

The LOI (loss of ignition) values of the samples were all lower than 1 wt.%. Moreover, there was no correlation between $\delta^{56}\text{Fe}$ and LOI, as shown in Fig. 7a. CIA (Al_2O_3 / (Al_2O_3 + Na_2O + K_2O + CaO) in a mole fraction) is the

chemical index of alteration, which can be used to indicate the degree of weathering (Nesbitt and Young 1982). All samples displayed low CIA values (Fig. 7b) and no sign of weathering was observed. Ba and U are active elements, while Th and Nb are inactive elements. With the progress of alteration, Ba and U should be lost in the first place, thus resulting in a decrease in the Ba/Th value and an increase in Nb/U. However, both indexes failed to present the trend of alteration. Therefore, it was determined that alteration and weathering may not be the mechanisms that resulted in the fractionation of Fe isotopes in the HFG.

5.2.2 Fluid exsolution

It has been established that the fluid in intrusive rock can be discharged in the form of steam or brine (Audétat and Pettke 2003; Baker and Alletti 2012). Heimann et al. (2008) reported that fluid exsolution occurs in the late stage of the magmatic process, thus resulting in the enrichment

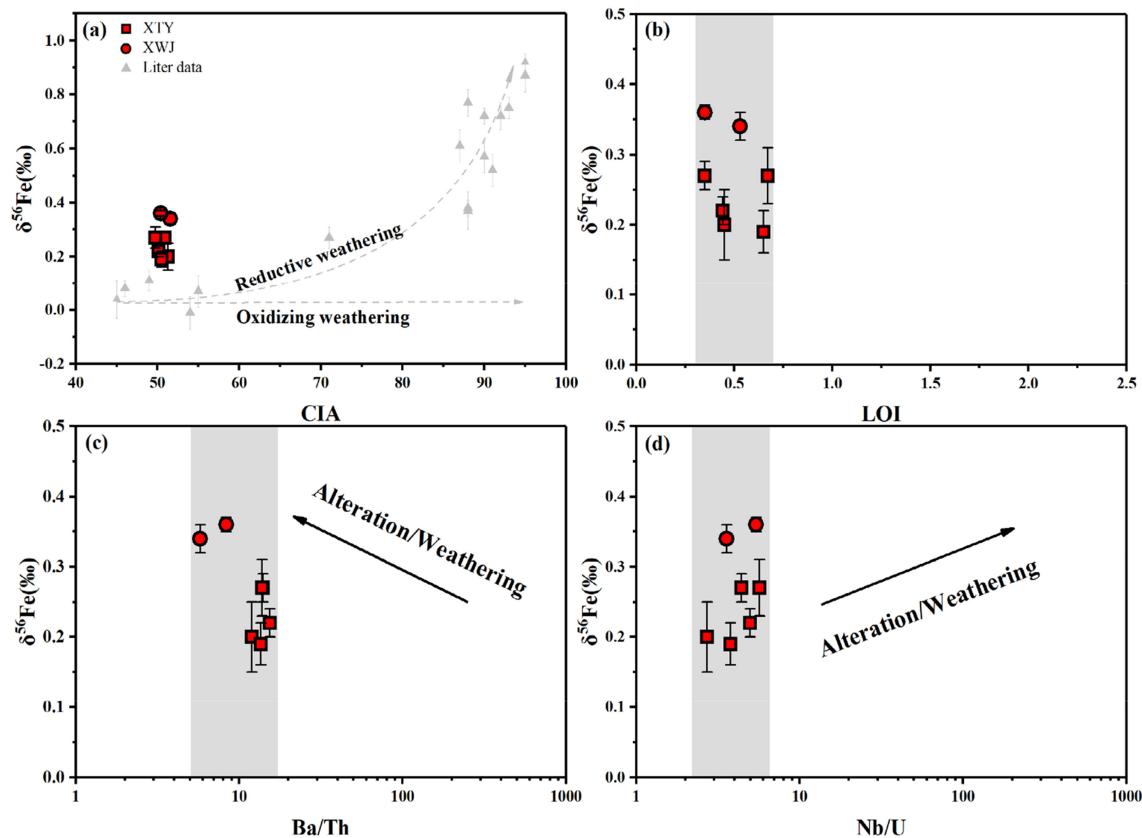


Fig. 7 $\delta^{56}\text{Fe}$ in HFG from Xinxian versus LOI (a), CIA (b), Ba/Th (c), Nb/U (d). The gray dotted line and literature data are from Liu et al. (2014). The gray dotted curve represents the reduced weathering trend, while the other dotted line represents the oxidizing weathering trend

of heavy iron isotopes in the melt by taking away the enriched light iron isotopes of Fe^{2+} .

The correlation between Zr/Hf, Nb/Ta, Rb/La, and $\delta^{56}\text{Fe}$ may provide constraints to the influences of fluid exsolution on iron isotope fractionation. The Zr/Hf ratios of igneous rock are generally less than 46 and higher than 26 (Bea et al. 2006), but significantly reduced after the exsolution of fluid because Zr is preferentially enriched in F-bearing fluid over Hf (Louvel et al. 2014). Previous studies (Ballouard et al. 2016) have argued that fluid exsolution occurred during the magmatic process, because the ratio of Nb/Ta is less than 5 (Du et al. 2019), as is also the case for Rb/La. The Th/U ratio used in this study is an index of fluid exsolution due to the higher mobility of U than Th in fluid (He et al. 2017).

The Zr/Hf ratios of two samples from XWJ were lower than 26, which may be explained by the effects of fluid exsolution, as shown in Fig. 8b. However, diagrams displaying another three parameters versus $\delta^{56}\text{Fe}$ show that the effects of fluid exsolution may be negligible. Both the XWJ granite and the XTY granite have Th/U values around the same levels as the mean lower continental crust ($\text{Th}/\text{U} \sim 5.9$, He et al. 2017) except for one sample (XTY-3-002, ~ 11.27). Recently, Du et al. (2019) argued that fluid

exsolution may produce $+ 0.07\ \text{‰}$ of $\delta^{56}\text{Fe}$ at most. Even if the high-level fluid exsolution occurred in the XTY or XWJ granite, the iron isotope fractionation of $\sim 0.35\ \text{‰}$ cannot be explained. Therefore, it was determined that another mechanism must be responsible for the enrichment of heavy iron isotopes in the Xinxian pluton.

5.2.3 Fractional crystallization and partial melting

The $\delta^{56}\text{Fe}$ of all the HFG samples increased with the increase in SiO_2 and $\text{Fe}^{3+}/\sum\text{Fe}$ and the decrease in Mg# and FeOt (Fig. 9), which suggested the influence of crystallization differentiation. The HFG trend can be modeled using the Rayleigh equation:

$$\delta^{56}\text{Fe}_{\text{melt}} = \delta^{56}\text{Fe}_{\text{melt0}} - \Delta^{56}\text{Fe}_{\text{melt-crystal}} \times \ln(f_{\text{Fe}}) \quad (2)$$

where f_{Fe} represents the Fe fraction in the remaining melt, and $\delta^{56}\text{Fe}_{\text{melt0}}$ refers to the value of $\delta^{56}\text{Fe}$ in the initial melt. XTY-4-002 was chosen on behalf of the composition of the initial melt to calculate the fractional factor of $\Delta^{56}\text{Fe}_{\text{melt-crystal}}$ due to its high FeOt content and low value of $\delta^{56}\text{Fe}$. The results of the modeling are presented in Fig. 9b, which shows that the trend of fractional crystallization of

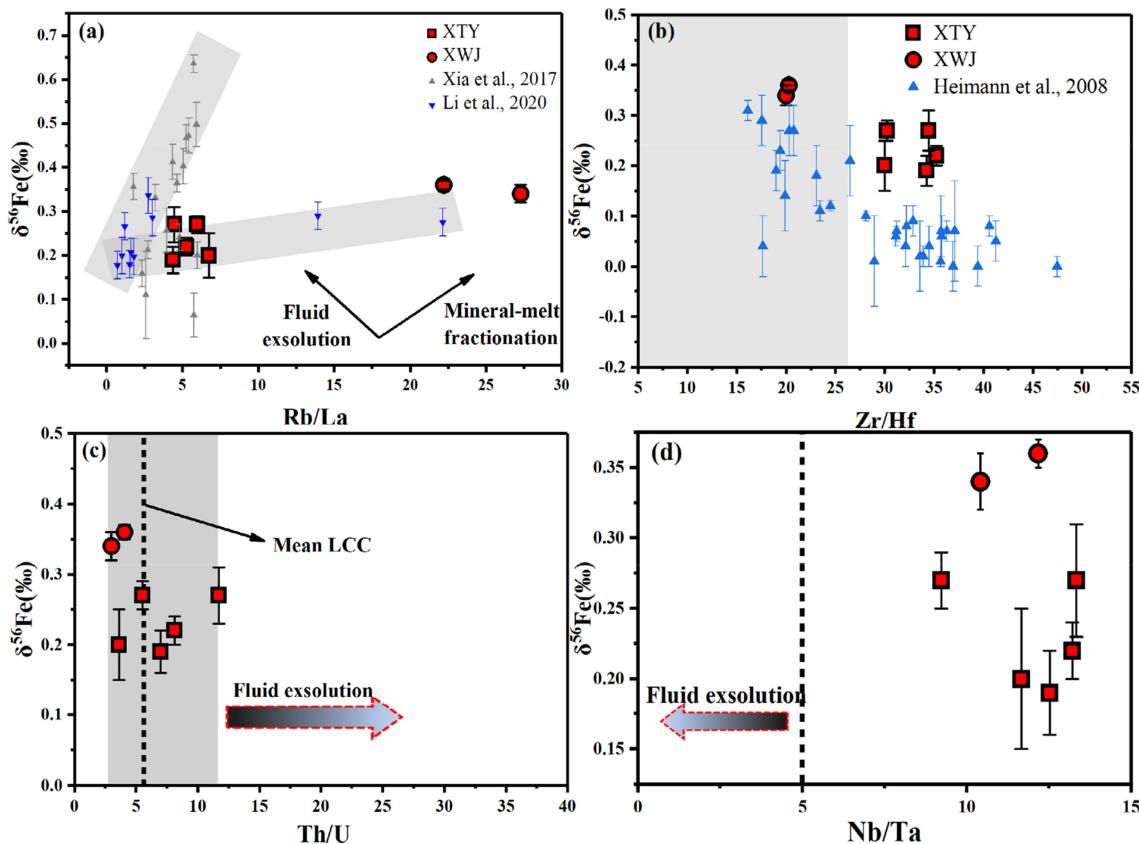


Fig. 8 $\delta^{56}\text{Fe}$ in HFG from Xinxian versus Rb/La (a), Zr/Hf (b), Th/U (c), Nb/Ta (d). The samples with Zr/Hf less than 26 in b tend to be affected by the dissolution of fluid. The gray field in a is the trend of $\delta^{56}\text{Fe}$ versus Rb/La ratios from Xia et al. (2017) and Li (2020), which is identical to the trend of HFG from the Xinxian pluton

HFG in the Xinxian pluton can be explained by $\Delta^{56}\text{Fe}_{\text{melt-crystal}} = \sim 0.15\text{\textperthousand}$.

It has been reported in previous studies (Heimann et al. 2008; Telus et al. 2012; Sossi et al. 2012; Wu et al. 2017a, b) that olivine, pyroxene, hornblende, biotite, and Fe-Ti oxides are enriched with light iron isotopes, while feldspar and magnetite are enriched with heavy iron isotopes.

Olivine and pyroxene are barely crystallized from felsic magmas, which means that the crystallization of the two minerals may not drive evident Fe isotope fractionation. Owing to the enrichment of middle REEs in amphiboles, the separation of an amphibole may lead to the U-shaped REE pattern (Bachmann et al. 2005). In this study, an obvious Eu negative anomaly can be observed with a slightly right-inclining shape, excluding the effects on Fe isotope fractionation by the crystallization of the amphiboles.

The iron isotopic composition of feldspar is significantly heavier than those of other rock-forming minerals. Wu et al. (2017a, b) reported the $\delta^{56}\text{Fe}$ of migmatites that are formed by the feldspar accumulation in Dabie ranged from $0.424 \pm 0.016\text{\textperthousand}$ to $0.567 \pm 0.016\text{\textperthousand}$. Evidence from

petrological observation showed that there was no sign of the accumulation of feldspar (Fig. 3). Furthermore, the accumulation of feldspar would produce a positive correlation between $\delta^{56}\text{Fe}$ value and Sr and Ba content as well as the Eu/Eu* ratio. Such correlation was not displayed in Fig. 10, so it was determined that the accumulation of feldspar may not account for the fractionation of iron isotopes in XWJ.

The crystallization of titanomagnetite, containing titanium spinel, might be the key factor affecting the Fe isotopic composition. Furthermore, the coefficient between titanomagnetite and the melt is $\sim -0.1\text{\textperthousand}$ (Schuessler et al. 2009). In conclusion, the crystallization of Fe-Ti oxide plays a significant role in the Fe isotopic composition of the residual melt, which has led to the enrichment of heavy Fe isotopes in the residual melt. The positive correlation between FeOt and TiO₂ in Fig. 11d shows the fractional crystallization of Fe-Ti oxide and the enrichment of heavy Fe isotopes in the residual melt.

Xu et al. (2017) suggested that FeOt/Al₂O₃ could be considered the enrichment index of feldspar for magmatic systems. As shown in Fig. 11a, the content of biotite in the samples was between 3 and 10%, and the Fe isotope

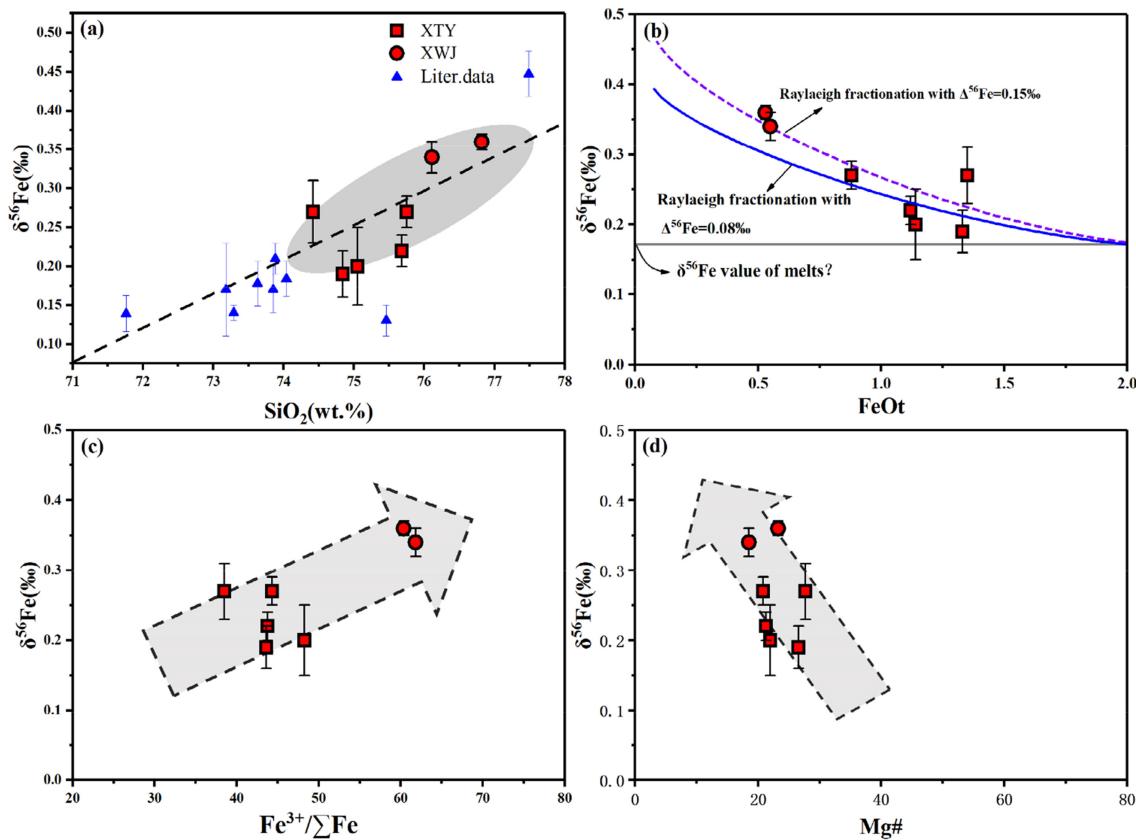


Fig. 9 $\delta^{56}\text{Fe}$ in HFG from Xinian versus SiO_2 (a), FeOt (b), $\text{Fe}^{3+}/\sum \text{Fe}$ (c), $\text{Mg}\#$ (d). The literature data in a are from He et al. (2017) and Foden et al. (2015). The two curves in b are modeled by Rayleigh equation

composition was significantly heavier than the estimated curve. We hold the view that this is probably due to the influence of the melt composition.

The magma evolves towards acidity with an increase in SiO_2 , while the alkali elements (Na and K) in feldspar tend to be enriched. The enrichment of these alkali metals, resulting in Fe^{3+} in the melt, becomes more stable in the tetra-coordination (Métrich et al. 2006; Guili et al. 2012), which leads to the accumulation of heavy iron isotopes. The positive correlation between $\delta^{56}\text{Fe}$ and $(\text{Na} + \text{K})/(\text{Ca} + \text{Mg})$ (in mole fraction), as presented in Fig. 11b, shows that the variable $\delta^{56}\text{Fe}$ from the HFG can be attributed to the magma composition, given that the HFG fall within the global HSG trend, thus leading to changes in $\Delta^{56}\text{Fe}_{\text{melt-crystal}}$.

Notably, once the composition of the melt changes, the composition of the mineral will change accordingly. Wu et al. (2017a, b) argued that the change in mineral composition would significantly affect the fractionation of iron isotopes between minerals. Wu et al. (2017a, b) created the equation between $\Delta^{56}\text{Fe}_{\text{Plagioclase-Biotite}}$, $\Delta^{56}\text{Fe}_{\text{Alkali-feldspar-Biotite}}$, $\Delta^{56}\text{Fe}_{\text{Plagioclase-Magnetite}}$, $\Delta^{56}\text{Fe}_{\text{Plagioclase-Magnetite}}$, $\Delta^{56}\text{Fe}_{\text{Alkali-feldspar-Magnetite}}$, and the content of Ab and Or:

$$\Delta^{56}\text{Fe}_{\text{Plagioclase-Biotite}} = 0.015 \times \text{Ab} - 0.48 \quad (3)$$

$$\Delta^{56}\text{Fe}_{\text{Alkali-feldspar-Biotite}} = 0.026 \times \text{Or} - 0.48 \quad (4)$$

$$\Delta^{56}\text{Fe}_{\text{Plagioclase-Magnetite}} = 0.022 \times \text{Ab} - 1.15 \quad (5)$$

$$\Delta^{56}\text{Fe}_{\text{Alkali-feldspar-Magnetite}} = 0.026 \times \text{Or} - 1.46 \quad (6)$$

The $\Delta^{56}\text{Fe}$ result according to this calculation is shown in Table 4. The calculated results of $\Delta^{56}\text{Fe}$ between feldspar and biotite/magnetite demonstrate that the change in the mineral composition may not be responsible for the iron isotopic composition of HFG, because the measured value of $\delta^{56}\text{Fe}$ is significantly heavier than the calculated result of $\Delta^{56}\text{Fe}$.

The effects of partial melting on iron isotopes are complex and involve oxygen fugacity and the degree of melting (Zhu et al. 2016). Xu et al. (2017) discovered that partial melting can trigger the fractionation of iron isotopes, averaging $0.093 \pm 0.056\text{\textperthousand}$ between leucosome (product of partial melting) and melanosome (residual of partial melting) in migmatite under the conditions of $P = \text{ca. } 5.0 \text{ kbar}$, $T = 705\text{--}744^\circ\text{C}$. Previous studies have shown that the source rocks of the Xinian HFG were Neoproterozoic TTG magmatic rocks from the north margin of the Yangtze block and were derived from the middle

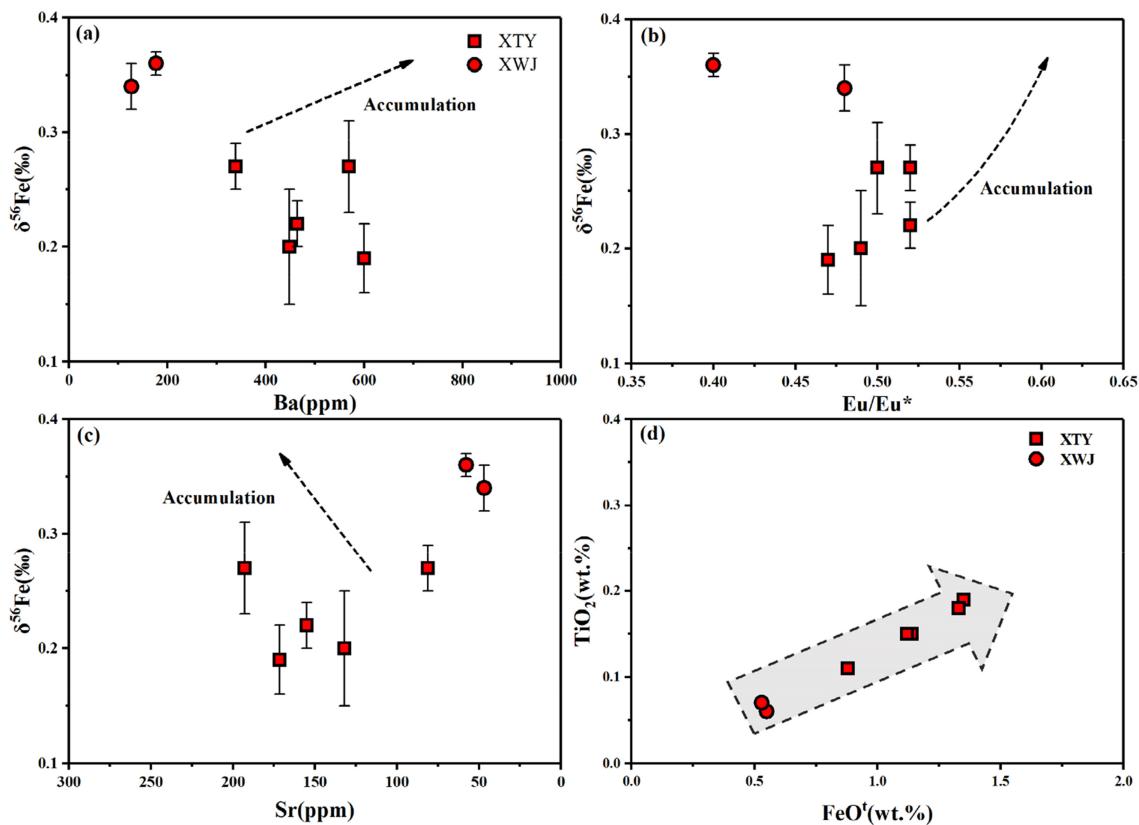


Fig. 10 $\delta^{56}\text{Fe}$ in HFG from Xinxian versus Ba (a), Eu/Eu* (b), Sr (c); diagram of TiO_2 vs. FeOt (d)

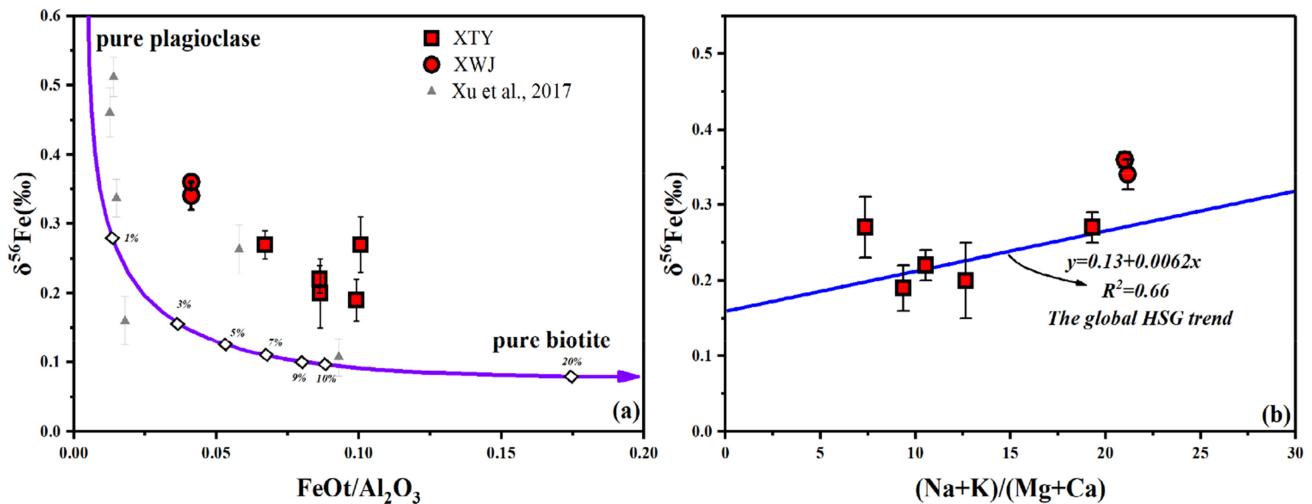


Fig. 11 $\delta^{56}\text{Fe}$ in HFG from Xinxian versus $\text{FeOt}/\text{Al}_2\text{O}_3$ (a), $(\text{Na} + \text{K})/\text{Ca} + \text{Mg}$ (b). The curve in a is from Xu et al. (2017), and the percentage numbers denote the fraction of biotite. The global HSG in b is from He et al. (2017)

and lower crust with a low temperature ($< 800^\circ\text{C}$) and a depressurized environment (Chen et al. 2013a; Zhao et al. 2013; Yang et al. 2020). Moreover, Li (2020) reported the iron isotopic composition of Early and late Neoarchean TTG type magmatic rocks, and the results illustrated that the values of ^{56}Fe in TTG magmatic rocks with different

ages and regions are similar, ranging from $0.075\text{\textperthousand}$ to $0.201\text{\textperthousand}$, with an average of $0.142 \pm 0.07\text{\textperthousand}$. In that case, partial melting may also lead to iron isotopic fractionation before the fractional crystallization. However, it is difficult to estimate the influences of partial melting using our current evidence.

Table 4 Calculation results of $\Delta^{56}\text{Fe}$ between feldspar and dark minerals

	XTY-1-001	XTY-2-001	XTY-3-002	XTY-3-003	XTY-4-002	XWJ-001	XWJ-003	Average
$\Delta^{56}\text{Fe}_{\text{Plg-Bi}}$	0.046	0.003	0.013	0.041	0.016	0.048	0.062	0.033
$\Delta^{56}\text{Fe}_{\text{Afs-Bi}}$	-0.616	-0.601	-0.63	-0.667	-0.608	-0.647	-0.661	-0.633
$\Delta^{56}\text{Fe}_{\text{Plg-Mgn}}$	-0.378	-0.441	-0.426	-0.385	-0.421	-0.374	-0.353	-0.397
$\Delta^{56}\text{Fe}_{\text{Afs-Mgn}}$	-0.726	-0.711	-0.74	-0.777	-0.718	-0.757	-0.771	-0.743

6 Conclusions

- (1) Major elements and mineralogical observations showed that the early Cretaceous granite in the Xinian pluton belongs to the highly fractionated I-type granite.
- (2) Further investigation demonstrated that the effect of fluid exsolution, along with alteration and weathering, on the Fe isotopes in HFG is negligible.
- (3) The results modeled using the Rayleigh equation showed that fractional crystallization may produce $\Delta^{56}\text{Fe}_{\text{melt-crystal}} = 0.08 \text{‰} \sim 0.15 \text{‰}$. The controlling factor that contributes to the Fe isotopic composition of our sample might be the fractional crystallization of Fe-Ti oxides.
- (4) The change in melt composition may also lead to the enrichment of heavy Fe isotopes in the residual melt, while the role of partial melting is consistent with the current observations.

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Declarations

Conflict of interest The authors declare that they have no conflict of interest.

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