

Elemental characteristics and paleoenvironment reconstruction: a case study of the Triassic lacustrine Zhangjitan oil shale, southern Ordos Basin, China

Delu Li¹ · Rongxi Li¹ · Zengwu Zhu² · Xiaoli Wu¹ · Futian Liu¹ ·
Bangsheng Zhao¹ · Jinghua Cheng¹ · Baoping Wang³

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Abstract Using trace elements to reconstruct paleoenvironment is a current hot topic in geochemistry. Through analytical tests of oil yield, ash yield, calorific value, total sulfur, major elements, trace elements, and X-ray diffraction, the quality, mineral content, occurrence mode of elements, and paleoenvironment of the Zhangjitan oil shale of the Triassic Yanchang Formation in the southern Ordos Basin were studied. The analyses revealed relatively high oil yield (average 6.63%) and medium quality. The mineral content in the oil shale was mainly clay minerals, quartz, feldspar, and pyrite; an illite-smectite mixed layer comprised the major proportion of clay minerals. Compared with marine oil shale in China, the Zhangjitan oil shale had higher contents of quartz, feldspar, and clay minerals, and lower calcite content. Silica was mainly in quartz and Fe was associated with organic matter, which is different from marine oil shale. The form of calcium varied. Cluster analyses indicated that Fe, Cu, U, V, Zn, As, Cs, Cd, Mo, Ga, Pb, Co, Ni, Cr, Sc, P, and Mn are associated with organic matter while Ca, Na, Sr, Ba, Si, Zr, K, Al, B, Mg, and Ti are mostly terrigenous. Sr/Cu, Ba/Al, V/(V + Ni), U/Th, AU, and δU of oil shale samples suggest the paleoclimate was warm and humid, paleoproductivity of the lake was relatively high during deposition of the shale—which mainly occurred in fresh water—and the paleo-redox condition was dominated by reducing conditions. Fe/Ti ratios of the oil shale samples suggest clear

hydrothermal influence in the eastern portion of the study area and less conspicuous hydrothermal influence in the western portion.

Keywords Trace elements · Occurrence mode · Paleoenvironment · Zhangjitan oil shale · Yanchang Formation · Ordos Basin

1 Introduction

Oil shale, as one of the most important unconventional petroleum resources, has attracted much attention in recent years (Liu and Liu 2005; Lü et al. 2015; Hakimi et al. 2016; Song et al. 2016). In general, shale with over 3.5% oil yield is defined as oil shale, and can release thermal energy and shale oil by low temperature carbonization (Fu et al. 2015; Liu et al. 2015). Prior research on oil shale has mainly focused on the marine environment, especially in northern China—including the Cretaceous oil shale in Qiangtang Basin (Fu et al. 2007, 2010a, b, 2015, 2016; Wang et al. 2010), middle-upper Neoproterozoic oil shale in Yanshan Region (Bian et al. 2005; Liu et al. 2011), etc. Based on elemental geochemistry, the oil shale in Qiangtang Basin was deposited in a tropical-subtropical environment (Fu et al. 2009; Wang et al. 2010). The Qiangtang oil shale formed at the edge of a gulf-lagoon, with fresh water input and high paleoproductivity (Fu et al. 2007; Wang et al. 2010); petrography and geochemistry indicate that hydrothermal fluid has impacted its formation (Sun et al. 2003; Chen and Sun 2004; Chen et al. 2004). Major and trace element analyses suggest hydrothermal activity was frequent during sedimentation of the Xiamaling Formation oil shale in the middle-upper Neoproterozoic (Sun et al. 2003). Lacustrine oil shale is widely distributed

✉ Rongxi Li
Rongxi99@163.com

¹ School of Earth Sciences and Resources, Chang'an University, Xi'an 710054, China

² Shaanxi Center of Geological Survey, Xi'an 710068, China

³ Yanchang Oilfield Co.,Ltd, Yan'an 716000, China

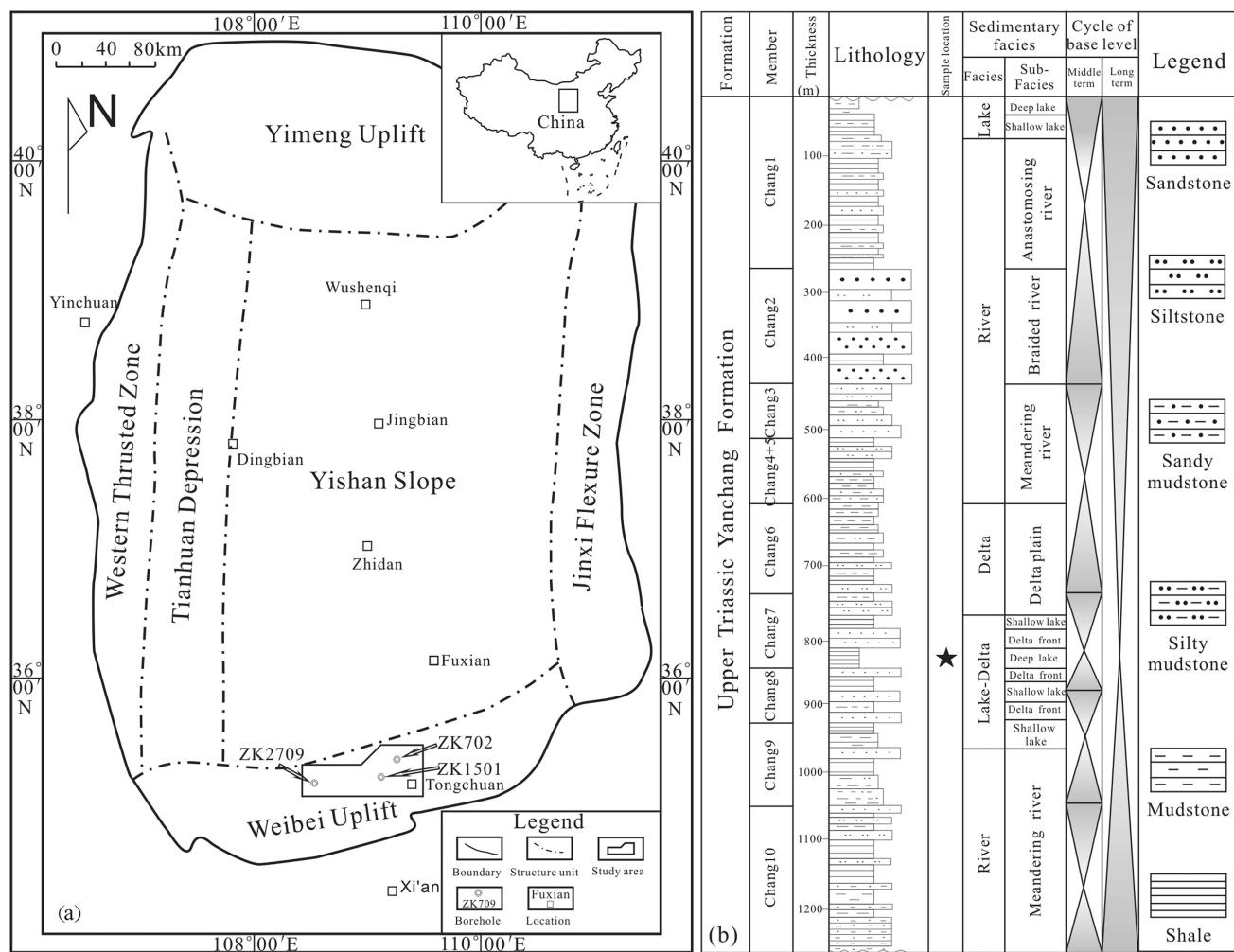


Fig. 1 **a** Geologic map of the Ordos Basin and location of studied section (Li et al. 2016) and **b** stratigraphic column of Upper Triassic Yanchang Formation in study area (modified after Qiu et al. 2014)

across China (Bai and Wu 2006; Bai et al. 2015; Liu et al. 2006). However, most previous studies have concentrated on calculating reserves (Bai and Wu 2006; Bai et al. 2009; Lu et al. 2006). Elemental and mineral characteristics of lacustrine oil shales, and implications for paleoenvironment, are badly in need of additional investigation.

A typical lacustrine oil shale in China, the Zhangjitan oil shale of the Triassic Yanchang Formation in the southern Ordos Basin has many merits, such as wide distribution, abundance, and shallow burial (Lu et al. 2006; Bai et al. 2009; Li et al. 2009a, b, 2014; Chang et al. 2012; Wang and Yan, 2012; Deng et al. 2013; Luo et al. 2014). Organic geochemistry analyses show that the Zhangjitan oil shale in Chang 7 Member of the Yanchang Formation is dominated by organic matter type II_I and approaches mature designation (Liu et al. 2009; Ma et al. 2016; Wei et al. 2016). However, few studies have focused on elemental geochemistry of the oil shale, particularly occurrence of trace elements. Although faunal

data, frambooidal pyrite, and the ratio of organic carbon to total phosphorus in the oil shale indicate deposition was dominated by oxidizing conditions (Yang et al. 2010; Yuan et al. 2016), biomarkers have shown reducing conditions (Deng et al. 2013). Additionally, discussions of oil shale paleosalinity are still unsettled (Luo et al. 2014), with a possible marine transgression event in the southern Ordos Basin. Elemental geochemistry of some oil shale outcrop samples, focusing on Tongchuan City in southern Ordos Basin, has improved understanding of the paleoenvironment (Sun et al. 2015). However, two concentrated oil shale samples from one location do not necessarily reflect the entire southern Ordos Basin as weathering and alteration can significantly impact elemental characteristics of samples.

In order to improve understanding, we evaluated the quality of oil shale and the occurrence mode of trace elements. Then, by using the specific or calculated value, paleoenvironment was reconstructed. Finally, relationships

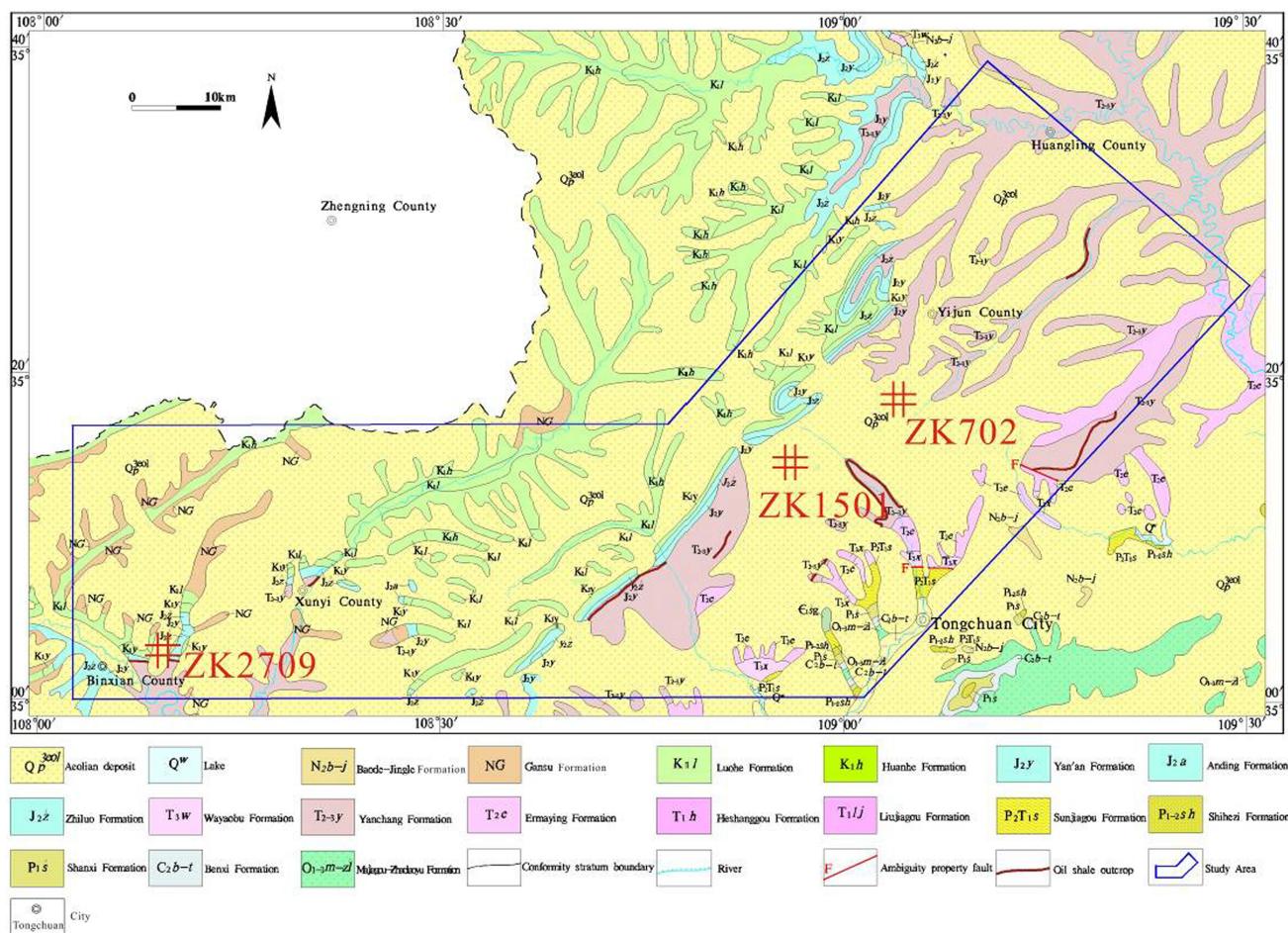


Fig. 2 Simplified geologic map of study area, showing the location of drill holes (modified after Ma et al. 2016)

between single elements and paleoenvironment were analyzed to determine the indicators of paleoenvironment of the lacustrine oil shale. This study fills gaps of lacustrine oil shale elemental geochemistry in the southern Ordos Basin and should help guide future exploration.

2 Geologic setting

The Ordos Basin is a superimposed basin with stable deposition and multiple sedimentary cycles (Fig. 1a) located in mid-western China (Liu et al. 2008). The Ordos formed as a marine basin of the North China Block by the Carboniferous to Permian and has a Proterozoic crystalline basement (Wan et al. 2013; Qiu et al. 2015). After the Triassic, the basin gradually departed from the North China Block and evolved into a large inland sedimentary basin with a relatively quiet tectonic setting (Li et al. 2006, 2008). Affected by the Indosinian Orogeny, the whole basin gradually uplifted and subducted in the Early Cretaceous and underwent reformation. According to

present tectonic characteristics, basement features and evolutionary history, the basin is divided into six first-order tectonic units: the Weibei Uplift, Yishan Slope, Yimeng Uplift, Jinxi Flexure Zone, Tianhuan Depression, and Western Thrusted Zone. The Triassic Yanchang Formation went through an integrated occurrence-extinction period and deposited a set of progradation aggradational retrogradation strata with a thickness of 1000–1300 m (Wu et al. 2004; Li et al. 2009a, b; Zou et al. 2012). According to sedimentary cycles and rock assemblages, the Yanchang Formation can be further divided into 10 Members: Chang 10 to Chang 1 (Qiu et al. 2010). During the initial stage of Chang 7, due to extension and sinking of the basin along with orogenies and paroxysmal eruption in the south, the lake reached its largest size (Qiu et al. 2015). There are two sets of oil shale in the Yanchang Formation, namely the Zhangjitan at the bottom of Chang 7 with a thickness of 20–30 m (Wang and Yan 2012) and the Lijiapan (Wang and Yan 2012; Dong et al. 2014) at the top of Chang 9 with a thickness of 5–15 m. The Zhangjitan is the main Mesozoic oil reservoir in the Ordos Basin (He 2003; Deng et al. 2013; Li et al. 2016).

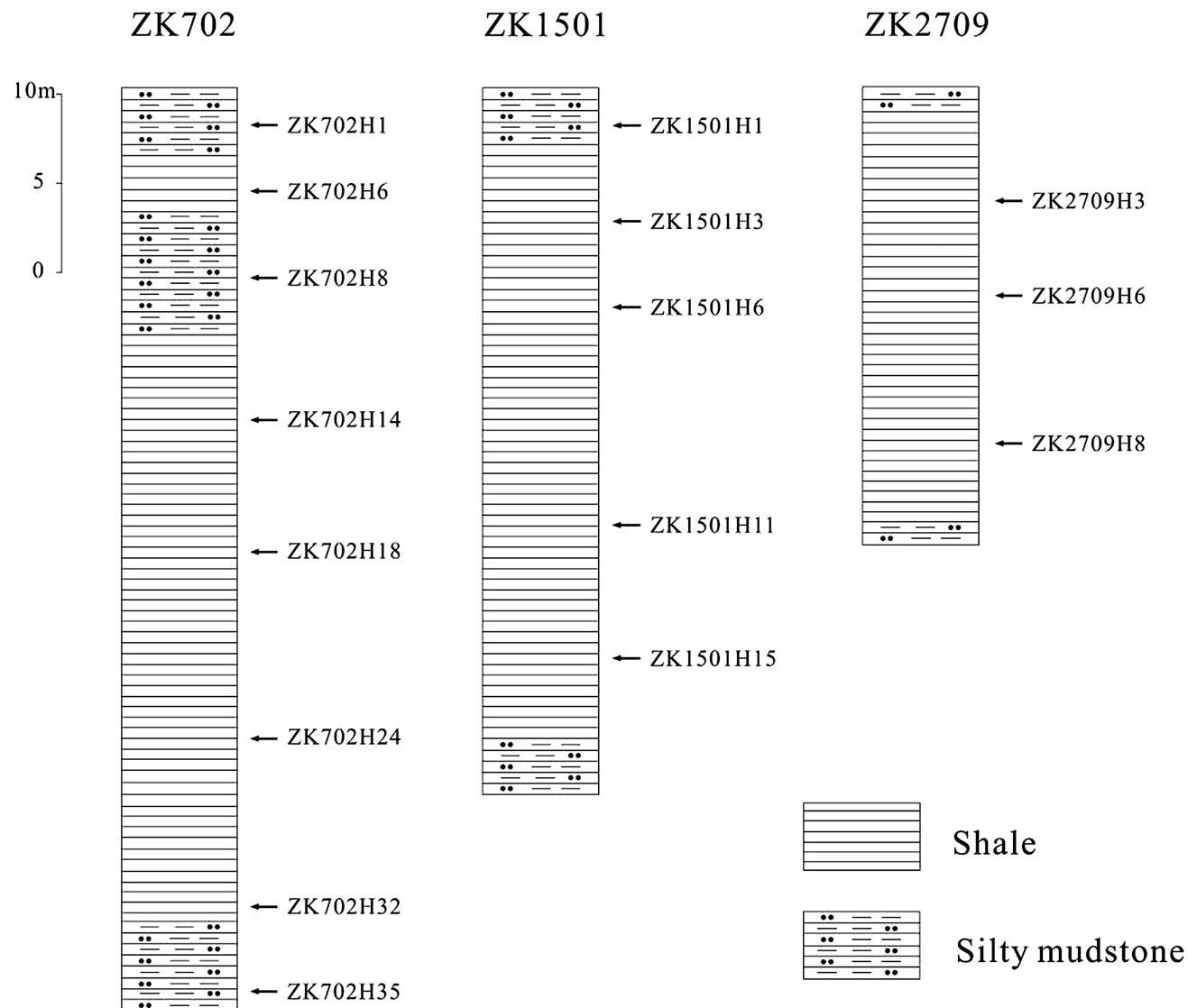


Fig. 3 Zhangjiatan oil shale sections, showing sampling locations

The study area is located in the Weibei Uplift, southern Ordos Basin. Due to the Indosinian Orogeny in the Late Triassic and the later Yanshan Orogeny, the study area experienced unbalanced uplift and lacks upper Yanchang strata. The bottom of Chang 7 contains oil shale, shale, mudstone, and silty mudstone (Fig. 1b). Shallow burial conditions and the significant thickness of the Zhangjiatan make it economically viable. Understanding the Zhangjiatan may improve oil shale exploration.

3 Samples and analytical methods

A total of 16 samples were collected from three oil shale sections in the lower Chang 7 Member in the southern Ordos Basin (Figs. 2, 3). All samples were analyzed for oil

yield (T_{ad}), ash yield (A_d), calorific value ($Q_{b,ad}$), total sulfur ($S_{t,d}$), major element oxides, and trace elements. Four samples were tested by X-ray diffraction (XRD) for mineral content.

For the oil yield analysis, samples were ground to a particle size of less than 3 mm, then 50 g of each sample was enclosed in aluminum retort by low temperature carbonization for analysis. The procedure followed the Chinese standard methods SH/T 0508-1992(1992).

The ash content analysis used the slow-ashing method. Each sample of about 1 g was ground to less than 0.2 mm and tiled to a cupel. Then the cupel was heated to $815 \pm 10^\circ\text{C}$ in a muffle furnace until the residue content stabilized. We followed the Chinese standard methods GB/T 212-2008 (2008a, b). Precision was within 5%.

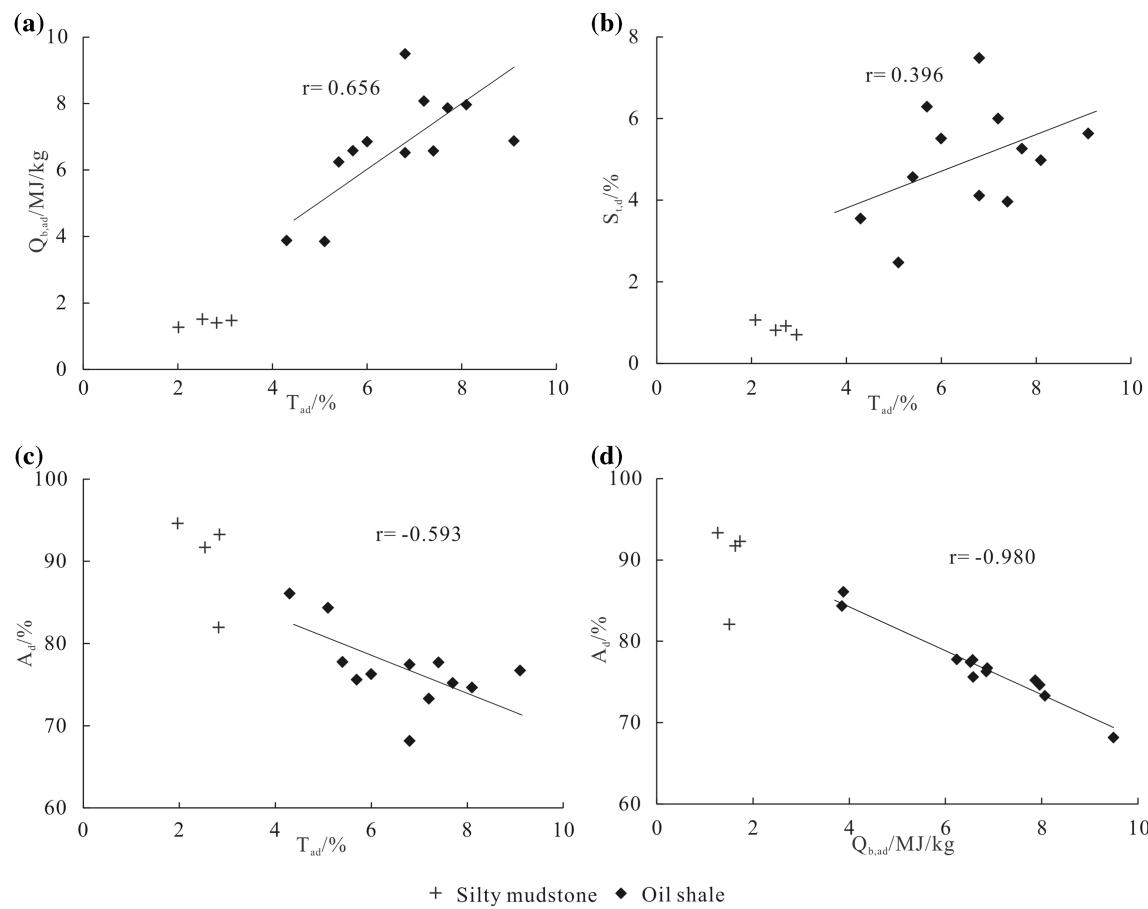


Fig. 4 Relationships between key parameters of oil shale and silty mudstone samples

Table 1 Mineral content tested by XRD in oil shale samples (%)

Sample no.	Quartz	Feldspar			Calcite	Pyrite	Siderite	Ankerite	Gypsum	Jarosite	Clay minerals
		Potassium feldspar	Anorthose	Buddingtonite							
ZK702H14	24.7		9.6	6.2	3.7	19.1			0.4	7.9	28.4
ZK702H24	31.0	9.0	10.0			5.0		4.0			41.0
ZK1501ZH6	24.0	3.0	11.0		5.0	7.0	1.0	3.0			46.0
ZK2709H3	26.0	6.0	10.0		1.0		7.0		8.0		42.0

For the determination of calorific value, samples of 1 g with grain size below 0.2 mm were put into a combustion boat and ignited electrically by oxygen with a purity of more than 99.5%. After 6–7 min, temperatures were recorded for 3 min at 1-min intervals and the highest temperature was recorded as the final temperature. The analytical method followed the Chinese National Standard GB/T 213-2008(GB/T 2008a, b). Analytical error was within 5%.

The samples for total sulfur analysis were powdered to less than 100 μm and heated in a pipe furnace to $1250 \pm 20^\circ\text{C}$ with fluxing agent of cupric oxide powder.

The method followed Chinese National Standard GB/T 6730.17-2014 (2014).

For mineral content analysis, XRD was performed with a D8 ADVANCE powder diffractometer. The analytical procedures followed Chinese National Standard SY/T 6210-1996(1996) and the precision was within 1%.

The above analyses were conducted at Shaanxi Coal Geological Laboratory Co., Ltd.

The samples for element analysis were all powdered to less than 200 mesh, and analyzed by X-ray fluorescence spectrometry (XRF) for major elements and inductively coupled plasma–mass spectrometry (ICP-MS) for trace

Fig. 5 Zhangjiatan oil shale outcrops of Bawangzhuang Profile in Tongchuan City, southern Ordos Basin

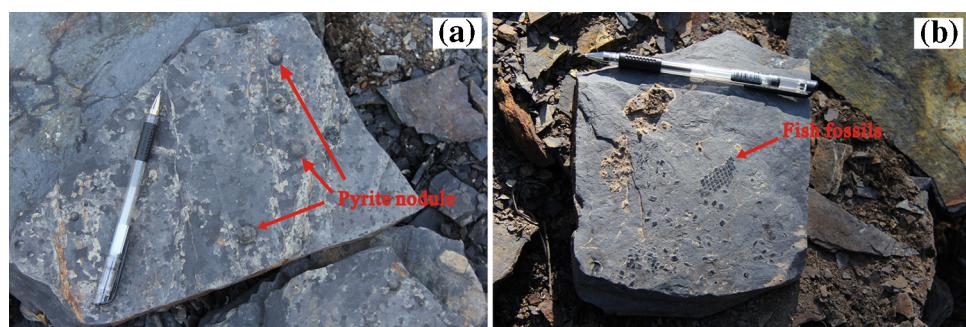


Table 2 Clay mineral content in oil shale samples (units in %)

Sample no.	Smectite	Illite	Kaolinite	Chlorite	Illite-smectite mixed layer	Chlorite-smectite mixed layer
ZK702H14	0	30	0	1	69	0
ZK702H24	0	33	3	4	60	0
ZK1501ZH6	0	38	5	2	55	0
ZK2709H3	0	14	4	3	79	0

Table 3 Oil yield, calorific value, ash yield, total sulfur, and major elements in oil shale samples (units of $Q_{b,ad}$ are MJ/kg; others are %)

Sample no.	Lithology	T_{ad}	$Q_{b,ad}$	A_d	$S_{t,d}$	SiO_2	Al_2O_3	MgO	CaO	Na_2O	K_2O	P_2O_5	MnO	TiO_2	TFe_2O_3
ZK702H1	Silty mudstone	2.10	1.22	94.39	1.03	68.10	13.06	0.66	1.16	2.32	2.96	0.07	0.03	0.19	1.56
ZK702H6	Oil shale	4.30	3.88	86.08	3.55	55.73	13.54	0.75	0.42	0.98	3.61	0.24	0.09	0.29	4.51
ZK702H8	Silty mudstone	3.00	1.63	93.08	0.53	66.39	12.91	0.60	1.52	1.69	3.04	0.11	0.17	0.19	0.90
ZK702H14	Oil shale	7.20	8.07	73.28	6.00	40.61	13.37	0.97	0.97	0.91	2.25	0.28	0.05	0.40	8.44
ZK702H18	Oil shale	9.10	6.87	76.70	5.63	46.90	10.69	0.86	1.61	1.30	1.85	0.45	0.06	0.36	7.84
ZK702H24	Oil shale	8.10	7.96	74.63	4.98	43.33	12.50	0.98	1.67	1.49	1.49	0.35	0.07	0.33	6.81
ZK702H32	Oil shale	7.70	7.87	75.21	5.26	44.26	12.66	0.93	1.25	1.00	2.37	0.29	0.09	0.34	7.92
ZK702H35	Silty mudstone	2.50	1.56	92.00	0.70	59.95	12.44	1.64	4.56	2.20	4.11	0.17	0.11	0.41	1.22
ZK1501H1	Silty mudstone	2.80	1.50	82.82	0.80	67.38	13.14	0.57	1.25	2.47	2.90	0.11	0.09	0.21	1.42
ZK1501H3	Oil shale	5.70	6.58	75.60	6.29	45.36	11.45	1.06	1.62	1.45	2.12	0.36	0.11	0.39	7.96
ZK1501H6	Oil shale	6.80	9.49	68.15	7.48	36.42	11.37	1.45	2.19	0.74	1.90	0.26	0.08	0.35	9.03
ZK1501H11	Oil shale	6.00	6.85	76.27	5.51	43.58	12.96	1.16	2.36	1.43	2.03	0.41	0.21	0.45	7.43
ZK1501H15	Oil shale	5.40	6.24	77.76	4.56	48.09	14.00	1.22	1.98	1.30	2.36	0.46	0.25	0.39	5.77
ZK2709H3	Oil shale	7.40	6.57	77.71	3.96	44.44	14.15	1.71	1.59	1.29	2.53	0.39	0.19	0.51	6.12
ZK2709H6	Oil shale	6.80	6.52	77.44	4.11	46.51	13.44	1.76	2.09	1.39	2.23	0.38	0.19	0.52	5.40
ZK2709H8	Oil shale	5.10	3.85	84.33	2.47	50.17	15.92	2.01	1.17	1.14	2.80	0.27	0.16	0.63	3.73

T_{ad} = oil yield; $Q_{b,ad}$ = calorific value; A_d = ash yield; $S_{t,d}$ = total sulfur

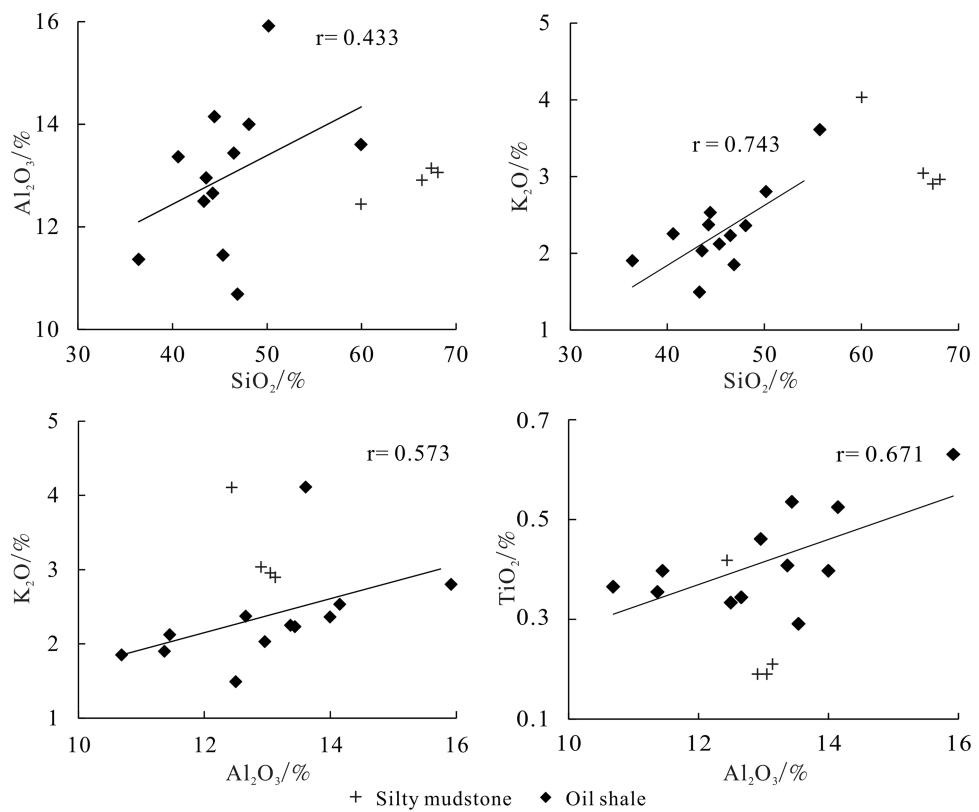
elements with AA-6800 atomic absorption spectroscopy, UV-2600 ultraviolet-visible spectrophotometer and Perkin Elmer SciexElan 6000. The analytical procedures followed Chinese National Standard GB/T 14506.1 ~ 14-2010(2010) and GB/T 14506.30-2010(2010). Analytical precision was within 5%. The analyses were conducted at the Analytical Center, No. 203 Research Institute of Nuclear Industry.

4 Results

4.1 The industrial quality of shale

In resource assessment of oil shale, four key parameters (oil yield, calorific value, ash yield, and total sulfur) are always considered (Liu et al. 2009). Oil yields of oil shale samples are in the range of 4.3%–9.1% (average 6.63%)

Fig. 6 Relationships among SiO_2 , Al_2O_3 , K_2O , and TiO_2 concentrations in oil shale and silty mudstone samples



and calorific values are from 3.85 to 9.49 MJ/kg (average 6.73 MJ/kg). The better the quality, the higher the values of the two parameters (Liu et al. 2009; Zhang et al. 2013; Sun et al. 2015). Ash yields of oil shale samples range from 68.15% to 86.08% (average 76.93%) and total sulfur from 2.47% to 7.48% (average 4.98%). The better the quality, the lower the value of these two parameters (Liu et al. 2009; Zhang et al. 2013; Sun et al. 2015). We found that oil yield correlated positively with calorific value (Fig. 4a) and total sulfur (Fig. 4b) and negatively with ash yield (Fig. 4c). Meanwhile, superb negative correlation was observed between calorific value and ash yield (Fig. 4d). According to National Resource Assessment of Oil shale (Dong et al. 2006), oil shale is classified as low, middle, and high grade by oil yield of 3.5%–5%, 5%–10%, and >10%, respectively. Thus, the oil shale in the southern Ordos Basin is middle grade.

4.2 Minerals in oil shale

The minerals identified by XRD in oil shale samples were clay minerals, quartz, feldspar, and pyrite (>5%) with minor amounts of calcite, siderite, ankerite, gypsum, and jarosite (Table 1). Pyrite was mostly in the form of nodules (Fig. 5a). For clay minerals, mixed-layer illite/smectite was predominant, followed by illite (Table 2). Kaolinite and

chlorite were measured in trace amounts (Table 2). Compared with marine oil shale, these samples had higher contents of quartz, feldspar, and clay minerals, and much lower average content of calcite (data from 13 marine oil shale samples in Quse Formation of the Early Jurassic, Qiangtang Depression show average quartz: 12.72%; feldspar: 3.38%; clay minerals: 14.28%; calcite: 68.53%,) (Fu et al. 2016). The relatively low calcite and high clay mineral contents are likely associated with terrigenous clastic input (Yu and Zhu, 2013; Zeng et al. 2013, 2014; Xie et al. 2014).

4.3 Major element geochemistry

Major elements can be used to construct associations between elements and minerals in oil shale, as demonstrated by prior study (Fu et al. 2010a, b). Table 3 lists major element concentrations of the oil shale samples. Average contents of SiO_2 (47.28%), Al_2O_3 (13.13%), and TiFe_2O_3 (6.00%) were relatively high and the rest were relatively low (<5%). Some major elements show positive correlation with ash yield at 95% confidence level, such as Si ($r = 0.978$), Al ($r = 0.571$), Mg ($r = 0.574$), Ca ($r = 0.397$), Na ($r = 0.683$), K ($r = 0.719$), Mn ($r = 0.123$), and Ti ($r = 0.355$), indicating that these elements have relationships with minerals. P_2O_5 ($r = 0.654$) and TiFe_2O_3

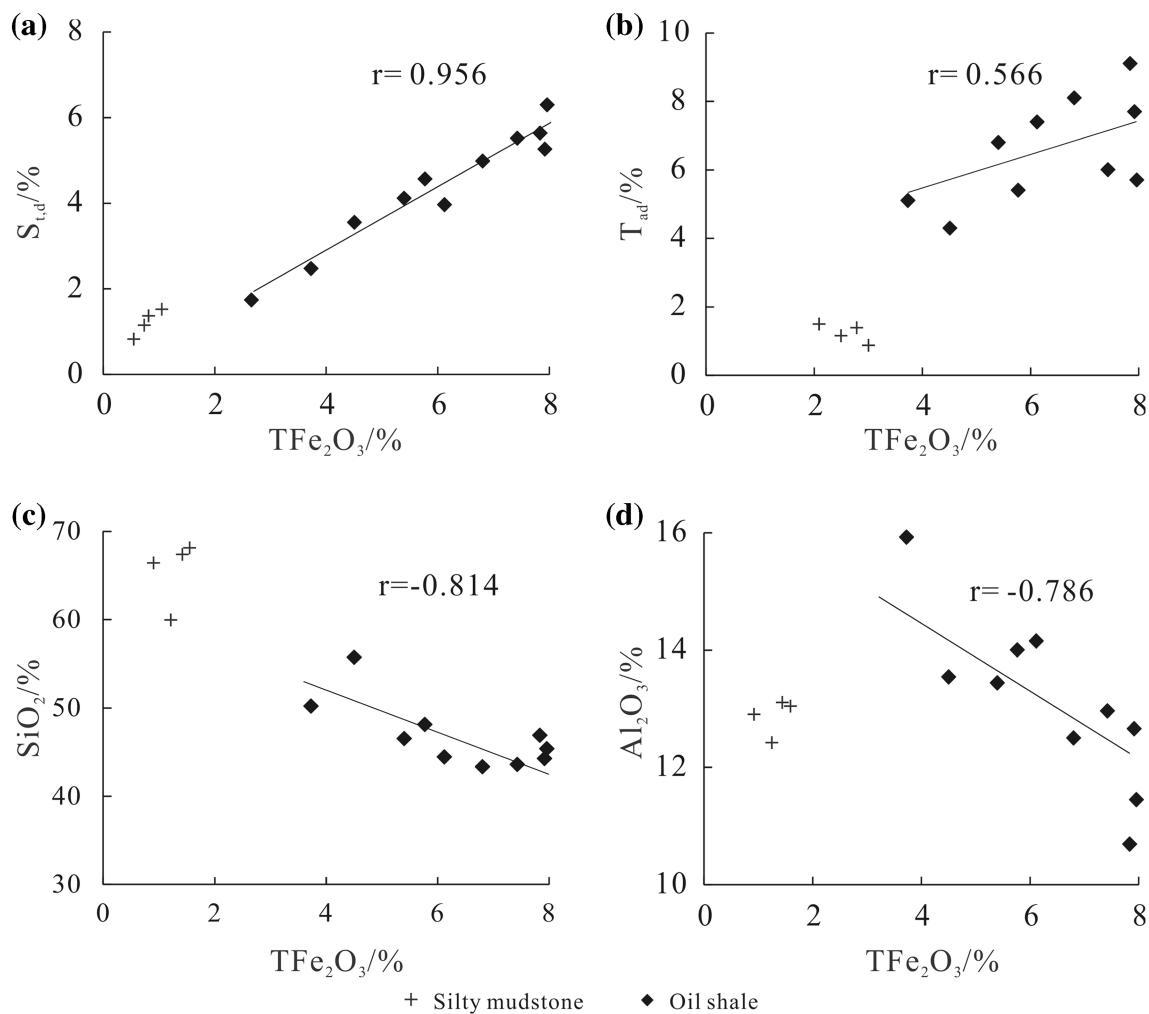


Fig. 7 Relationships between TFe₂O₃ concentration and total sulfur, oil yield, SiO₂, and Al₂O₃ concentrations in oil shale and silty mudstone samples

($r = 0.836$) show positive correlation with oil yield, suggesting P and Fe are mainly associated with organic matter.

The elements Si, Al, Ti, and K have some connection with quartz, feldspar, and clay minerals (Fu et al. 2010a,b). Relatively higher correlations among them (Fig. 6) illustrate that they mainly stem from a mixed clay assemblage, in accordance with clay mineral analysis. The Al/Si ratio can reflect the source of SiO₂ (Fu et al. 2010a,b). The Al/Si of oil shale samples varied from 0.26 to 0.37 (average 0.32), implying most SiO₂ is in the form of quartz, with some as clay minerals, which is consistent with the high abundance of quartz in XRD analysis.

The element Fe is generally associated with pyrite in oil shale (Fu et al. 2010a, b). The significantly positive correlation between TFe₂O₃ and total sulfur indicates a high proportion of Fe arising from sulfide oxidation and organic sulfur as the dominant component of total sulfur (Fig. 7a). This is corroborated by the positive correlation between TFe₂O₃ and oil yield (Fig. 7b). TFe₂O₃ showed

negative correlations with SiO₂ and Al₂O₃ (Fig. 7c, d), indicating a low frequency of iron in the clay minerals, which contrasts with observations made of marine oil shale (Fu et al. 2010a, b). Moreover, according to XRD, there was a small amount of Fe present as siderite, ankerite, and jarosite, demonstrating the occurrence of segmental Fe.

Calcium presents in various forms (Mukhopadhyay et al. 1998). In marine oil shale, a high abundance of calcite indicates that Ca is associated with calcite (Fu et al. 2016). However, in the Zhangjiatan oil shale samples, the calcite content was relatively low, suggesting different forms of Ca are present. Low correlation ($r = 0.396$) between CaO and Al₂O₃ suggests presence of Ca not only in clays, but also in other minerals. XRD analysis showed Ca in gypsum. Generally, low CaO content indicates sparse fossil remains in oil shale (Mukhopadhyay et al. 1998). Fish fossil remains observed in Zhangjiatan samples (Fig. 5b) support Ca being of biological origin.

Table 4 Trace element contents (ppm) and corresponding UCC values

Sample no.	Cr	Sr	Zr	Ba	Sc	V	Cu	Zn	Ga	As	Pb	Rb	Cs	Th
ZK702H1	18.30	163.30	125.80	546.30	7.80	43.50	22.60	58.60	18.70	17.90	36.60	169.00	20.30	28.50
ZK702H6	38.90	110.40	138.70	380.90	10.80	123.80	90.30	98.00	22.10	82.40	51.90	215.30	8.33	22.20
ZK702H8	18.30	117.20	111.40	536.40	5.00	39.60	20.90	48.20	17.60	18.90	32.40	189.10	10.60	31.50
ZK702H14	80.40	162.00	120.40	515.70	12.70	282.80	139.20	117.90	23.70	101.60	45.90	107.60	9.72	14.90
ZK702H18	60.20	229.80	135.50	463.40	13.90	235.30	144.50	107.00	19.20	80.10	35.40	88.90	7.88	10.80
ZK702H24	58.10	384.90	132.60	774.60	16.50	305.90	142.10	116.70	22.10	104.40	47.00	82.90	7.56	9.56
ZK702H32	68.40	328.20	146.80	792.60	13.10	316.80	132.30	122.10	24.10	153.30	37.90	117.30	11.30	12.70
ZK702H35	50.90	506.50	159.70	830.60	11.40	82.70	28.90	65.80	18.50	24.80	25.70	142.70	3.79	10.50
ZK1501H1	16.50	137.20	137.50	462.10	7.30	48.60	25.80	57.20	17.10	21.40	36.20	191.30	19.00	31.80
ZK1501H3	56.10	185.00	120.30	562.40	10.10	255.50	139.30	99.80	21.00	81.80	43.80	85.30	10.20	15.80
ZK1501H6	78.60	151.70	116.90	438.50	13.20	286.70	150.10	101.20	22.00	109.20	47.70	81.90	8.57	12.60
ZK1501H11	71.30	438.80	127.20	765.10	16.90	299.70	141.90	121.90	22.50	84.50	48.00	88.70	8.14	9.96
ZK1501H15	55.50	425.00	146.30	849.80	11.60	249.20	123.10	117.80	22.70	113.20	52.80	114.00	9.69	15.10
ZK2709H3	68.60	192.40	127.90	649.50	16.20	189.50	97.50	100.50	19.00	65.60	31.90	115.70	7.70	13.10
ZK2709H6	76.30	182.00	134.00	612.00	14.50	179.00	95.90	95.10	17.90	44.30	29.70	105.00	7.24	12.00
ZK2709H8	78.10	177.70	132.80	662.00	16.60	170.00	64.20	101.00	20.50	37.90	33.60	122.90	8.23	14.00
UCC ^a	35	350	190	550	13.6	60	25	71	17	1.5	20	112	3.7	10.7
Sample no.	U	Co	Ni	Nb	Cd	Hf	Ta	Li	Be	In	Bi	Mo	B	
ZK702H1	13.00	6.18	11.90	32.10	0.44	10.10	0.94	36.90	3.00	0.08	0.79	10.10	21.06	
ZK702H6	28.50	13.30	22.00	21.70	0.79	7.08	0.97	35.60	3.17	0.08	0.89	44.97	22.81	
ZK702H8	16.20	4.37	8.04	25.60	0.21	6.15	0.94	30.80	3.99	0.06	0.91	8.03	15.21	
ZK702H14	54.10	21.60	34.20	13.80	0.87	5.17	0.78	43.20	2.47	0.07	0.80	52.32	32.08	
ZK702H18	61.30	15.10	29.10	9.20	0.78	4.44	0.86	25.30	1.93	0.06	0.56	76.38	15.51	
ZK702H24	61.10	14.60	26.90	7.86	0.75	4.57	0.57	26.70	1.82	0.06	0.48	76.81	14.70	
ZK702H32	47.00	14.70	28.30	12.50	0.90	4.84	0.81	25.00	2.56	0.07	0.63	74.14	19.57	
ZK702H35	6.80	7.06	15.10	11.10	0.18	3.05	0.38	16.10	2.00	0.05	0.32	10.13	18.91	
ZK1501H1	16.50	5.03	9.93	17.70	0.44	5.86	0.90	28.30	3.25	0.05	0.74	12.70	14.50	
ZK1501H3	68.20	15.80	27.70	12.30	0.93	4.98	0.72	23.80	2.16	0.06	1.16	52.65	18.44	
ZK1501H6	61.70	18.90	31.20	20.00	0.92	4.15	0.52	24.70	1.91	0.06	0.75	24.64	20.86	
ZK1501H11	51.10	17.70	30.00	6.69	1.08	3.90	0.54	26.10	2.00	0.06	0.52	58.32	20.00	
ZK1501H15	47.30	17.40	31.00	20.00	1.12	6.00	0.66	28.20	2.56	0.08	0.62	62.02	26.56	
ZK2709H3	17.70	17.80	36.80	19.90	0.60	5.86	2.11	34.60	2.45	0.08	0.59	14.40	29.50	
ZK2709H6	21.60	17.50	36.90	18.10	0.68	5.90	1.76	35.30	2.25	0.07	0.55	23.00	43.80	
ZK2709H8	12.70	17.10	36.70	17.40	0.51	5.51	1.52	40.60	2.47	0.08	0.53	25.00	51.00	
UCC ^a	2.8	10	20	25	0.098	5.8	2.2	20	3	0.05	0.127	1.5	15	

^a data cited from McLennan (2001)

4.4 Trace element geochemistry

The trace element contents of samples are presented in Table 4. Ba, Sr, V, Zr, Cu, Rb, and Zn were in relatively high abundance (>100 ppm on average). Compared with the average value of the upper continental crust (UCC) (McLennan 2001), As, U, Cd, and Mo were highly enriched, while Sr, Nd, and Ta were depleted. Element enrichment was quantified by the enrichment factor (EF): EF = (X/Al)_{sample}/(X/Al)_{ucc}. All analyzed trace elements from oil shale samples were classified into three groups by

EF (Fig. 8; Table 5): (1) Cu, As, U, Cd, Bi, and Mo (average EF ≥ 5); (2) Cr, Ba, Sc, V, Zn, Ga, Pb, Rb, Cs, Th, Co, Ni, Hf, Li, In, and B (1 ≤ average EF < 5); and (3) Sr, Zr, Nb, Ta, and Be (average EF < 1). In general, Cu is associated with biological growth and Mo with chalcophile elements (Sun et al. 2015). The highly enriched Cu and Mo imply that there was high productivity during oil shale formation. The extremely high U concentration in oil shale is probably due to redox conditions (Cao et al. 2012; Bai et al. 2015).

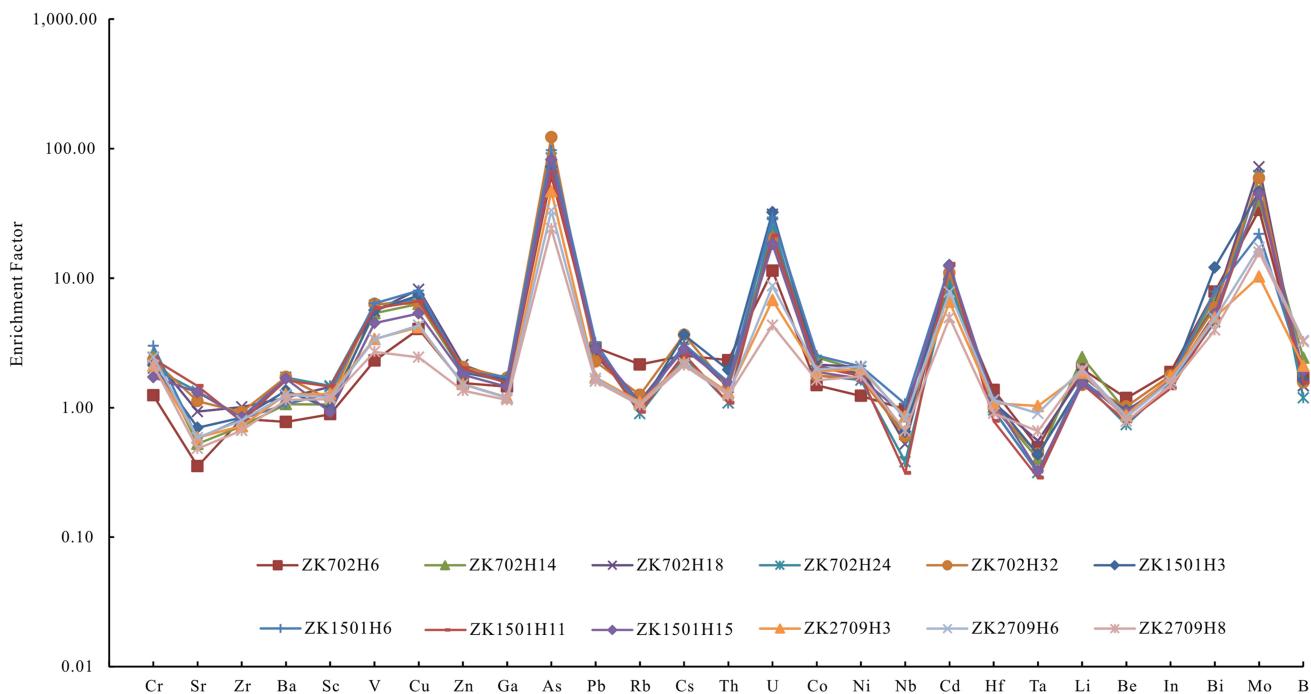


Fig. 8 Spider diagram of trace elements of oil shale samples

Ba content gradually decreases with increased distance from the lakeshore (Sun et al. 1997). In addition, Ba is often associated with paleoproductivity and can reflect organic matter content (Sun et al. 1997). The average EF of Ba (1.34) in this study indicates that the oil shale was deposited in a depocenter with high organic matter.

The Sr EF of oil shale samples was higher than that of mudstone from Chang 6 to Chang 3 in the southern Ordos Basin (EF < 1.0; Qiu et al. 2015). Increased Sr is usually associated with increased salinity and calcite (Zhang et al. 2004). This suggests the paleosalinity of southern Ordos Basin water gradually reduced after Chang 7 deposition. In addition, the EF of U in oil shale samples (Table 5) was clearly higher than that in mudstone from Chang 6 to Chang 3 (EF just above 1.0), suggesting that reducibility of water gradually declined. Compared with marine oil shale from Bilong Co in China (Fu et al. 2016), lacustrine oil shale from Chang 7 returned a significantly lower EF of Sr and higher EF of U. The variation of EF may be due to mineral content, water depth, redox conditions, or paleoclimate (Zhang et al. 2004, 2011; Ma et al. 2016).

5 Discussion

5.1 Element associations

Vertical distributions of selected elements are shown in Fig. 9. Compared with silty mudstone, most oil shale

samples returned high Cu, U, and V concentrations and low Al and Si concentrations. The variations of Fe, Cu, U, and V roughly paralleled oil yield and total sulfur, indicating that the four elements are closely associated with organic matter. The vertical distributions of Al and Si are similar to that of ash yield, suggesting quartz and clay minerals are the bases of ash yield and Si is present in clay minerals. The element patterns are relatively consistent, indicating that these elements are controlled by a common sedimentary environment (Fu et al. 2010a, b, c).

According to the established sedimentary cycle of the Yanchang Formation, the lake reached its largest scale during oil shale deposition (He 2003), leading to different sedimentary environments between oil shale and silty mudstone. The variation of environment may have influenced the presence of Al and Si.

Correlation coefficients and tree diagrams in cluster analysis help quantify correlations and attribute relationships in samples by grouping samples to optimize within-group similarity (Zhu et al. 2000; Zhao et al. 2015). Statistical Program for Social Sciences (SPSS) version 20.0 developed by IBM was used to analyze element associations. By cluster analysis, trace elements, major elements, total sulfur, oil yield, and ash yield were classified into two groups (Fig. 10).

Group A includes total sulfur, Fe, Cu, U, V, Sr, Ba, Zn, Ga, As, Mo, Pb, Cd, Cs, Na, P, Ca, Mn, and T_{ad} . The correlation coefficients of Fe-Cu (0.918), U-Cu (0.944), and V-Cu (0.847) were all higher than 0.80, indicating a

Table 5 Enrichment factors of trace elements

Sample no.	Cr	Sr	Zr	Ba	Sc	V	Cu	Zn	Ga	As	Pb	Rb	Cs	Th
ZK702H1	0.61	0.54	0.77	1.15	0.67	0.84	1.05	0.96	1.28	13.86	2.13	1.75	6.37	3.09
ZK702H6	1.25	0.35	0.82	0.78	0.89	2.31	4.05	1.55	1.46	61.55	2.91	2.15	2.52	2.32
ZK702H8	0.61	0.39	0.69	1.15	0.43	0.78	0.98	0.80	1.22	14.81	1.90	1.98	3.37	3.46
ZK702H14	2.61	0.53	0.72	1.06	1.06	5.35	6.32	1.88	1.58	76.85	2.60	1.09	2.98	1.58
ZK702H18	2.44	0.93	1.01	1.20	1.45	5.57	8.20	2.14	1.60	75.78	2.51	1.13	3.02	1.43
ZK702H24	2.01	1.33	0.85	1.71	1.47	6.19	6.90	1.99	1.58	84.47	2.85	0.90	2.48	1.08
ZK702H32	2.34	1.12	0.93	1.73	1.15	6.33	6.34	2.06	1.70	122.46	2.27	1.25	3.66	1.42
ZK702H35	1.77	1.76	1.02	1.84	1.02	1.68	1.41	1.13	1.33	20.16	1.57	1.55	1.25	1.20
ZK1501H1	0.54	0.45	0.84	0.97	0.62	0.94	1.19	0.93	1.16	16.47	2.09	1.97	5.93	3.43
ZK1501H3	2.12	0.70	0.84	1.35	0.98	5.64	7.38	1.86	1.64	72.25	2.90	1.01	3.65	1.96
ZK1501H6	3.00	0.58	0.82	1.06	1.29	6.38	8.01	1.90	1.73	97.13	3.18	0.98	3.09	1.57
ZK1501H11	2.38	1.47	0.78	1.63	1.45	5.85	6.64	2.01	1.55	65.94	2.81	0.93	2.58	1.09
ZK1501H15	1.72	1.32	0.83	1.67	0.92	4.50	5.34	1.80	1.45	81.77	2.86	1.10	2.84	1.53
ZK2709H3	2.10	0.59	0.72	1.27	1.28	3.39	4.18	1.52	1.20	46.89	1.71	1.11	2.23	1.31
ZK2709H6	2.46	0.59	0.80	1.26	1.20	3.37	4.33	1.51	1.19	33.33	1.68	1.06	2.21	1.27
ZK2709H8	2.13	0.48	0.67	1.15	1.16	2.70	2.45	1.36	1.15	24.08	1.60	1.05	2.12	1.25
Average	2.21	0.83	0.82	1.32	1.19	4.80	5.84	1.80	1.48	70.21	2.49	1.15	2.78	1.48
Sample no.	U	Co	Ni	Nb	Cd	Hf	Ta	Li	Be	In	Bi	Mo	B	
ZK702H1	5.39	0.72	0.69	1.49	5.22	2.02	0.50	2.14	1.16	1.74	7.23	7.82	1.63	
ZK702H6	11.40	1.49	1.23	0.97	9.03	1.37	0.49	1.99	1.18	1.88	7.85	33.59	1.70	
ZK702H8	6.80	0.51	0.47	1.20	2.52	1.25	0.50	1.81	1.56	1.39	8.42	6.29	1.19	
ZK702H14	21.92	2.45	1.94	0.63	10.07	1.01	0.40	2.45	0.93	1.59	7.15	39.58	2.43	
ZK702H18	31.07	2.14	2.06	0.52	11.29	1.09	0.55	1.80	0.91	1.70	6.26	72.26	1.47	
ZK702H24	26.48	1.77	1.63	0.38	9.29	0.96	0.31	1.62	0.74	1.53	4.59	62.14	1.19	
ZK702H32	20.11	1.76	1.70	0.60	11.00	1.00	0.44	1.50	1.02	1.77	5.94	59.23	1.56	
ZK702H35	2.96	0.86	0.92	0.54	2.24	0.64	0.21	0.98	0.81	1.20	3.07	8.24	1.54	
ZK1501H1	6.80	0.58	0.57	0.82	5.18	1.17	0.47	1.63	1.25	1.15	6.73	9.77	1.12	
ZK1501H3	32.27	2.09	1.83	0.65	12.57	1.14	0.43	1.58	0.95	1.56	12.10	46.50	1.63	
ZK1501H6	29.40	2.52	2.08	1.07	12.53	0.95	0.32	1.65	0.85	1.65	7.88	21.92	1.86	
ZK1501H11	21.36	2.07	1.76	0.31	12.90	0.79	0.29	1.53	0.78	1.40	4.79	45.51	1.56	
ZK1501H15	18.30	1.89	1.68	0.87	12.38	1.12	0.33	1.53	0.92	1.65	5.29	44.80	1.92	
ZK2709H3	6.78	1.91	1.97	0.85	6.56	1.08	1.03	1.85	0.88	1.74	4.98	10.29	2.11	
ZK2709H6	8.71	1.98	2.08	0.82	7.83	1.15	0.90	1.99	0.85	1.60	4.89	17.31	3.30	
ZK2709H8	4.32	1.63	1.75	0.66	4.96	0.91	0.66	1.93	0.78	1.50	3.98	15.88	3.24	
Average	19.34	1.98	1.81	0.69	10.04	1.05	0.51	1.78	0.90	1.63	6.31	39.08	2.00	

$$\text{EF} = (\text{X/Al})_{\text{sample}} / (\text{X/Al})_{\text{ucc}}$$

close relationship among Fe, Cu, U, and V (Table 6). Elements from this group generally had negative correlation coefficients with ash yield (15 out of 17 elements had negative correlation, with values ranging from -0.890 to 0.277) and generally positive correlation coefficients with oil yield (12 out of 17 elements had positive correlation, with values ranging from -0.424 to 0.566) (Table 6). The positive correlations with oil yield indicate that these elements are mainly associated with organic matter. Previous

research has suggested an extremely high positive correlation between total organic carbon and oil shale yield (Liu et al. 2009; Zhang et al. 2013).

Group B includes T_{ad}, Si, K, Rb, Be, In, Hf, Th, Mg, Ti, Ni, B, Cr, Al, Li, Co, Nb, Ta, Bi, Zr, and Sc. These elements generally had positive correlation coefficients with ash yield (13 out of 17 elements had positive correlation, with values ranging from -0.469 to 0.954) and generally negative correlation with oil yield (14 out of 17 elements

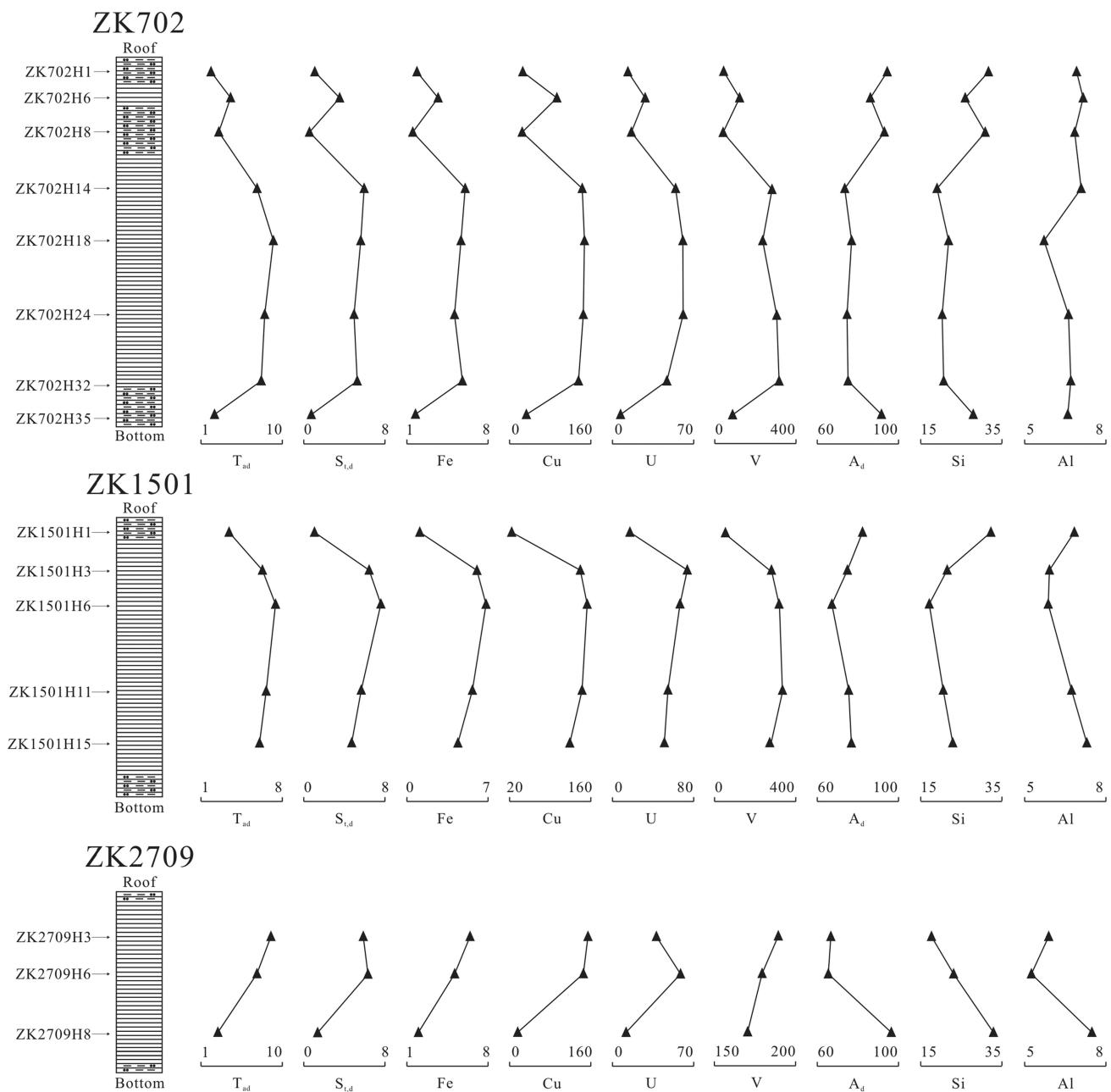


Fig. 9 Vertical variation of oil yield, total sulfur, ash yield, certain trace elements, and major elements in oil shale sections (trace elements in ppm, others in %)

had negative correlation, with values ranging from -0.707 to 0.326 (Table 6). In addition, the elements in this group showed a positive relationship with Al (with the exception of Bi), indicating a terrigenous source.

5.2 Reconstruction of paleoenvironment

Previous studies have suggested that the ratios and calculated values of certain elements can be used to reconstruct

lacustrine paleoenvironment (Fan et al. 2012; Bai et al. 2015; Sun et al. 2015; Fu et al. 2016; Jemaï et al. 2016).

The Sr/Cu ratio is sensitive to paleoclimate (Deng and Qian 1993; Fu et al. 2010a, b, c; Liang et al. 2015). Sr/Cu ratios from 1.3 to 5.0 indicate a warm and humid paleoclimate; >5.0 , dry and hot. Sr/Cu ratios of our oil shale samples ranged from 1.01 to 12.16, with an average of 3.26 (Table 7), implying that the integral paleoclimate was warm and humid.

The Ba/Al ratio can be used to semiquantitatively reconstruct paleoproductivity of lakes (Dean et al. 1997;

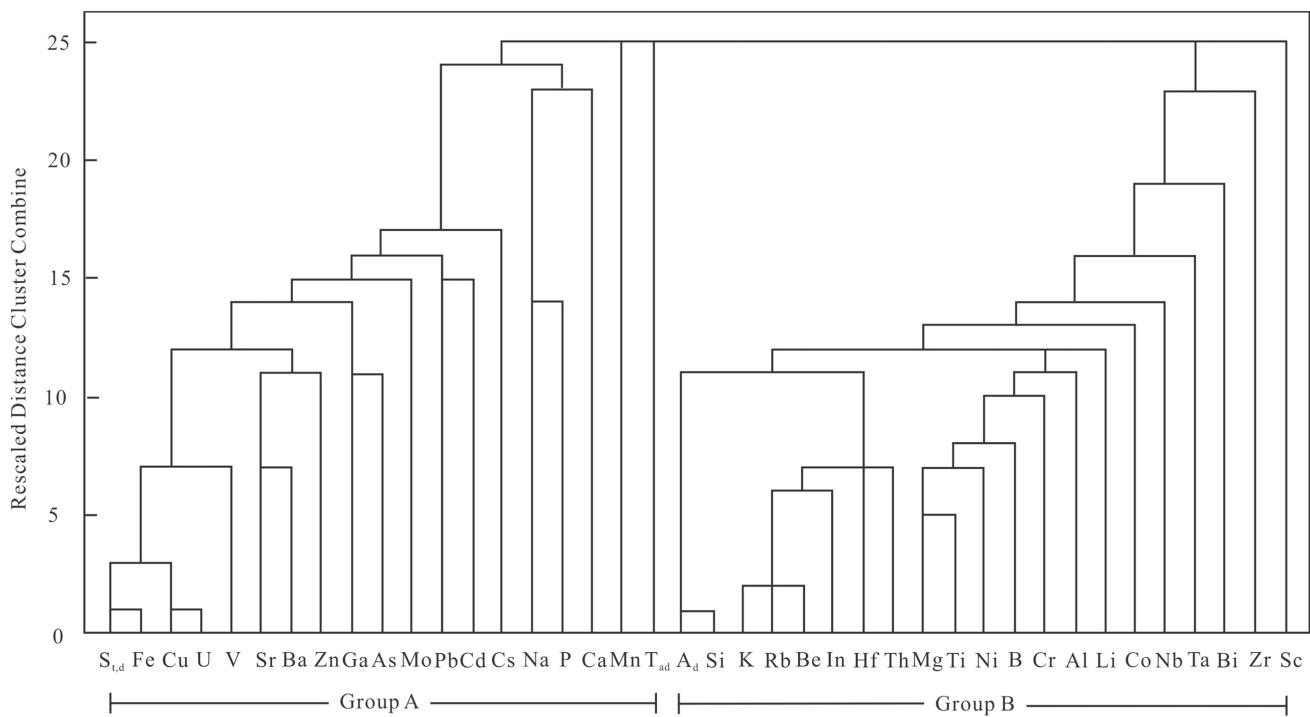


Fig. 10 Cluster analysis of major elements, trace elements, oil yield, ash yield, and total sulfur in oil shale samples (cluster method, Furthest Neighbor; interval, Person correlation; transform values, maximum magnitude of 1)

Luo et al. 2013). Higher values are associated with higher paleoproductivity; a ratio of 0.010 to 0.012 has been designated as high productivity (Dean et al. 1997; Sun et al. 2015). Ba/Al in our samples varied from 0.005 to 0.012, with an average of 0.009 (Table 7), suggesting relatively high paleoproductivity.

The Sr/Ba ratio can be used to reconstruct paleosalinity (Wang et al. 1979, 2014; Wang and Wu 1983; Li and Chen 2003; Guo et al. 2015; Li et al. 2015). Sr/Ba > 1, 0.6–1, and <0.6 reflect sea water, brackish water, and fresh water, respectively. Sr/Ba ratios of our samples were generally lower than 1.0 (from 0.27 to 0.69, with an average of 0.38) (Table 7), suggesting that the oil shale mainly formed in fresh water. This could support the opinion that there was no large-scale marine transgression in the southern Ordos Basin during the Triassic (Yin and Lin 1979; Zhang 1984; Zhang et al. 2011; Qiu et al. 2015).

V, Ni, U, and Th are sensitive to redox conditions and V/(V + Ni), U/Th, AU(U-Th/3), and δU(2U/(Th/3 + U)) can reflect paleo-redox conditions (Ernst 1970; Deng and Qian 1993; Jones and Manning 1994; Dypvik and Harris 2001; Teng et al. 2005; Tribouillard et al. 2006; Zhao et al. 2016a, b). V/(V + Ni) < 0.5 and δU < 1 both indicate oxidation conditions; V/(V + Ni) > 0.5 and δU > 1, reducing conditions (Deng and Qian 1993; Zhao et al. 2016a, 2016b). Strong oxidation conditions are reflected by U/Th < 0.75 and AU < 5 ppm; strong reducing conditions by U/Th > 1.25 and AU > 12 ppm; intermediate values

indicate weak oxidation–weak reducing conditions (Qiu et al. 2015; Sun et al. 2015). Average values of V(V + Ni), δU, U/Th, and AU in oil shale samples were 0.88, 1.71, 3.13, and 35.32 ppm, respectively (Table 7), indicating that the Zhangjitan oil shale was mainly deposited in reducing conditions.

The Fe/Ti ratio can be used to measure hydrothermal influence (Boström 1983; Zhong et al. 2015; Chu et al. 2016). Fe/Ti > 20 reflects definite hydrothermal influence, with higher values indicating greater influence (Chu et al. 2016). Six samples (ZK702H14, ZK702H18, ZK702H24, ZK702H32, ZK1501H3, and ZK1501H6) from the eastern study area had Fe/Ti > 20 (Table 7), meaning there was hydrothermal influence during oil shale sedimentation. However, three samples (ZK2709H3, ZK2709H6, and ZK2709H8) from the western study area had Fe/Ti < 20 (Table 7), indicating less hydrothermal influence.

6 Conclusions

The oil yields of Zhangjitan oil shale samples ranged from 4.3% to 9.1% (average 6.63%), classifying the oil shale as middle grade.

Minerals in the oil shale were mainly clay minerals, quartz, feldspar, and pyrite (>5%), with illite/smectite dominating the clay minerals. In contrast with marine oil

Table 6 Correlation coefficients of oil yield, calorific value, ash yield, total sulfur, and major elements in oil shale samples

Cases	T _{st}	A _d	S _{st}	Si	Al	Mg	Ca	Na	K	P	Mn	Ti	Fe	Cr	Sr	Zr	Ba	Sc	V	Cu	Zn	Ga	As	Pb	Rb	Cs	Th	U	Co	Ni	Nb	Cd	Hf	Ta	Li	Be	In	Bi	Mo	B
T _{st}	1.000	-0.573	0.396	-0.533	-0.528	-0.152	0.238	0.114	-0.703	0.313	-0.424	-0.184	0.566	0.287	0.156	-0.049	0.055	0.326	0.491	0.523	0.276	-0.162	0.768	-0.402	-0.583	-0.110	-0.707	0.372	0.048	0.130	-0.511	-0.129	-0.547	-0.020	-0.268	-0.603	-0.442	-0.397	-0.363	-0.361
A _d	1.000	-0.886	0.954	0.626	-0.130	-0.372	0.164	0.192	0.277	0.304	-0.890	-0.466	-0.176	0.489	-0.042	0.007	-0.795	-0.817	-0.316	-0.227	-0.503	-0.122	0.763	-0.211	0.557	-0.687	-0.469	-0.112	0.297	-0.420	0.719	0.431	0.462	0.685	0.670	-0.038	-0.145	0.421		
S _{st}	1.000	-0.778	-0.824	-0.420	0.406	-0.227	-0.631	0.037	-0.428	-0.543	0.956	0.185	0.067	-0.531	-0.217	-0.310	0.726	0.914	0.261	0.330	0.544	0.349	-0.583	0.335	-0.286	0.866	0.299	-0.278	-0.340	0.603	-0.087	-0.655	-0.573	-0.564	-0.261	0.369	-0.287	-0.631		
Si	1.000	0.433	-0.067	-0.546	0.210	0.743	0.007	0.229	0.090	-0.814	-0.657	0.177	0.400	-0.071	-0.198	-0.757	-0.687	-0.305	-0.210	-0.377	-0.027	0.768	0.133	0.607	-0.876	-0.609	-0.204	0.381	-0.267	0.741	0.395	0.785	0.678	0.588	-0.057	0.020	0.247			
Al	1.000	0.585	-0.288	-0.018	0.573	-0.195	0.538	0.671	-0.786	0.176	0.342	0.334	-0.499	-0.825	-0.064	-0.033	-0.409	-0.176	0.436	-0.130	0.260	-0.827	0.189	0.480	0.443	-0.411	0.596	0.554	0.730	0.555	0.768	-0.031	-0.461	0.784						
Mg	1.000	0.402	-0.546	0.210	0.743	0.007	0.229	0.090	-0.814	-0.657	0.177	0.400	-0.071	-0.198	-0.757	-0.687	-0.305	-0.210	-0.377	-0.027	0.768	0.133	0.607	-0.876	-0.609	-0.204	0.381	-0.267	0.741	0.395	0.785	0.678	0.588	-0.057	0.020	0.247				
Ca	1.000	0.402	-0.694	0.599	0.482	0.188	0.332	0.375	0.498	-0.212	0.388	0.340	0.434	0.409	0.131	-0.240	-0.021	0.054	-0.754	-0.238	-0.711	0.277	0.291	0.316	-0.247	0.374	-0.579	-0.133	0.535	-0.736	-0.887	-0.344	-0.061	-0.119						
Na	1.000	-0.354	0.737	0.438	0.326	-0.210	-0.247	0.500	0.146	0.491	0.308	0.009	0.005	0.054	-0.277	-0.341	-0.193	0.319	-0.327	-0.398	0.033	-0.201	0.030	-0.482	-0.010	-0.086	0.130	-0.271	-0.366	-0.291	-0.277	0.745	-0.091							
K	1.000	-0.444	0.158	0.116	-0.660	-0.361	-0.454	0.315	-0.393	-0.312	-0.760	-0.718	-0.373	0.024	-0.244	0.006	-0.466	0.078	0.855	-0.366	-0.213	0.126	0.653	-0.252	0.820	0.382	0.535	0.936	0.831	0.275	-0.375	-0.392								
P	1.000	0.525	0.103	0.080	-0.134	0.573	0.233	0.431	0.161	0.136	0.248	0.211	-0.373	-0.101	-0.122	-0.346	-0.016	-0.465	0.174	-0.020	0.150	-0.324	0.292	-0.177	0.037	-0.388	-0.391	0.313	-0.330	0.290	-0.209									
Mn	1.000	0.555	-0.477	0.063	0.416	0.268	0.569	0.245	-0.238	-0.395	-0.088	-0.004	-0.170	-0.077	-0.477	-0.157	0.423	-0.271	0.149	0.255	0.366	0.045	0.130	0.138	0.288	-0.318	-0.338	0.399												
Ti	1.000	-0.497	0.584	-0.125	-0.174	0.278	0.559	-0.374	-0.643	-0.303	-0.521	-0.736	-0.647	-0.150	-0.335	-0.275	-0.678	0.390	0.845	0.165	-0.539	0.101	0.701	0.511	-0.079	0.247	-0.359	-0.625	0.843											
Fe	1.000	0.266	0.116	-0.469	-0.124	-0.192	0.799	0.918	0.393	-0.357	0.601	0.204	-0.616	0.394	-0.388	0.825	0.311	-0.155	-0.459	0.521	-0.737	-0.571	-0.521	-0.556	-0.718	0.254	0.563	-0.632												
Cr	1.000	-0.090	-0.432	0.106	0.492	-0.294	0.073	-0.082	-0.082	-0.151	-0.460	-0.541	-0.050	0.545	-0.134	0.763	0.801	-0.067	0.196	-0.454	0.227	0.290	-0.392	0.183	-0.345	-0.406	0.504													
Sr	1.000	0.420	0.852	0.316	0.595	0.383	0.788	0.368	0.429	0.309	-0.373	0.129	-0.525	0.304	-0.125	-0.154	-0.513	0.555	-0.394	-0.426	-0.489	-0.320	-0.270	-0.492	0.006	-0.362														
Zr	1.000	0.468	-0.063	-0.170	-0.298	0.278	0.107	0.254	-0.036	0.417	0.173	0.134	-0.295	-0.569	-0.234	0.094	0.046	0.410	0.072	-0.082	0.457	0.486	-0.386	0.403	0.012															
Ba	1.000	0.477	0.063	0.416	0.268	0.569	0.245	-0.238	-0.395	-0.088	-0.004	-0.170	-0.077	-0.477	-0.157	0.423	-0.271	0.149	0.255	-0.170	-0.045	-0.219	-0.140	0.034	-0.507	0.310	-0.267													
Sc	1.000	0.090	-0.189	0.159	-0.280	-0.347	-0.434	-0.336	-0.599	-0.702	-0.344	-0.142	0.483	-0.340	-0.436	-0.363	0.341	0.142	-0.433	0.084	-0.227	-0.157	-0.112	0.624	-0.656															
V	1.000	0.847	0.767	0.590	0.222	0.301	-0.698	0.448	-0.400	0.760	0.221	-0.139	-0.646	0.6	0.590	-0.808	-0.681	-0.553	-0.597	-0.658	-0.089	0.577	-0.505	-0.376	0.434															
Cu	1.000	0.533	0.419	0.655	0.425	-0.609	0.310	-0.423	0.844	0.131	-0.365	-0.588	0.682	-0.736	-0.764	-0.666	0.613	-0.306	0.157	0.604	-0.700																			
Zn	1.000	0.736	0.694	0.399	-0.276	0.442	-0.379	0.437	0.116	-0.189	-0.581	0.588	-0.452	-0.580	-0.243	-0.156	-0.234	-0.318	0.717	-0.394																				
Ga	1.000	0.295	0.712	0.108	0.690	0.180	0.435	0.072	-0.456	-0.184	0.612	-0.194	-0.270	-0.123	0.209	-0.018	0.195	0.600	0.660	-0.600	-0.211	0.125																		
As	1.000	0.507	-0.100	0.697	-0.049	0.598	-0.103	-0.401	-0.207	0.638	-0.304	-0.668	0.500	0.000	-0.157	0.112	0.624	-0.656																						
Pb	1.000	0.191	0.280	0.512	-0.032	-0.649	-0.012	0.273	-0.020	-0.783	-0.241	0.137	-0.139	0.343	-0.387	-0.508																								
Rb	1.000	0.028	0.844	-0.540	-0.385	-0.355	0.561	-0.200	0.822	0.234	0.453	0.915	0.788	-0.217	-0.169	0.188																								
Cs	1.000	0.269	0.371	0.026	-0.251	-0.041	0.538	-0.066	-0.422	-0.213	0.286	0.009	0.498	0.383	-0.214	-0.214																								
Tb	1.000	-0.265	-0.198	-0.369	0.636	-0.636	-0.027	0.778	0.063	-0.372	0.854	0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600	-0.600											
U	1.000	-0.001	-0.509	-0.575	0.658	-0.662	-0.062	-0.851	-0.069	-0.561	-0.812	0.332	0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662	-0.662											
Co	1.000	0.698	0.140	0.117	-0.209	0.084	0.428	-0.182	-0.100	-0.415	0.401	-0.155	0.112	0.624	-0.656																									
Ni	1.000	-0.001	-0.426	-0.010	0.654	0.499	-0.160	-0.163	-0.391	-0.624	0.755																													
Nb	1.000	-0.234	0.736	0.462	0.419	0.626	0.722	-0.700																																
Cd	1.000	-0.319	-0.770	-0.547	-0.109	-0.098	-0.277	-0.227																																
Ta	1.000	0.551	0.292	0.586	-0.240	-0.760																																		
Li	1.000	0.524	0.624	-0.654	-0.069	-0.497	0.776																																	
Be	1.000	0.869	0.286	-0.194	0.312																																			
In	1.000	-0.086	-0.408	0.515																																				
Bi	1.000	-0.056	-0.226																																					
Mo	1.000	-0.672																																						

shale, quartz, feldspar, and clay minerals in lacustrine oil shale were clearly higher while calcite content was much lower. Si mainly occurred in quartz and in clay minerals.

Fe was mainly associated with organic matter and barely present in the clay minerals, which is opposite to marine oil shale. Ca occurred in various forms.

According to cluster analysis, Fe, Cu, U, V, Sr, Ba, Zn, Ga, As, Mo, Pb, Cd, Cs, Na, P, Ca, and Mn were generally associated with organic matter, while Si, K, Rb, Be, In, Hf, Th, Mg, Ti, Ni, B, Cr, Al, Li, Co, Nb, Ta, Bi, Zr, and Sc were not; all showed a positive relationship with Al (with the exception of Bi), indicating these elements are closely associated with a terrigenous source.

The Sr/Cu ratio of oil shale samples ranged from 1.01 to 12.16, with an average of 3.26, indicating a warm and humid paleoclimate. Ba/Al was between 0.005 and 0.012, suggesting that paleoproductivity was high. Sr/Ba varied from 0.27 to 0.69, suggesting that the oil shale was mainly deposited in fresh water and there was no large-scale marine transgression in the southern Ordos Basin. The average values of V/(V + Ni), U/Th, AU, and δ U of oil shale samples were 0.88, 3.13, 35.32 ppm, and 1.71, respectively, indicating that the shale was mainly deposited under reducing conditions. The Fe/Ti ratio of the six oil shale samples from the southern study area was >20, indicating clear hydrothermal influence; the ratio of samples from the western study area was <20, suggesting lesser hydrothermal influence.

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